Seat No.: _____ Enrolment No. _____

GUJARAT TECHNOLOGICAL UNIVERSITY B.PHARM - SEMESTER - 2 EXAMINATION - SUMMER -2019

Subject Code: 2220002 Date: 31-05-2019

Subject Name: Pharmaceutical Chemistry-II (Physical Chemistry)

Time: 10:30 AM TO 01:30 PM Total Marks: 80

Instructions:

- 1. Attempt any five questions.
- 2. Make suitable assumptions wherever necessary.
- 3. Figures to the right indicate full marks.

Q.1	(a)	What is order of a reaction? Derive integrated rate equation for zero order reaction.	06
	(b) (c)	Explain the collision theory with its limitations. Write a note on Parachor.	05 05
Q.2	(a)	Define quantum efficiency. Explain in detail causes of high & low quantum yield with examples.	06
	(b) (c)	Write a note on One and two component systems. What is the basic principle of Joule –Thomson effect?	05 05
Q.3	(a) (b)	Define: Thermodynamics. Explain first law of thermodynamics. Write a note on partition coefficient. Why benzoic acid forms dimer in benzene but not in water.	06 05
	(c)	Explain the laws of photochemistry.	05
Q.4	(a)	Write a note on homogeneous and hetrogeneous catalyst with suitable examples.	06
	(b) (c)	What is Raoult's law? Describe the deviations from Raoult's law. Define: surface tension. What are its units? Explain drop weight method for the determination of the surface tension.	05 05
Q.5	(a) (b) (c)	Define: Adsorption. Explain Freundlich Adsorption isotherms. Define: Molarity and Normality. Explain state and limitation of Henry's law. Write a note on Jablonski diagram.	06 05 05
Q. 6	(a) (b)	What is enthalpy? How enthalpy of chemical reaction can be calculated? Define: Colligative property. Describe in detailed lowering of the Vapour pressure.	06 05
	(c)	Explain "Beer's Lambert's law of photochemistry".	05
Q.7	(a)	Write a note on Debye-Huckle theory.	06
	(b) (c)	Write a note on "Acid- Base Enzyme Catalysis". Describe different methods to determine the order of reaction in brief	05 05
