

Roll No.					Total No.	of Pages	: 02

Total No. of Questions: 11

B.Sc. (Honours) Chemistry (Sem.-1)
ORGANIC CHEMISTRY-I
Subject Code: BHCL-102-19

M.Code: 77224

Time: 3 Hrs. Max. Marks: 60

INSTRUCTIONS TO CANDIDATES:

- SECTION-A is COMPULSORY consisting of EIGHT questions carrying TWO marks each.
- 2. SECTION-B contains EIGHT questions carrying FOUR marks each and students have to attempt any SIX questions.
- 3. SECTION-C will comprise of two compulsory questions with internal choice in both these questions. Each question carries TEN marks.

SECTION-A

1) Write briefly:

- a) What is the optical activity?
- b) What are localized and de-localized chemical bonds?
- c) What are aromaticity?
- d) Explain the third law of thermodynamics.
- e) What are geometrical isomers?
- f) What is markowniff's rule?
- g) Explain the boat and chair form in cyclohexane.
- h) Explain the Saytzeff rule.

SECTION-B

- 2) Explain the Clatherates with examples.
- 3) Explain the reactivity order of carbocation and factors affect this order.
- 4) What are enantiomers and properties of enantiomers?
- 5) Explain Thero and Erethro isomer form.

1 M-77224 (S111)-2385



- 6) Explain the D and L system with examples.
- 7) Explain the Bayer's strain theory.
- 8) Explain the 1,2-addition in dienes.
- 9) Explain the synthesis of DDT.

SECTION-C

10) Explain the Hydroboration-oxidation of alkenes.

OR

Explain Diel's Alder reaction with suitable example.

11) What do you understand by SN1 and SN2 mechanism? Illustrate your answer by considering the hydrolysis of alkyl halides.

Explain the Birch Reduction with example.

NOTE: Disclosure of Identity by writing Mobile No. or Making of passing request on any page of Answer Sheet will lead to UMC against the Student.

2 | M-77224 (S111)-2385