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## GUJARAT TECHNOLOGICAL UNIVERSITY <br> B. Pharm. - SEMESTER-3 • EXAMINATION - SUMMER -2018

## Subject Code: 230001 <br> Subject Name: Physical Pharmaceutics - II Time: 02:30 PM TO 05:30 PM

Date: 28/04/2018

## Instructions:

1. Attempt any five questions.
2. Make suitable assumptions wherever necessary.
3. Figures to the right indicate full marks.
Q. 1 (a) Describe the methods for determination of boiling point elevation. ..... 06
(b) Describe osmotic pressure as a colligative property. ..... 05
(c) Define Raoult's law. Explain positive and negative deviations from ..... 05
Raoult's law.
Q. 2 (a) Define: molarity, normality, molality, osmolality and vapor pressure. ..... 06
(b) Describe Debye-Heckel theory for strong electrolytes. ..... 05
(c) Describe Arrhenius theory of electrolytic dissociation. ..... 05
Q. 3 (a) Describe properties of solutions of electrolytes. ..... 06
(b) Explain Faraday's law. ..... 05
(c) Derive equations for first order reaction \& its half-life. ..... 05
Q. 4 (a) Write a short note on accelerated stability study. ..... 06
(b) Describe the factors which govern the rate of a chemical reaction. ..... 05
(c) Discuss various methods to determine the order of reactions. ..... 05
Q. 5 (a) Write applications of complexes in pharmacy. ..... 06
(b) Write note on protein binding. ..... 05
(c) Give classification of complexes. ..... 05
Q. 6 (a) Write in short on metal complexes. ..... 06
(b) Give detailed classification of polymers. ..... 05
(c) Write a note on hydrogel drug delivery system. ..... 05
Q. 7 (a) Describe pharmaceutical applications of polymers. ..... 06
(b) Explain Fick's first law of diffusion. ..... 05
(c) Describe the USP type-II dissolution apparatus with labelled diagram. ..... 05
