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GUJARAT TECHNOLOGICAL UNIVERSITY

B.PHARM – SEMESTER – 4- EXAMINATION –WINTER - 2018			
Subj	ect N	Code:2240004 Date: 10/12/2018 Fame: Pharmaceutical Chemistry - VI (Organic Chemistry - II) Fig. 30 PM TO 05:30 PM Total Marks: 80	
Instr	uction	s:	
	1.	Attempt any five questions.	
	2.	Make Suitable assumptions wherever necessary.	
	3.	Figures to the right indicate full marks.	
Q.1	(a)	Explain the following synthesis with reaction mechanism.	06
•	. ,	1. Fischer's indole synthesis, 2. Skraup Quinoline synthesis.	
	(b)	Define stereoselective and stereospecific reactions. Explain with suitable	05
	` /	examples.	
	(c)	Enumerate different derivatives of carboxylic acids. How are they prepared from	05
	` '	carboxylic acids?	
Q.2	(a)	Comment on following.	06
		1. Pyridine is less basic than aliphatic amines.	
		2. Thiophene is more basic than furan.	
		3. Trichloroacetic acid is more acidic than acetic acid.	
	(b)	Explain Hell Wolhard Zelinsky reaction with mechanism.	05
	(c)	Write a note on Riemer Tiemann reaction.	05
Q.3	(a)	Explain following terms giving suitable examples.	06
		1. Configuration, 2. Chiral centre, 3. Mesomer,	
		4. Resolution, 5. Optical rotation, 6. Geometric isomers	
	(b)	How is phenol prepared? Give any two methods.	05
	(c)	Write a note on conformational isomers of cyclohexane.	05
Q.4	(a)	Give 1 preparation and 1 reaction of Thiophene, pyrrole, pyridine.	06
	(b)	Enumerate methods for preparation of carboxylic acid. Explain any two methods	05
		from them.	
	(c)	What is diazonium salt? Give their preparation and reactions.	05
Q.5	(a)	Give structure of following compounds.	06
		1. Pyrazine 2. Furan 3. Isooxazole	
	4 \	4. Pyrimidine 5. Indole 6. Pyridine	
	(b)	Define nucleophilic aromatic substitution reaction. Explain benzyne mechanism.	05
0 ((c)	Write about application of nanochemistry in pharmacy.	05
Q. 6	(a)	Define green chemistry. Explain the principles of green chemistry with suitable	06
	(1.)	examples.	0.5
	(b)	Write reaction and mechanism of aldol condensation.	05
	(c)	Give structure of following compounds.	05
		1. Phenyl acetic acid, 2. para toluidine, 3. Benzoyl chloride	
0.5	(.)	4. Malonic acid, 5. Phthalic anhydride.	0.0
Q.7	(a)	Describe the principle of microwave synthesis. Explain applications of	06
	(b)	microwave synthesis in chemistry.	05
	(b)	Define α,β -unsaturated carbonyl compounds. Explain Michael addition reaction in detail.	05
	(c)	Explain Hoffmann degradation of amide in detail.	05
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