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GUJARAT TECHNOLOGICAL UNIVERSITY

B. Pharm. - SEMESTER-3 • EXAMINATION – SUMMER -2018

Subject Code: 230001 Date: 28/04/2018

Subject Name: Physical Pharmaceutics - II

Time: 02:30 PM TO 05:30 PM Total Marks: 80

Instructions:

1. Attempt any five questions.

- 2. Make suitable assumptions wherever necessary.
- 3. Figures to the right indicate full marks.

Q.1	(a) (b) (c)	Describe the methods for determination of boiling point elevation. Describe osmotic pressure as a colligative property. Define Raoult's law. Explain positive and negative deviations from Raoult's law.	06 05 05
Q.2	(a) (b) (c)	Define: molarity, normality, molality, osmolality and vapor pressure. Describe Debye-Heckel theory for strong electrolytes. Describe Arrhenius theory of electrolytic dissociation.	06 05 05
Q.3	(a) (b) (c)	Describe properties of solutions of electrolytes. Explain Faraday's law. Derive equations for first order reaction & its half-life.	06 05 05
Q.4	(a) (b) (c)	Write a short note on accelerated stability study. Describe the factors which govern the rate of a chemical reaction. Discuss various methods to determine the order of reactions.	06 05 05
Q.5	(a) (b) (c)	Write applications of complexes in pharmacy. Write note on protein binding. Give classification of complexes.	06 05 05
Q. 6	(a) (b) (c)	Write in short on metal complexes. Give detailed classification of polymers. Write a note on hydrogel drug delivery system.	06 05 05
Q.7	(a) (b) (c)	Describe pharmaceutical applications of polymers. Explain Fick's first law of diffusion. Describe the USP type-II dissolution apparatus with labelled diagram.	06 05 05
