

Code No. 6157/Non-CBCS

FACULTY OF PHARMACY

B. Pharmacy 3/4 I-Sem. (Non-CBCS) (Backlog) Examination, January 2020

Subject: Physical Pharmacy - I

Time: 3 Hours Max. Marks: 70 Note: Answer all questions. All questions carry equal marks. 1. (a) Write and explain the postulates of the kinetic molecular theory. 7 (b) Write the importance of polymorphism with examples and its applications. OR 2. (a) What is phase rule? Explain the phase diagram for two component systems. 9 (b) Write a note on Differential Scanning Calorimetry with applications. 5 3. (a) Define i) Specific heat & Latent heat ii) Enthalpy iii) Entropy/ 6 (b) State and explain first and second law of thermodynamics with applications. 8 OR 4. (a) Define and explain Hess's law of heat summation and its application. 8 (b) Explain Gibbs free energy and applications. 5. (a) What are colligative properties? Explain suitable colligative properties for molecular weight determination. 8 (b) Discuss the modern theory of strong electrolytes and Debye-Huckel theory. 6 6. (a) Explain Arrhenius theory of electrolytic dissociation. 5 (b) Derive the equation for determination of acidity & basicity constant and write its 9 use. 7. (a) Define buffer. Derive buffer equation for weak acids. 4 (b) What are the methods for adjustment of tonicity? Explain freezing point depression method for adjusting isotonicity. 10 8. (a) Derive Henderson-Hasselbalch buffer equation for a weak acid and its salt. 7 (b) Write a note on pharmaceutical buffers and their preparation. Influence of pH on tissue irritation. 7 9 9. (a) Explain different types of electrodes . (b) How to measure the EMF of cells. 5 OR 10. (a) Draw and explain Daniel cell. 5 (b) Write a note on: (i) Catalysis and Catalyst (ii) Promoters and Inhibitors (iii) Applications of Redox potentials in pharmacy. 9