

Seat No.: \_\_\_\_\_ Enrolment No.\_\_\_\_

## **GUJARAT TECHNOLOGICAL UNIVERSITY** B.PHARM - SEMESTER - 2- EXAMINATION -WINTER - 2018

Subject Code:2220002 Date: 13/12/2018

**Subject Name: Pharmaceutical Chemistry-II (Physical Chemistry)** 

Time:02:30 PM TO 05:30 PM **Total Marks: 80** 

## **Instructions:**

1. Attempt any five questions.

- 2. Make Suitable assumptions wherever necessary.
- 3. Figures to the right indicate full marks.

Q.1	(a) (b) (c)	What is Refractive Index? Give its importance in Pharmacy. Enlist the factors affecting refractive index.  Define: Viscosity and fluidity. Describe principle of Ostwald's viscometer.  Discuss the measurement of surface tension by drop formation method.	06 05 05
Q.2	(a) (b) (c)	Define: Fluorescence, Phosphorescence and Chemiluminescence.  Draw a Jablonski diagram & explain consequences of Light absorption.  Define: photochemistry. Explain Lambert – Beer law.	06 05 05
Q.3	(a) (b) (c)	Explain in detail Debye-Huckel theory. Define & Enlist Colligative Properties. Explain any two in detail. Define: Molarity and Normality. Explain Henry's law in brief.	06 05 05
Q.4	(a) (b) (c)	State & explain first law of Thermodynamics with various modifications. Enlist various methods for the determination of order of reaction. Define enthalpy. How enthalpy of a chemical reaction can be calculated?	06 05 05
Q.5	(a) (b) (c)	Differentiate between: (i) Adsorption and Absorption (ii) Physical adsorption and Chemical adsorption.  Explain in detail Langmuir Adsorption isotherms.  Discuss applications of adsorption in pharmacy.	06 05 05
Q. 6	(a) (b) (c)	State and explain phase rule. Describe phase diagram of water.  Derive Reaction rate constant & half-life for first order reaction kinetics.  Write a note on "The Carnot Cycle".	06 05 05
Q.7	(a) (b)	State and explain Raoult's law of dilute solution. Discuss deviation of real solution from the Law.  Write a note on Parachlor.	06 05
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Discuss Acid-base Enzyme Catalysis in detail.