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Code No.: 6502

## **FACULTY OF PHARMACY**

B. Pharmacy I Year (Supplementary) Examination, November/December 2011

## PHARMACEUTICAL INORGANIC CHEMISTRY

Time: 3 Hours] [Max.]		s : 70
	Answer all questions. All questions carry equal marks.	
1.	(a) (i) Explain the principle and procedure involved in the limit test for He metals.	eavy 8
	(ii) Give the two characteristic tests for two anions and two cations.	6
	$\mathbf{Or}$	
	(b) (i) Classify Pharmaceutical Inorganic compounds based on their application with examples.	ions 8
	(ii) Give the principle and procedure involved in the limit test for sulphates chlorides.	and 6
2.	(a) (i) Give the preparation, properties and uses of magnesium hydroxide a calcium carbonate.	<b>and</b> 10
	(ii) Write a brief note on Acid base regulators. Or	6
	(b) (i) What are Adsorbents? Write a note on Light Kaolin.	6
	(ii) Give the principle and procedures involved in the assay of (A) Calci gluconate (B) Magnesium Oxide.	ium
3.	(a) (i) What are suspending agents? Write a note on Bentonite.	7
	(ii) Write a note on Excipients.	7
	(b) (i) Write the preparation and uses of Sodium metabisulphite and Sodi Phosphate.	ium 6
	(ii) Write a brief note on purified water.	8

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4.	(a) (i) What are expectorants? Give the preparation, tests for purity Ammonium Chloride.	y and essay of 8
	(ii) Write a note on Inhalants.	6
	Or	
	(b) (i) Write the preparation properties and uses of Sodium Thiosulpha sulpahte.	ate and copper 8
	(ii) Give the principle and procedure involved in the Assay of Potassi Tartrate.	ium Antimonyl 6
5.	(a) (i) Write a note on Antidepressants?	6
	(ii) Give the preparation and uses of	9
	(A) Calcium phosphate.	
	(B) Barium Sulphate.	
	(C) Potassium bromide	
	Or	
	(b) (i) Write the principle and procedure involved in the assay of b Zinc oxide.	poric acid and 8
	(b) (i) Write the principle and procedure involved in the assay of be Zinc oxide.  (ii) Write a note on Dentifrices.	6