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# First/Second Semester B.E. Degree Examination, Dec.2016/Jan.2017 **Engineering Physics**

Time: 3 hrs.

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Max. Marks: 80

15PHY12/22

(07 Marks)

(05 Marks)

Note: 1. Answer FIVE full questions, choosing one full question from each module. 2. Physical Constants : Velocity of light,  $c = 3 \times 10^{-8} \text{ ms}^{-1}$ , *Planck's constant*,  $h = 6.625 \times 10^{-34} J.S.$ Mass of electron,  $m_e = 9.1 \times 10^{-31} \text{ kg}$ ,

> Avogadro number,  $NA = 6.02 x 10^{26}$ /Kmol, Boltzmann constant,  $k = 1.38 \times 10^{-23} \text{ J/K}$ ,

Charge of an electron,  $e = 1.602 \times 10^{-19} C$ 

### Module-1

- a. State Planck's radiation law. Show how Planck's law could be reduced to Wien's law and Rayleigh-Jeans law. (07 Marks)
  - b. State Heisenberg's uncertainty principle and show that electron does not exist inside the nucleus by this principle. tormik **1** Wes. 'IMO (05 Marks)
  - c. Find deBroglie wavelength of a particle of mass 0.58 MeVA<sup>2</sup>Thas a kinetic energy 90 eV, where c is velocity of light. (04 Marks)

# OR

- 2 a. Using Schrodinger's time independent wave equation obtain eigen values and eigen function for a particle in a one dimensional potential well of infinite height. (07 Marks)
  - b. Define phase velocity and group velocity. Show that group velocity is equal to particle velocity. (05 Marks)
  - The inherent uncertainty in the measurement of time spent by Iridium 191 nuclei in the c. excited state is found to be  $1.4 \times 10^{1^{\circ}}$  s. Estimate the uncertainty that results in its energy in eV in the excited state. (04 Marks)

### Module-2

- Explain Meissner effect. Write any three differences between Type-I and Type-II 3 a. superconductors. (07 Marks) (05 Marks)
  - b. Explain the failure of classical electron theory.
  - c. For intrinsic Gallium Arsenide, the electric conductivity at room temperature is  $10^{-6}$  ohm' m'. The electron and hole mobilities are respectively 0.85 m<sup>2</sup>/V.S and  $0.04 \text{ m}^2/\text{V.S.}$  Calculate the intrinsic carrier concentration at room temperature. (04 Marks)

#### OR

- a. State law of mass action. Obtain an expression for electrical conductivity of semiconductors.
  - b. Explain the BCS theory of super conductivity.
  - Calculate the probability of finding an electron at an energy level 0.02 eV above Fermi level c. at 200 K. (04 Marks)

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## Module-3

- 5 a. Describe construction and working of carbon dioxide laser with suitable diagrams. (07 Marks)
  - b. Obtain an expression for the numerical aperture of an optical fiber. (05 Marks)
  - c. Find the ratio of population of two energy levels in a medium at thermal equilibrium, if the

wavelength of light emitted at 291 K is 6928 A.

### OR

- 6 a. Describe the recording and reconstruction process in holography with the help of suitable diagrams. (07 Marks)
  - b. Discuss point to point optical fiber communication system. (05 Marks)
  - c. Calculate the numerical aperture and angle of acceptance for an optical fiber having refractive indices 1.563 and 1.498 for core and cladding respectively. (04 Marks)

#### $Module_4$

7	a. I	Describe briefly the seven crystal systems.	(07 Marks)
	b.	Describe with a neat diagram the crystal structure of diamond.	(05 Marks)
	c.	Draw the crystal planes (102) (111) (011) and (002) in a cubic crystal.	(04 Marks)

### OR

- 8 a. Define atomic packing factor. Calculate the atomic packing factor for sc, bcc and fcc structures. (07 Marks)
  - b. Describe the construction and working of a Bragg's x-ray spectrometer. (05 Marks)
  - c. An x-ray beam of wavelength 0.7 A undergoes first order Bragg's reflection from the plane (302) of a cubic crystal at glancing angle 35°, calculate the lattice constant. (04 Marks)

## Module\_

9	a. Explain Ball Milling method of synthesis of nano materials.	(06 Marks)
	h. Describe hand operated Reddy shock tube with diagram.	(05 Marks)
V	U. Define shock waves. Mention its applications.	(05 Marks)

# OR

10	a. I	Explain the working of SEM with the help of a neat diagram.	(07 Marks)
	b.	Mention Rankine-Hugonit shock equations and expand the terms.	(05 Marks)
	c.	Write any four applications of carbon nano tubes.	(04 Marks)

(04 Marks)

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