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**B.Sc.(MLS) (2014 to 2017) (Sem.-4)**

**Subject Code : BMLS-405**

**M.Code : 48120**

**Max. Marks : 60**

1. **SECTION-A is COMPULSORY** consisting of **TEN** questions carrying **TWO** marks each.
2. **SECTION-B** contains **FIVE** questions carrying **FIVE** marks each and students has to attempt any **FOUR** questions.
3. **SECTION-C** contains **THREE** questions carrying **TEN** marks each and students has to attempt any **TWO** questions.

**1) Answer briefly :**

- What is the basic principle of disposal of lab wastes?
- Explain the terms accuracy and specificity.
- What are disaccharides?
- How ammonia is made less toxic in our body?
- What are the instruments to detect radioactivity?
- List some common causes of fire in lab.
- What do you mean by blank and control?
- What is full form of ELISA?
- What is the purpose of conjugation of bilirubin?
- What is denaturation of proteins?

### SECTION-B

- 2) Describe the principle and method of determination of phosphorous in a blood sample.
- 3) Describe briefly what are the basic requirements for lab management?
- 4) Write short notes on values of central tendency in statistics.
- 5) What are radioisotopes and give their use in a clinical biochemistry lab.
- 6) Write a short note on the handling and proper disposal of radioactive materials.

### SECTION-C

- 7) What do you mean by quality control? List two phases of internal quality control.
- 8) Discuss various methods for the estimation of urea in a blood sample. Explain enzymatic method in detail along with its advantages.
- 9) Describe in detail the principle and procedure for determining the level of sodium in a clinical sample.

**NOTE : Disclosure of Identity by writing Mobile No. or Making of passing request on any page of Answer Sheet will lead to UMC against the Student.**