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Seat No.:		Enrolm	Enrolment No		
GUJARAT TECHNOLOGICAL UNIVERSITY B.PHARM - SEMESTER – 5 EXAMINATION – WINTER -2019					
Subject Code: 2250003Date: 20-11-201Subject Name: Pharmaceutical Analysis IIITotal MarksTime: 02:30 PM TO 05:30 PMTotal Marks			Date: 20-11-2019)	
			Total Marks: 80		
Instruc 1. 2. 3.	ctions Atte Mak Figu	: empt any five questions. we suitable assumptions wherever necessary. ares to the right indicate full marks.			
Q.1	(a)	Define: wavelength, frequency, wave number, fluorescence, Fermi resonance, 06 base ion			
	(b) (c)	Explain various types of deviations from Beers Lar Derive Beer-Lambert's law equation.	nbert's law.	05 05	
Q.2	(a) (b)	Explain various monochromators used in UV-Visib One tablet of drug (mol. wt.: 225, ϵ = 570) when a ml with water gave absorbance of 0.625 at 27 present in one tablet.	le spectrophotometer. dissolved and diluted to 1500 5 nm. Calculate mg of drug	06 05	
	(c)	Explain different types of molecular vibrations in I	R spectroscopy.	05	
Q.3	(a) (b) (c)	Explain various factors affecting vibrational freque Explain briefly principle and working of thermal de Write a detailed note on FTIR along with its advan	ency in IR spectroscopy. etectors. itages.	06 05 05	
Q.4	(a)	What is interference in atomic spectroscop interference and explain each in brief.	y? List different types of	06	
	(b) (c)	Write a brief note on hollow cathode lamp. Differentiate : premix burner and total consumpti photometry	on burner used in flame	05 05	
Q.5	(a) (b) (c)	Explain in detail chemical ionization source used Write a detailed note on quadrupolar analyzer. Explain briefly principle of mass spectrometry an mass spectrometer.	in mass spectrometer. Id draw a labeled diagram of	06 05 05	
Q. 6	(a) (b) (c)	 Explain chemical shift and magnetic anisotropy. Explain in detail spin spin coupling alongwith exar Differentiate: Proton and C13 NMR spectroscopy Hard source and soft source 	nple.	06 05 05	
Q.7	(a) (b) (c)	Explain various factors affecting fluorescence inte Write a brief note on applications of flourimetry. Explain principle and applications of atomi spectrophotometry.	nsity. c absorption and emission	06 05 05	