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[LD 805]

OCTOBER 2013

Sub. Code: 3805

DOCTOR OF PHARMACY (PHARM. D) DEGREE EXAMINATION

FIRST YEAR

PAPER V – PHARMACEUTICAL INORGANIC CHEMISTRY

Q.P. Code: 383805

Answer All questions

Maximum: 70 marks

 $(2 \ge 20) = 40$

 $(10 \times 3 = 30)$

I. Elaborate on:	I.	Elaborate	on:
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Time: Three Hours

- 1. a) Describe the principle and procedure involved in the limit test for Iron.
 - b) Explain the procedure for the assay of oxygen and carbon dioxide.
- 2. a) Explain the theory of redox titration.
 - b) Explain the various concepts of Acid-base.
 - c) Give an account on Neutralization curve.

II. Write notes on:

- 1. How will you prepare and standardize 0.05 M disodium EDTA.
- 2. Explain types of solvent used in non aqueous titration.
- 3. Describe the principle involved in Mohr's method
- 4. What are the applications of buffer solution in pharmacy.
- 5. Write a note on the assay and uses of aluminium hydroxide gel.
- 6. Explain the physiological role of copper & selenium.
- 7. What are cathartics? Give an example.
- 8. Explain Geiger-Muller counter
- 9. What are antacids? Classify them with examples.
- 10. Write molecular formula and uses of the following:
 - a) Potassium Iodide.
 - b) Sodium Fluoride.
 - c) Ferrous fumarate.
