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GUJARAT TECHNOLOGICAL UNIVERSITY

BE- SEMESTER-V (NEW) EXAMINATION – WINTER 2020 Date:22/01/2021

Subject Code:3153622

Subject Name: Chemical synthesis for Technologists Time:10:30 AM TO 12:30 PM **Total Marks: 56 Instructions:** 1. Attempt any FOUR questions out of EIGHT questions.

- 2. Make suitable assumptions wherever necessary.
- 3. Figures to the right indicate full marks.

MARKS

07

04

07

03

04

- Q.1 (a) Write short note on Benzidine rearrangement. 03 (b) Write down Nucleophilic substitution reaction of Pyridine. 04 (c) Define Aromaticity. Explain Huckel rule in detail showing the behavioural approach of 07 different Organic molecules in proving their aromatic characteristics. 03
- Q.2 (a) State whether the following compounds are aromatic or non-aromatic.



(b) Name the following reaction and explain its mechanism and application in detail. 04



cyclohexanone

- (c) Explain why,
 - 1. Pyridine is more basic than pyrrole.
 - 2. Pyridine is less basic than aliphatic amine.
 - 3. Pyridine is more basic than aniline
- Q.3 (a) Explain Isomerism in substituted Naphthalene with example of it. 03

cyclohexanoxime

caprolactam

- Which of the following compound will exhibit optical activity? **(b)**
 - 1. Meso-2,3-Dihydroxybutane
 - 2. Mixture of 1 gm of (+)-erythro-2-bromo-3-chloro butane & 0.5 gm (-)-erythro-2-bromo-3-chloro butane
 - (c) Write detailed mechanism and applications of Darzens reaction.
- Write a note on Annulene. **Q.4** (a)
 - Give E & Z Notation to the following: **(b)**



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(c) Write detailed mechanism and applications of Rosenmund reaction.
(a) Explain only mechanism of Leuckart reaction.
(b) Write methods of synthesis for Furan.
(c) Write detailed mechanism and applications of Knoevenagel condensation.
(d) 03
(e) Write detailed mechanism and applications of Knoevenagel condensation.
(f) 03
(g) 04
(g) 04
(h) 05
(h) 05
(h) 06
(h) 07
(h) 07
(h) 07
(h) 07
(h) 03
(h) 04
(h



- (b) Write down electrophilic substitution reaction of Pyrrole. 04
- (c) Explain detailed mechanism of MPV reduction with one application.
- Q.7 (a) Differentiate between Enantiomers and Diastereomers. 03
 - (b) Explain why, methyl (-CH₃ group) acts as ortho-para directing.
 - (c) Name the following reaction and explain its mechanism and application in detail. 07



Q.8(a) Write a short note on Geometrical Isomerism.03(b) Write a note on: Opposite behaviour of Halogen group.04(c) Name the following reaction and explain its mechanism and application in detail.07





07

04