Code No. 6033 / S

FACULTY OF PHARMACY

B. Pharmacy I – Year (Supplementary) Examination, November 2015

Subjects Phermacoutical Ingraphic Chemistry

		Subject: Pharmaceutical inorganic Chemistry	
Time: 3 Hours Max.Marks: 70			
Note: Answer All questions. All questions carry equal marks.			
1		Discuss about sources of impurities in pharmaceuticals. Give four characteristic tests for group IV cations. OR 8	
		Define limit test and give its significance. 2+4 Give the principle and procedure in the limit test for mercury and iron. 4+4	
2		Discuss the importance of sodium in the body. Write the preparation, properties and uses of sodium chloride and sodium bicarbonate. 2+3+3 Give the preparation, assay methods and limit tests for following compounds: i) Calcium gluconate 3 ii) Sodium citrate	
	,	OR Explain in detail about dialysis fluids and their significances. Add a note on acid base regulators. 10	
3	•	Define desiccants with examples. Give the preparation, properties, test for purity of silica gel. Add a note on haematinics.	
	,	OR Define the following terms with examples. i) Excipients (ii) Suspending agents (iii) Absorbents Give the preparation and uses of following compounds. i) Ferrous sulphate (ii) Magnesium steasate	
4	a) b)	iii) Sodium metabisulfite (iv) Sodium phosphate	
	,	OR	
	c)	Give the preparation, properties, assay method, limit test and uses of any two of the following: 2x5 i) Potassium iodide (ii) Sodium thiosulphate (iii) Ammonium chloride	
	d)	What are inhalants and give their significance.	
5	•	Give the preparation, test for purity and uses of following compounds. 3x3=9 i) Silvernitrate (ii) Hydrogen peroxide (iii) Boric acid	
	,	Add a note on dentifrices. 5	
	,	Write the significance of cements and fillers. What are anti-infectives? Give the preparation, properties and uses of following:1+3+3+i) Zinc oxide (ii) Potassium permanganate (iii) Zinc stearate. ****	-3