

Water And Electrolytes

Balance

And

Imbalance

In Human Body

Body Water

- Water is the **chief constituent of human body.**
- Water is the **chief solvent of body.**
- Water **comprises 60-70%** of total body weight
- **Human body cannot exist without Water the.**

Sources Of Body Water

Exogenous Sources Of Water

- **Drinking Water**
,Beverages -1000-1500 ml
- **Water from Cooked Foods**

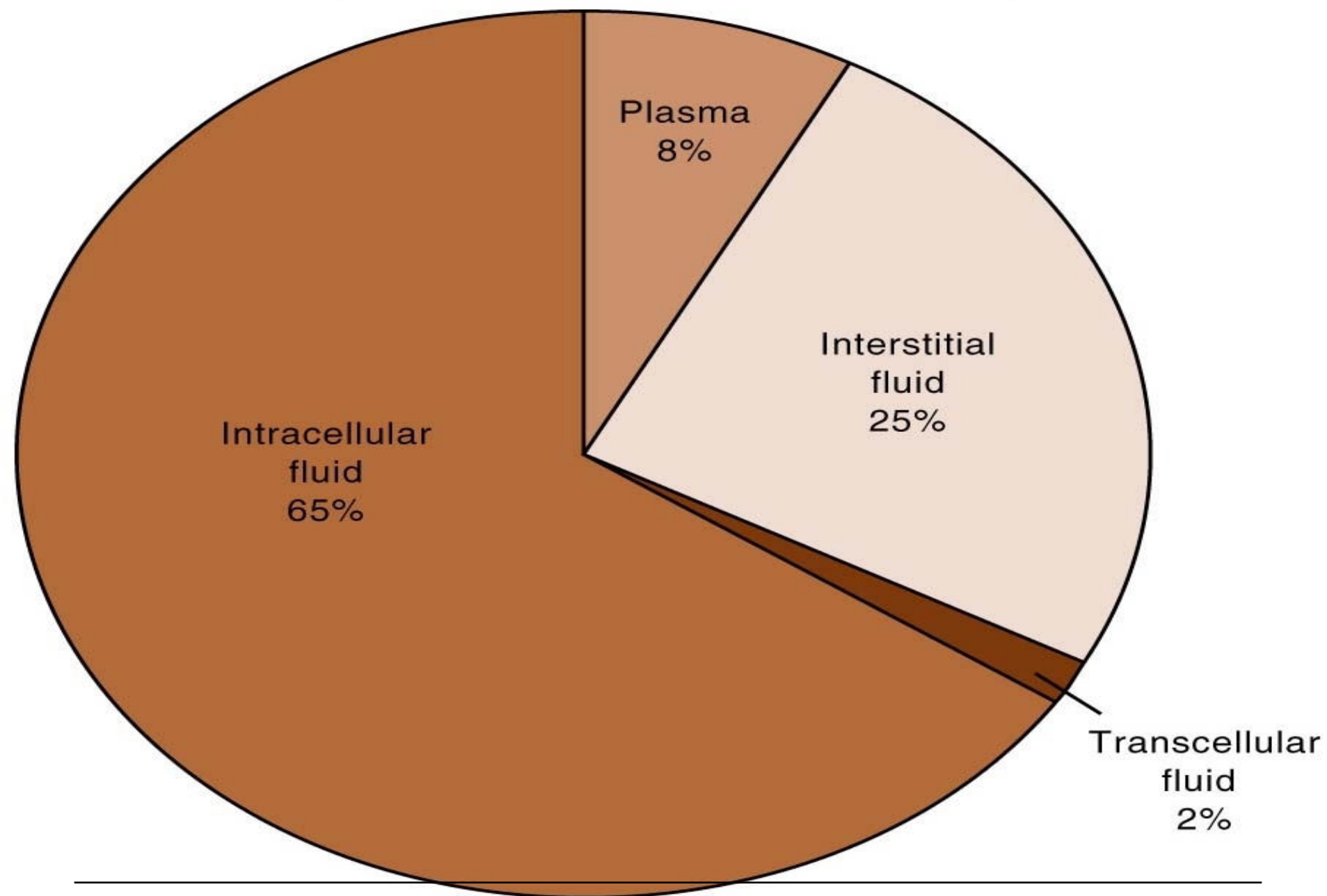
- **Water intake** through mouth is highly variable 1-5 Litres this depend on :
 - **Social habits**
 - **Climatic condition**

Endogenous Sources Of Water

- **Metabolic Water - 400 ml**
- Produced during metabolism oxidation of food substances.
(At end of ETC Process)

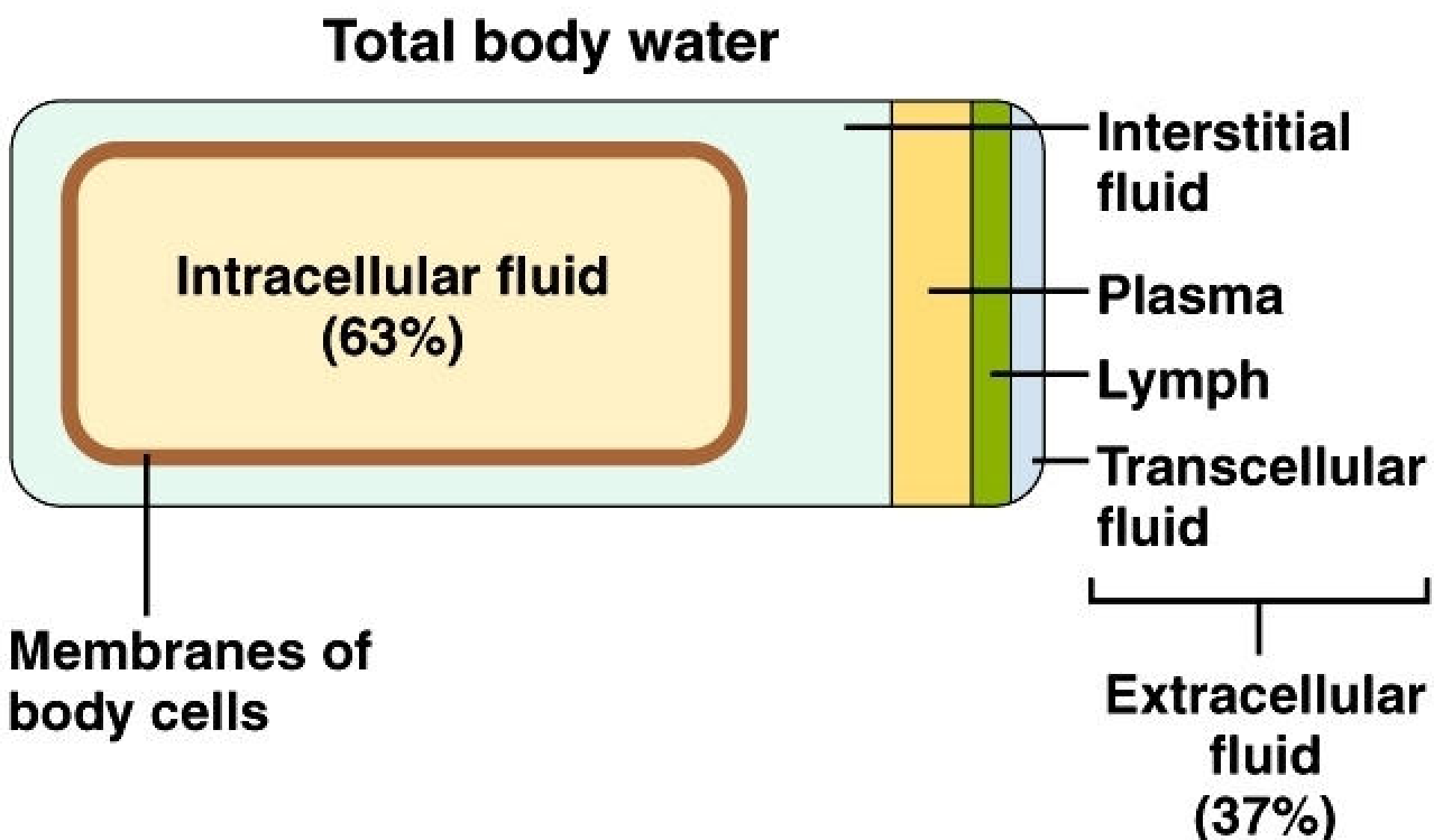
Distribution Of Body Water

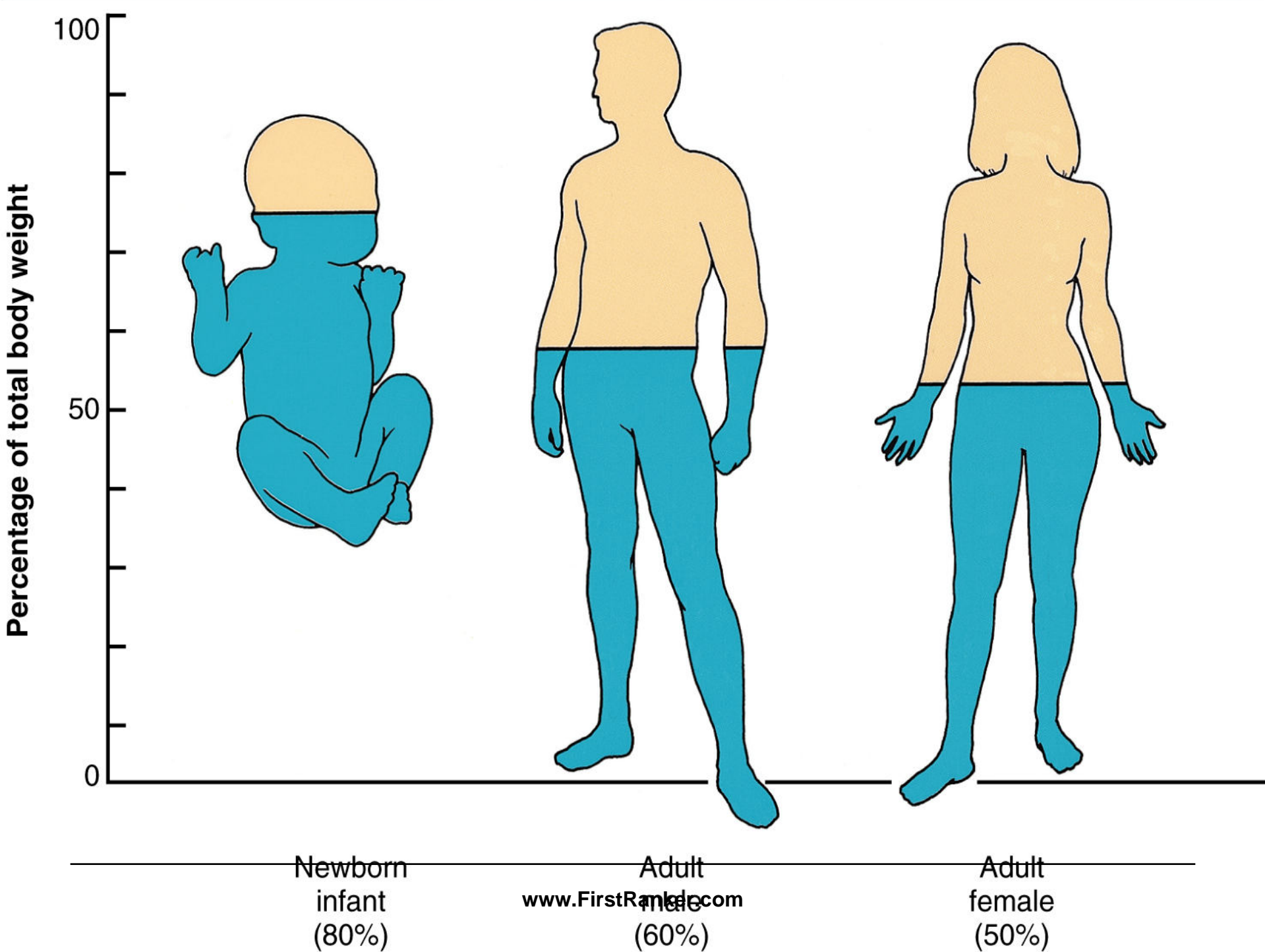
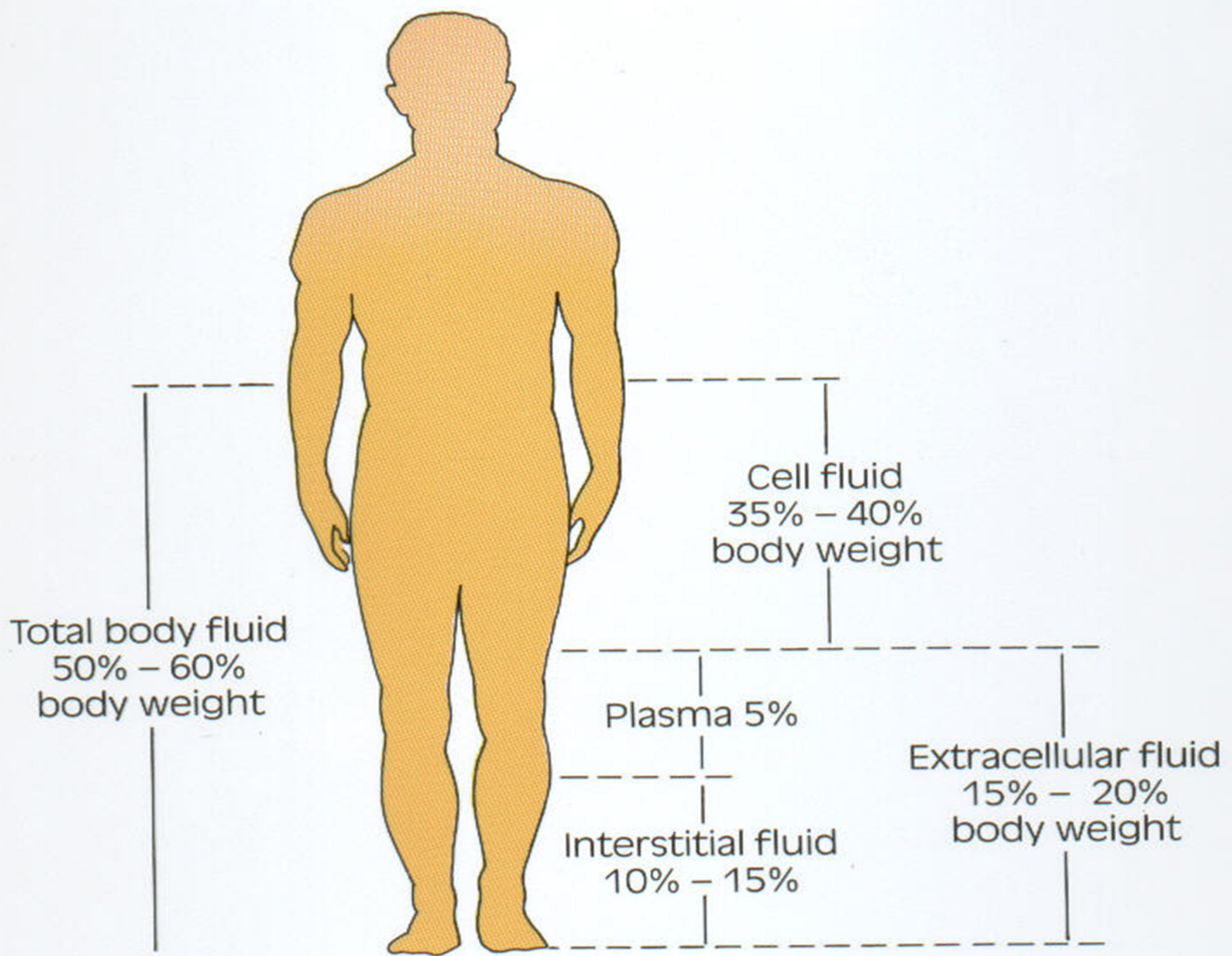
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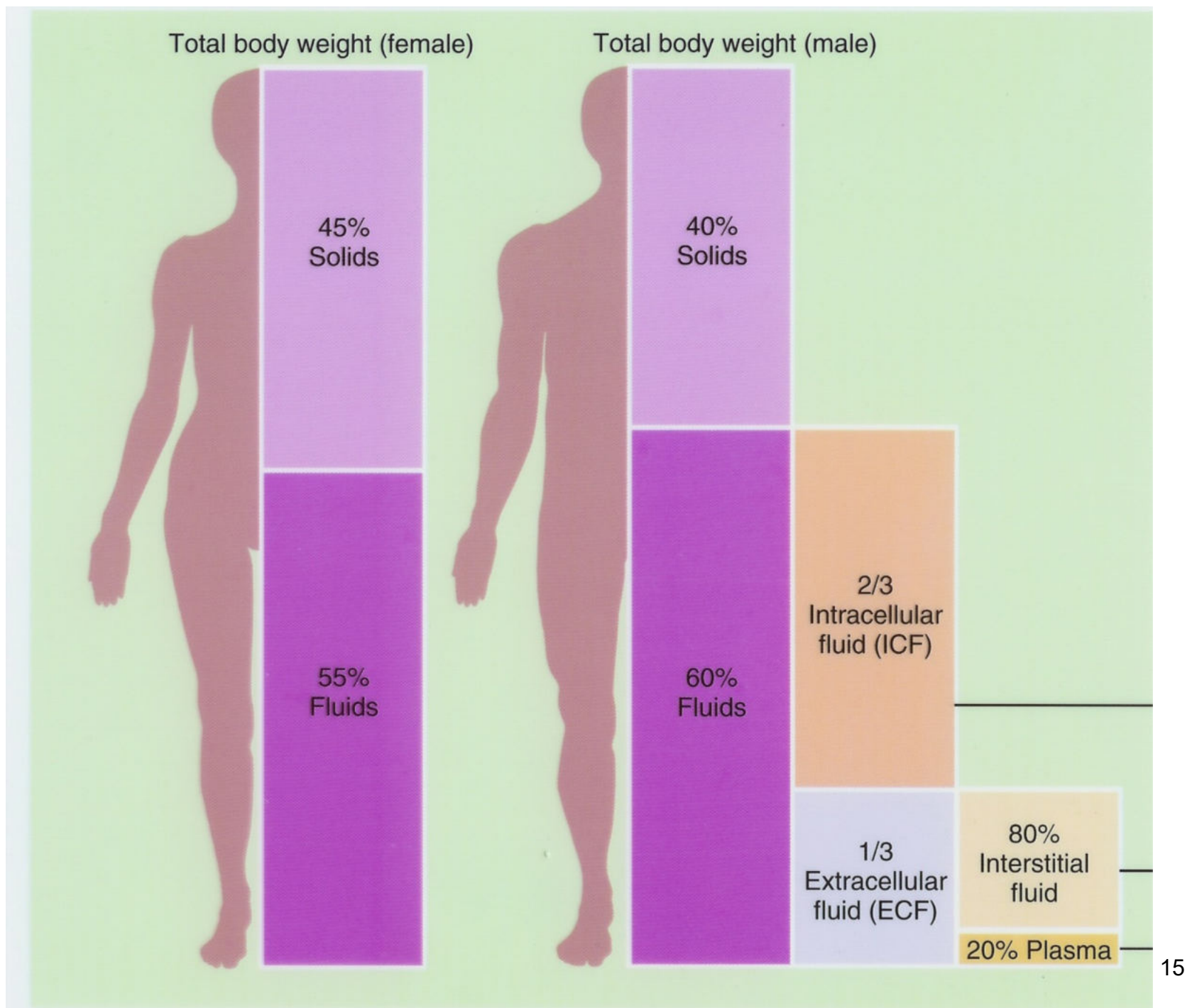


- In an adult of 70 kg body
- **Total Body Water -60- 70% /36-49 Lt**
- **Intracellular Fluid -65 % - 35 L**
- **Extracellular Fluid -35% -14 L**
 - **Interstitial Tissue Fluid -25% -11L**
 - **Plasma /Intra Vascular Fluid -8% -3L**
 - **Transcellular Fluid- 2%**

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• **Body water content in percentage of a body weight is **lowest** in.**

(A) Well built man

(B) Fat woman

(C) Well nourished child

(D) Fat Man

Functions Of Body Water

- Involved in Biochemical reactions
 - Water act as reactant in many hydration
Hydrolytic reactions of metabolic pathways.
- Transporting media of body:
 - Transportation of nutrients and waste metabolites through aqueous media of blood and tissue floods.
- Regulates body temperature

- Water transports **Hormones**, Enzymes, **blood platelets**, and red and white blood cells
- Water act as a solvent for **Electrolytes** and **Non electrolytes**
- Water Facilitates **Digestion** and promoting **Elimination of ingested food**
- Water serve as a tissue **Lubricant**

Body Water Input and Output

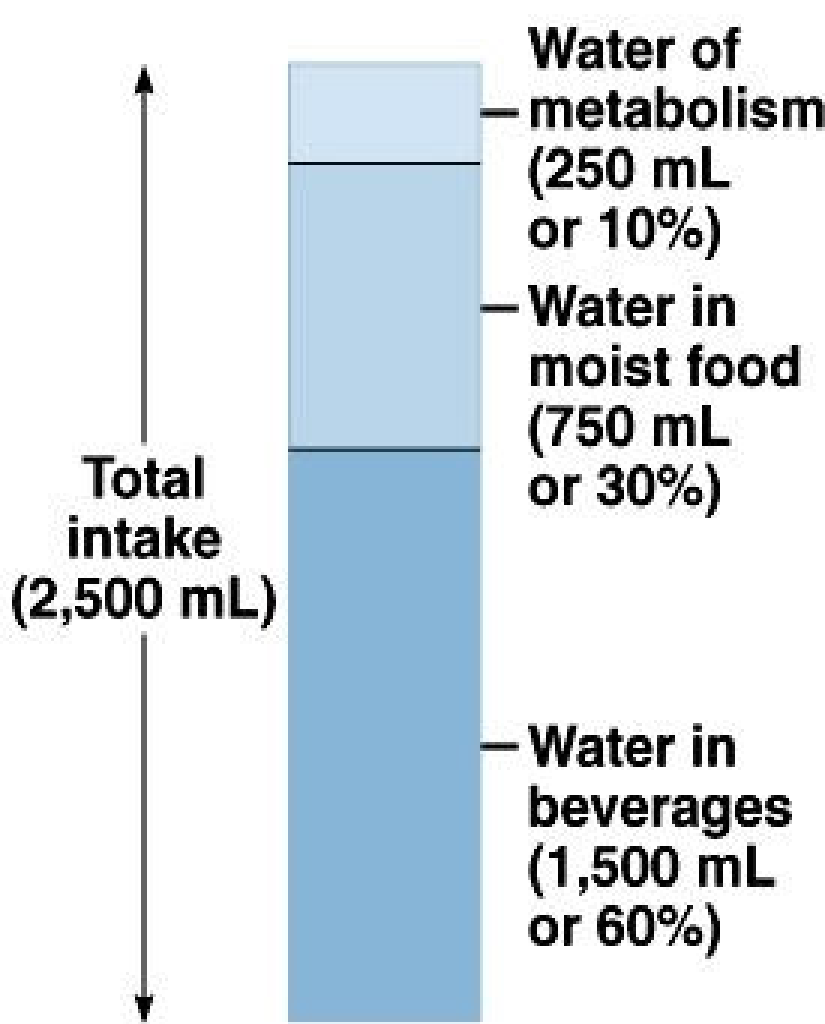
Body Water Input

- Body can gain water by
 - **Ingestion of liquids and moist foods (2300mL/day)**
 - **Metabolic synthesis of water during cellular respiration (200mL/day)**

Body Water Output

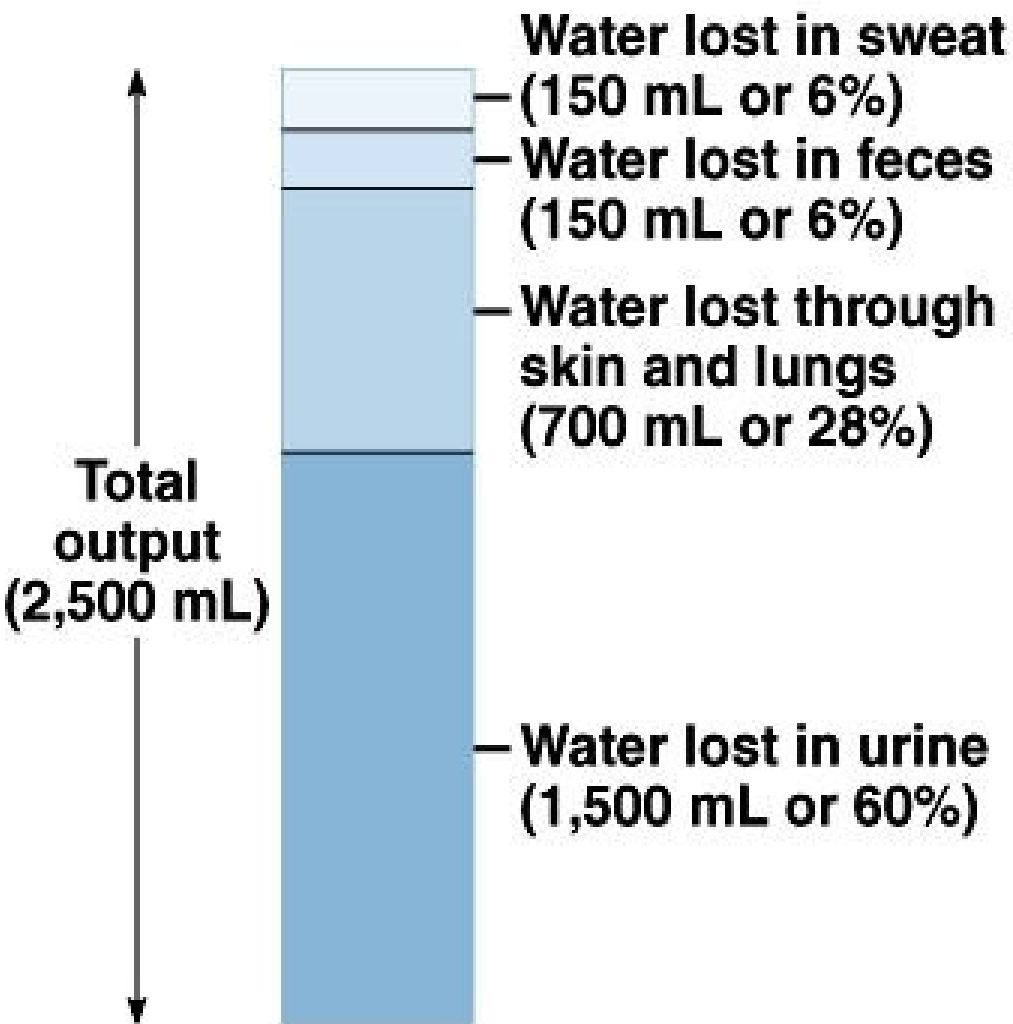
- Body losses water through:
 - **Kidneys (1500mL/day)**
 - **Evaporation from Skin (600mL/day)**
 - **Exhalation from Lungs (300mL/day)**
 - **Feces (100mL/day)**

Average daily intake of water



(a)

Average daily output of water



(b)

BODY ELECTROLYTES

What Are Electrolytes?

- Substance when dissolved in solution **dissociates into ions**
- These ions are able to **carry an electrical current**
- **An Electrolyte** is a substance
- Which **develops an electrical charge when dissolved in water**

Body Electrolytes

- Salts like **NaCl** and **KCl** in aqueous solutions gets dissociated to
- **Charged ions Na^+ and Cl^- called as Electrolytes.**
- The **concentration of these Electrolytes is expressed** as **mEq/L.**

Types Of Electrolytes

- **CATION** - Positively charged Electrolyte
- **ANION** - Negatively charged Electrolyte

- **Water molecules completely surround these dissociated ions**
- These **prevents union of Cations and Anions.**

Distribution Of Body Electrolytes In ECF and ICF

ELECTROLYTES IN BODY FLUID COMPARTMENTS

INTRACELLULAR Electrolytes	EXTRACELLULAR Electrolytes
POTASSIUM	SODIUM
MAGNESIUM	CHLORIDE
PHOSPHOROUS	BICARBONATE

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To Maintain Electrical Neutrality In
Each Fluid Compartments

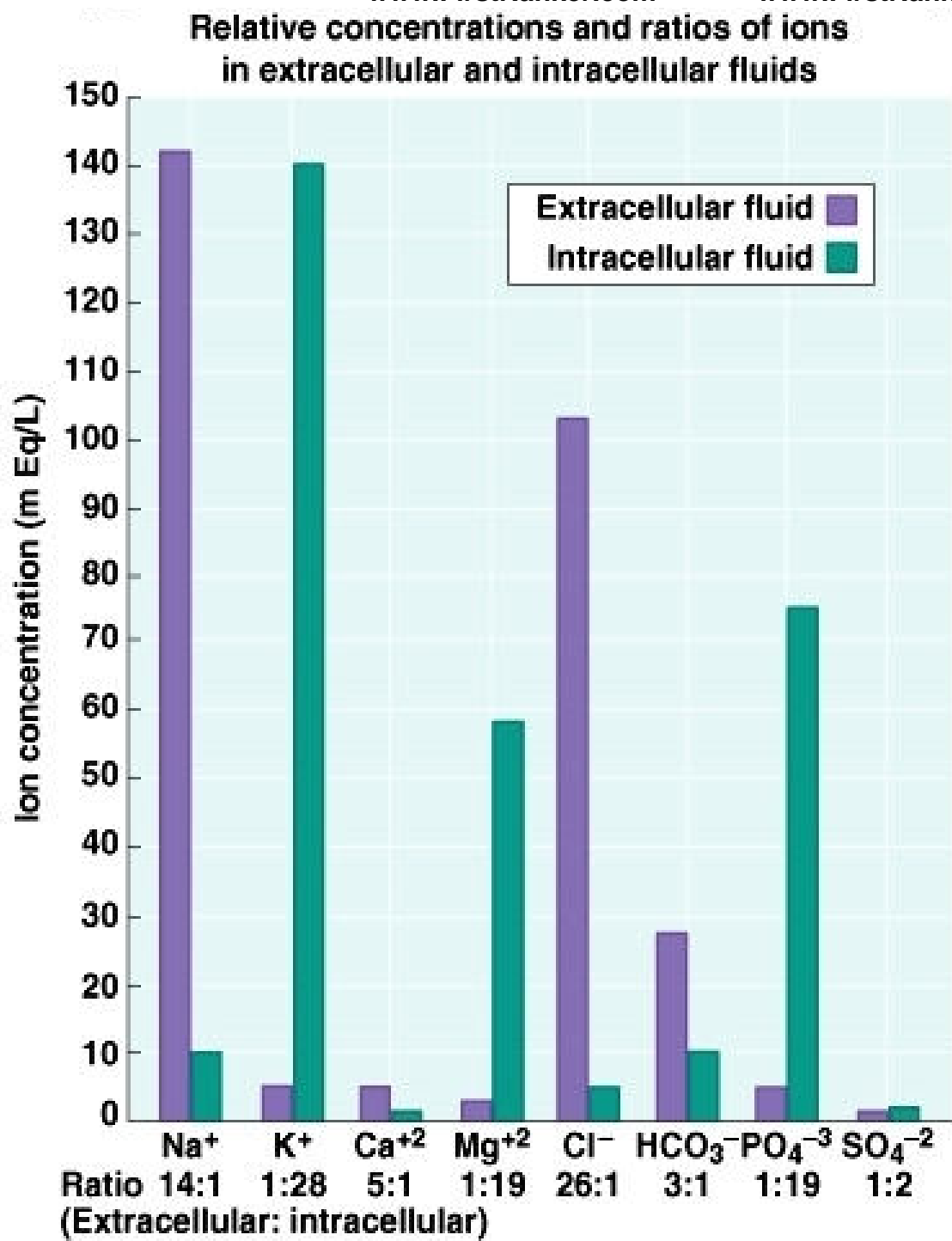
Number Cations =Number Anions

ECF Cations	ECF Anions
Na ⁺ (140 mEq/L)	Cl ⁻ (103 mEq/L)
K ⁺	HCO ₃ ⁻
Ca ⁺	HPO ₄ ⁻⁻
Mg ⁺	SO ₄ ⁻⁻
Total Cations 155 mEq/L	Total Anions 155 mEq/L

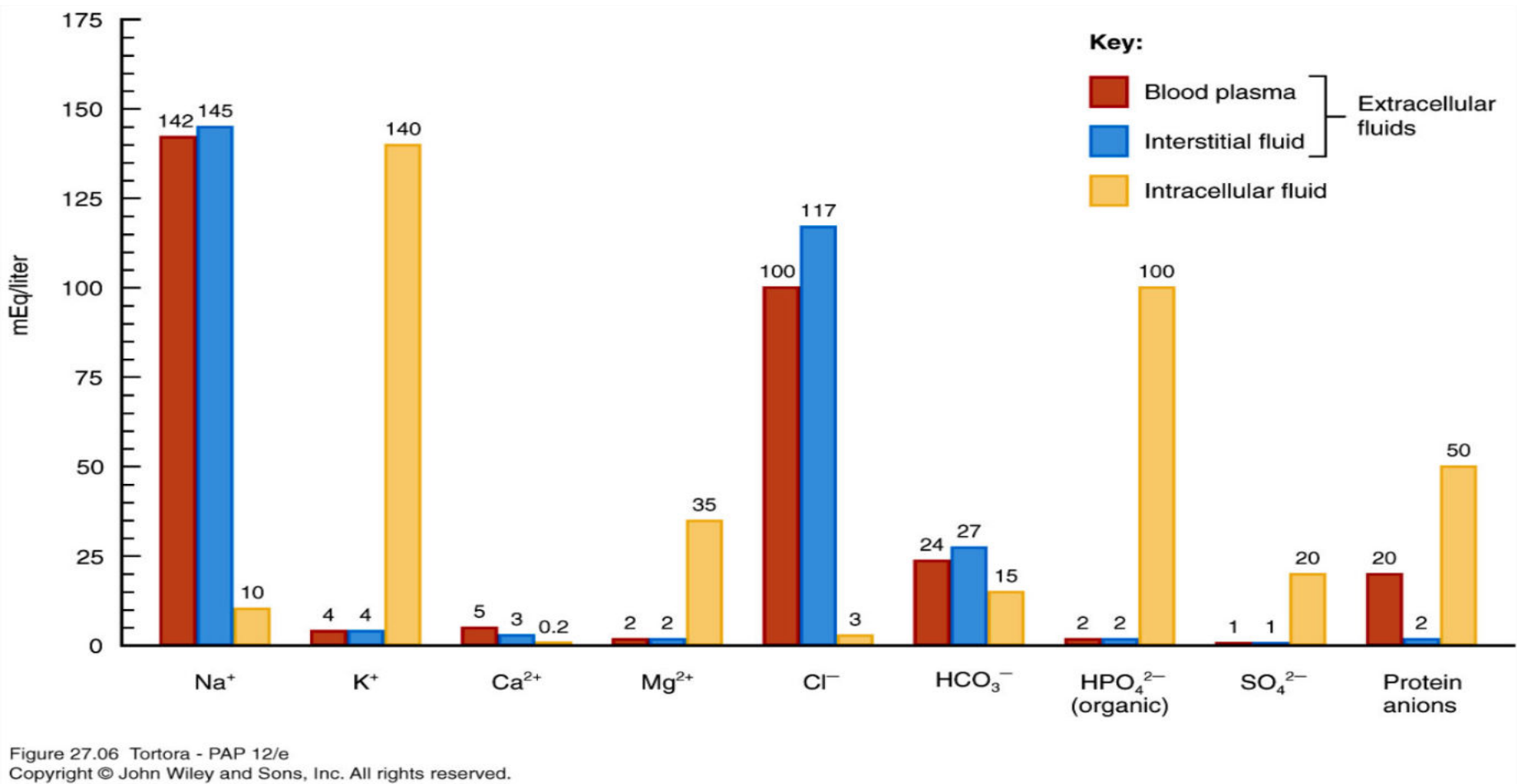
- Predominant Cations and Anions of ECF:
- Na⁺ and Cl⁻ respectively.

ICF Cations	ICF Anions
Na ⁺	Cl ⁻
K ⁺ (150 mEq/L)	HCO ₃ ⁻
Ca ⁺	HPO ₄ ^{- -} (140 mEq/L)
Mg ⁺	SO ₄ ^{- -}
Total Cations 195 mEq/L	Total Anions 195 mEq/L

- Thus the predominant Cations and Anions of ICF
- **K⁺ and HPO₄^{- -} respectively.**



Electrolyte and protein anion concentrations



Functions Of Body Electrolytes

- **Electrolytes are well distributed in the body compartments.**
- Electrolytes in the medium/compartments **produce osmotic pressure.**
- This osmotic pressure helps in **maintaining water balance.**

ELECTROLYTES

- **Na⁺**: Most abundant electrolyte in the ECF.
- **K⁺**: Essential for normal membrane excitability for nerve impulse
- **Cl⁻**: Regulates osmotic pressure and assists in regulating acid-base balance
- **Ca²⁺**: Promotes nerve impulse and muscle contraction/relaxation
- **Mg²⁺**: Plays role in carbohydrate and protein metabolism, storage and use of intracellular energy and **neural transmission. Important in the functioning of the heart, nerves, and muscles**

- For more detail functions of Electrolytes
- **Refer Minerals Chapter.**

Movement of Water and Electrolytes

Diffusion – movement of particles down a concentration gradient.

Diffusion: the random movement of particles in all directions through a solution

- **Osmosis:** movement of water across a membrane from a less concentrated solution to a more concentrated solution
- **Osmosis** – diffusion of water across a selectively permeable membrane

- **Osmolarity:** The number of moles **per liter of solution**
- **Osmolality:** The number of moles **per Kg of Solvent.**
- **Sodium** and its associated ions in plasma make the **largest contribution** (90%) for plasma Osmolality.
- **Osmolality is measured by Osmometer**

- **Plasma Osmolality = $2 \times \text{Plasma Na}^+$ mmol/L**

- **Osmolality**: concentration of a solution determined by the number of dissolved particles per kilogram of water.
- **Osmolality controls water movement and distribution in body fluid compartments**

- **Active transport:** Movement of solutes across membranes;
 - Requires transporters and **expenditure of energy**
 - Movement of particles **is up a concentration gradient**
-
- **Filtration:** transfer of water and solutes through a membrane
 - From a **region of high pressure to a region of low pressure**

Normal Fluid and Electrolytes Exchanges In Body

**Water And Electrolytes Movement
INN and OUT of Cells.**

Remember

- Normally in a healthy body there **is osmotic equilibrium maintained in each compartment.**

- In a healthy body the **semipermeable cell membrane**
- **Allows only passage of Water** but not **Electrolytes** through it.
- **Disturbance in osmotic equilibrium of compartments**
- **Draws water from the compartment with lower osmotic pressure (Hypotonic)**
- **Into the compartment with higher osmotic pressure (Hypertonic)**
- **Until equilibrium is restored.**

- In the **concentration gradient of K^+ and Cl^-**
- **K^+ tends to diffuse out of the cells** and **Cl^- enters into cells.**
- During difference **in electrical potential**
- For example in relative **negativity inside the cells**
- **There tend to keep Cl^- out and K^+ inn.**

Remember

- Cells do not allow accumulation of Na^+
 - **Na rapidly enters in the cells**
 - **By the Sodium pump, Na is effectively extrudes out from ICF**
 - **By active transport of Na out of cells.**
-
- However in case of Na^+ diffusion into cells
 - It is **favored by both the concentration gradient and electrical potential.**

Remember

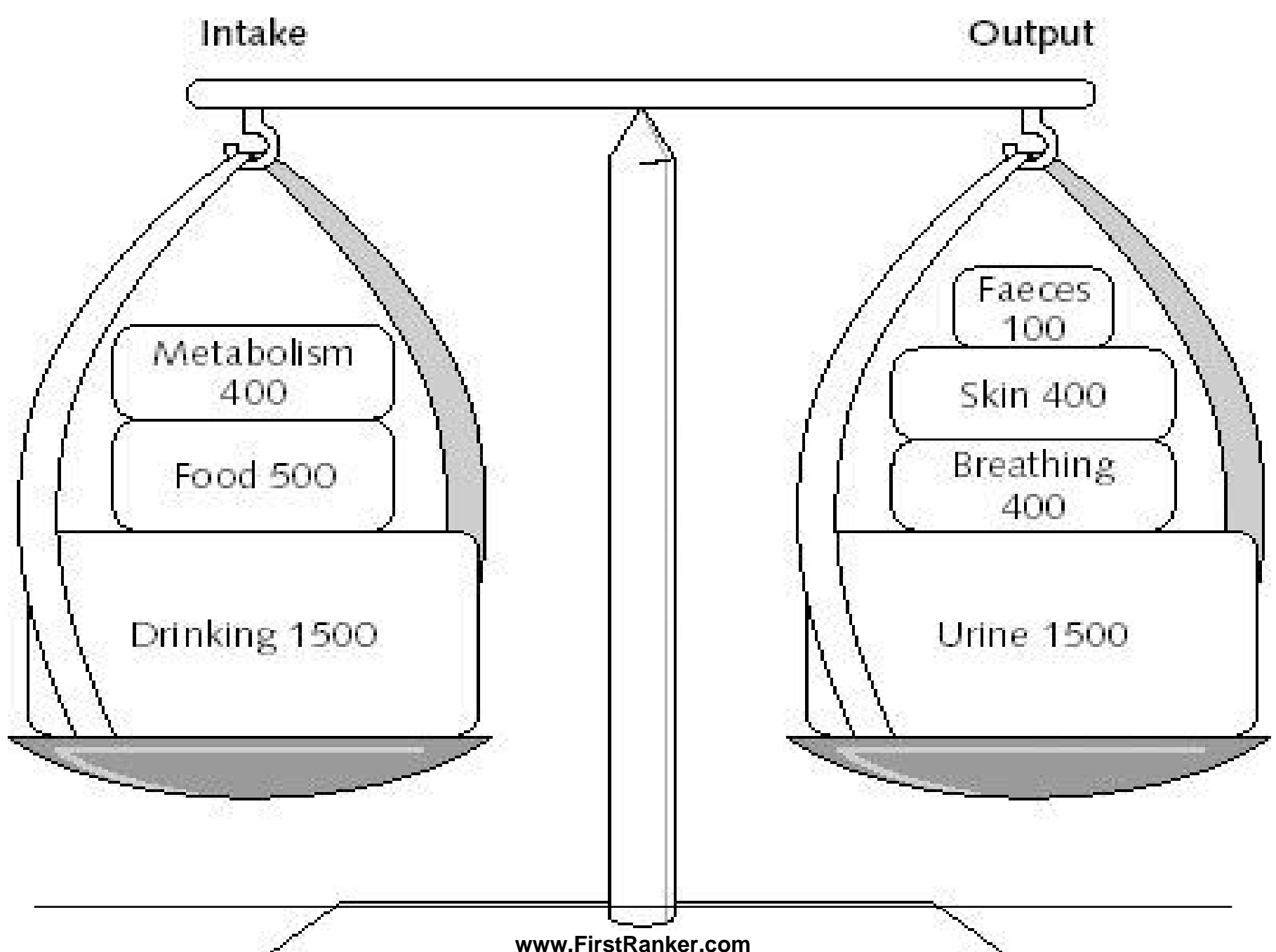
- **Where Sodium goes,
Water follows.**

**Homeostasis Of Body Water
OR
Regulation Of Body Water**

Water Balance

- An healthy adult individual always try
- **To maintain water balance** by the homeostatic mechanisms.
- Since **Water balance is vital for human body**
- A body is said to be in **water balance** In a day
- When the amount of **water intake in the body is equal**
- **To the amount of water output by that body.**

- A healthy body tries to regulate the body water
- **Proportionately distribute the water in ICF and ECF**



- **Specific concentrations of Electrolytes in the body compartments are of most important**
- **In distribution and retention of body water.**
- **In healthy state the Osmotic pressure due to Cations is equal to the osmotic pressure due to Anions.**
- **Which is predominantly due to equal concentrations of Cations and Anions in each compartment.**

Factors Regulating Water And Electrolyte Balance

- In human body **Water and Electrolytes go togetherly.**
- That means the **osmotic equilibrium** created by **Electrolytes** help in maintaining **water balance.**
- If there is **imbalance in Electrolytes** it directly affects **Water balance.**

- Since the **Electrolytes and Water balance** go hand in hand in human body.
- Hence **factors regulating the water and electrolytes are same.**

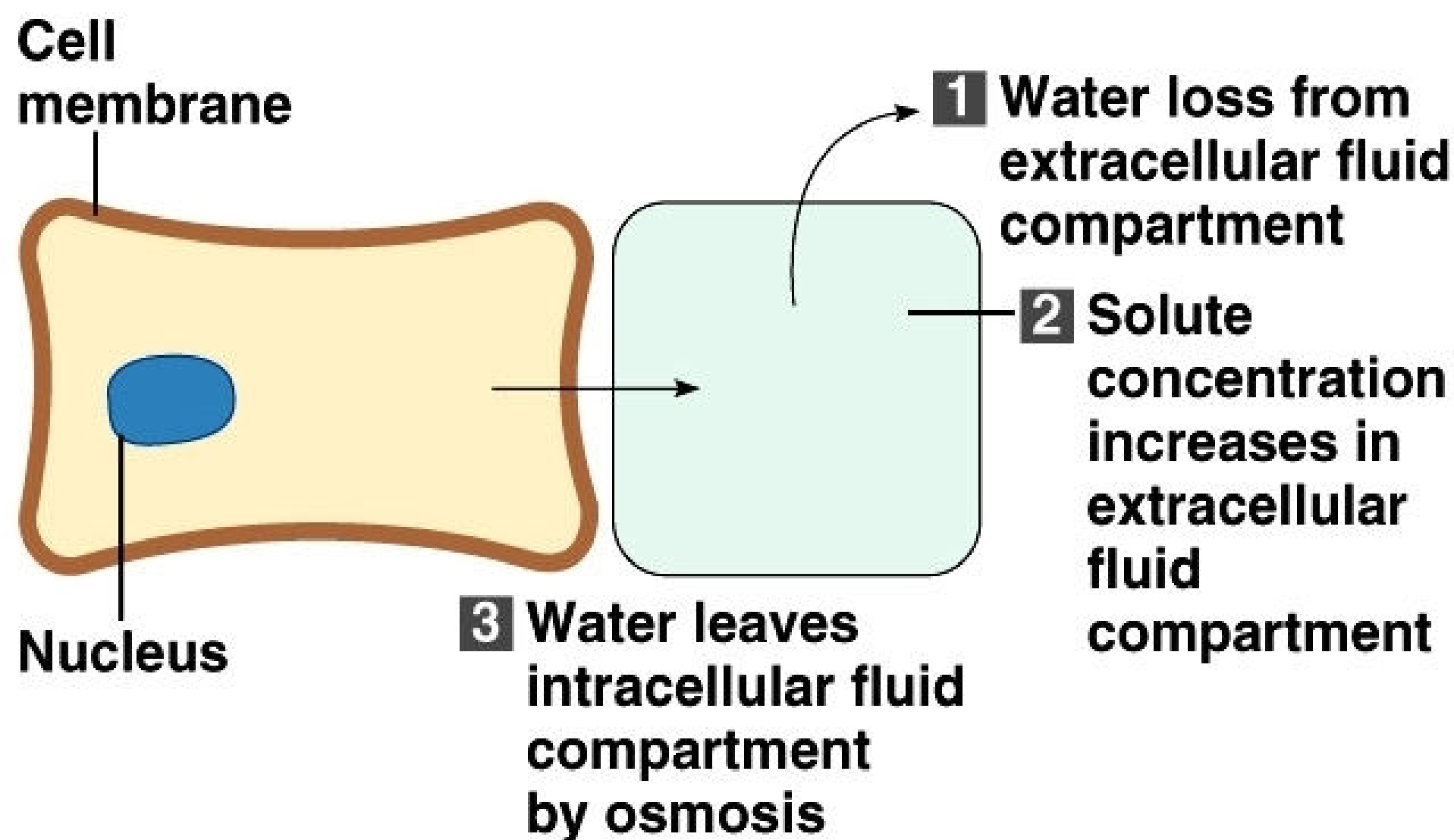
**Homeostasis of
Water And Electrolytes
Is Maintained By**

Solute Homeostasis Maintained by:

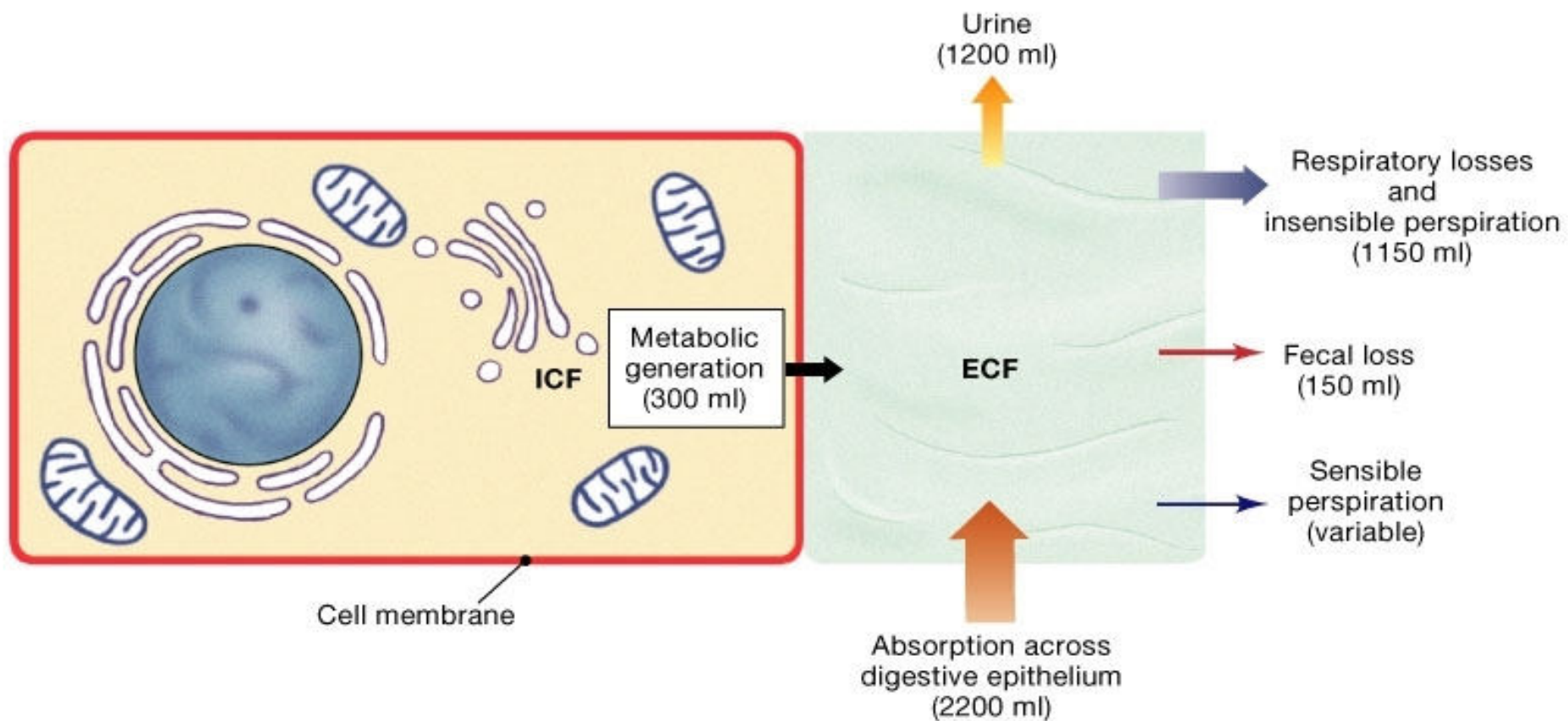
- Ion transport
- Water movement
- Kidney function

These functions act to keep body fluids:

- Electrically neutral
- Osmotically stable



Fluid Exchanges



- A body consume fluids and food items variably **depending upon habits and climatic condition.**
- **Intake of water and electrolytes is rarely proportional.**

- **Kidneys play a predominant role**
- In **regulating water and electrolyte balance** in the body normally.

–Kidneys play role to excrete
–Excess water through urine
(Dilute form of Urine)

OR

–Excess Electrolytes through urine

(Concentrated form of Urine)

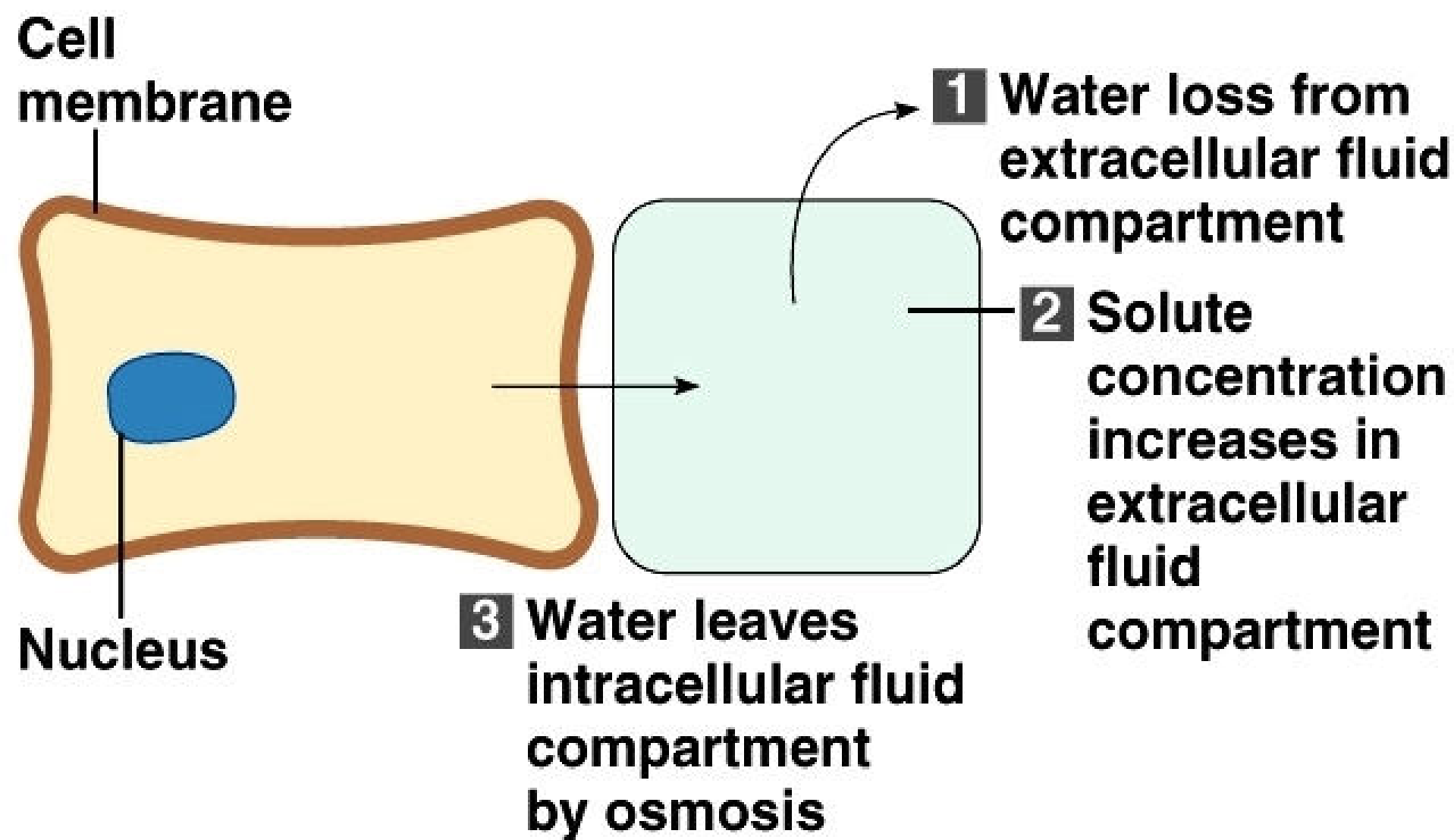
Biochemical Factors Regulating Water And Electrolyte Balance

- 1. Neural Mechanism- Thirst Mechanism**
 - 2. Antidiuretic Hormone/Vasopressin**
 - 3. Renin Angiotensin System**
 - 4. Aldosterone**
 - 5. Atrial Natriuretic Peptide(ANP)**
 - 6. Kinins (Increases Salt and Water excretion)**
-

1. Neural Mechanism/Thirst Mechanism

Regulate Low Body Water

- When the **body water** is lowered due to:
 - No intake of fluids
 - Body fluids lost through obligatory losses (Urine and Feces).
- This leads to **decrease in volume of body fluids with respect to solutes and rise in osmotic pressure.**



- The **ECF volume decreases and becomes hypertonic.**
- This **tends to draw water from ICF causing cellular dehydration.**

- The **cellular dehydration stimulates**
- **The thirst centre located in hypothalamus.**
- In response to the stimulus to thirst center
- **There occurs dryness of mouth and Pharynx .**
- Feeling of thirst makes drink water
- **Water ingested orally quench the thirst to regulate the body water.**

2. Antidiuretic Hormone/Vasopressin

Antidiuretic Hormone/Vasopressin

- Anti Diuretic Hormone(ADH) is **produced in Hypothalamus**
- **Stored in posterior pituitary gland**

- The action of Peptide Hormone ,ADH is **via cyclic AMP.**

- An increase in Osmolality (**Na⁺ conc**) of plasma
- **Promotes ADH secretion and vice versa.**

- ADH is **water conservation hormone**
- It acts on renal collecting tubule
- **For reabsorption of water by renal tubules.**

Role Of Anti Diuretic Hormone (ADH)

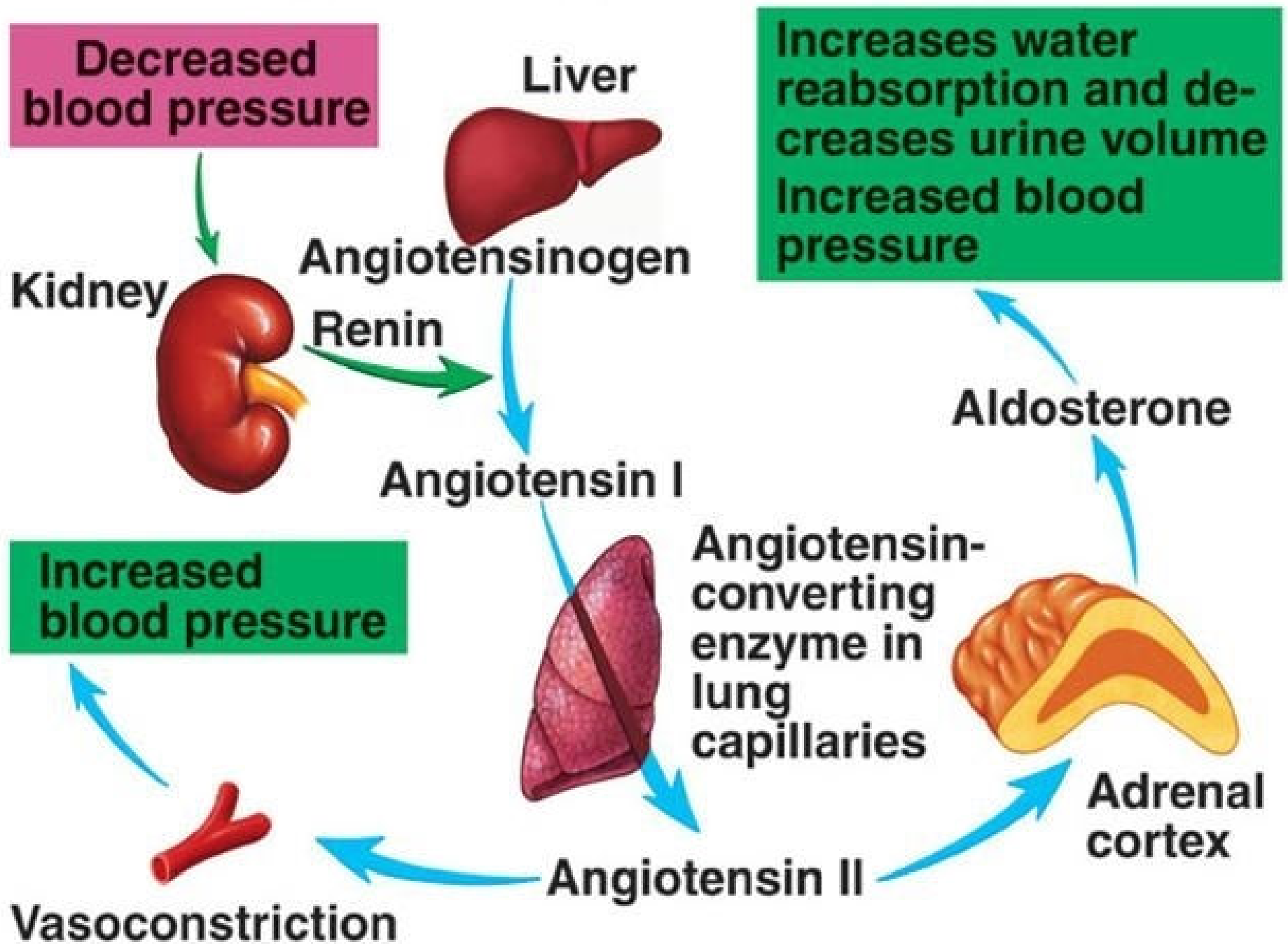
- When the **body water is depleted**
- ADH exerts **Antidiuretic effect.**

- ADH affects renal tubules
- Provides for the **facultative reabsorption of water from distal tubules.**
 - Urine output will be lower
 - Urine concentration will be increased
 - Body water will be maintained

3. RENIN ANGIOTENSIN MECHANISM

- **Renin-Angiotensin System works when the:**
 - **Blood volume is low**
 - **Blood pressure is low**

- **Kidney, Liver and Lungs are involved in Renin Angiotensin System.**



- **Renin is released by kidneys** in response to decreased blood volume

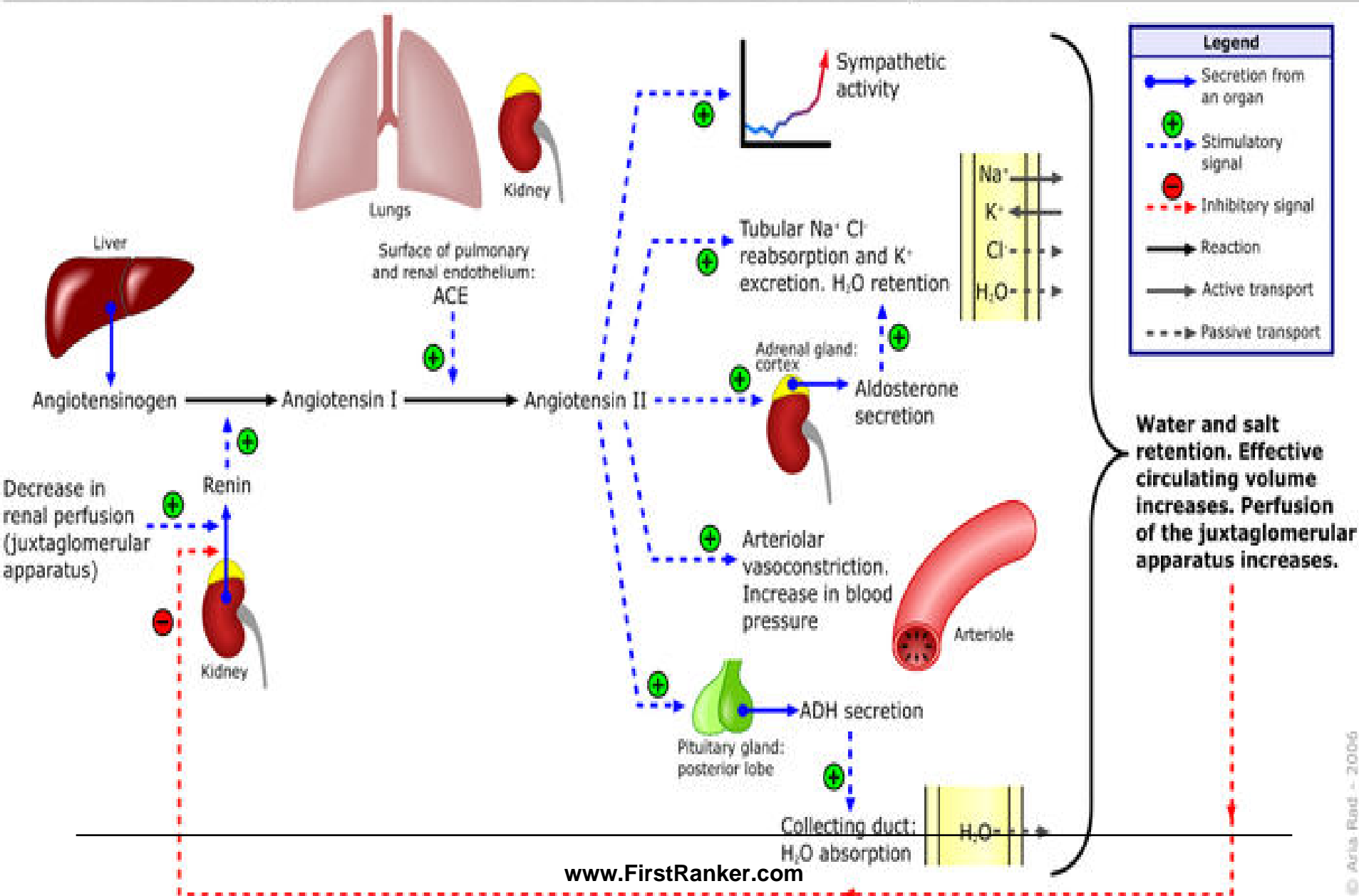
- **Renin causes Angiotensinogen (plasma protein formed in Liver) to split & produce Angiotensin I**
- **Lungs convert Angiotensin I to Angiotensin II**
- **By Angiotensin Converting Enzyme.**

- **Angiotensin II** then stimulates adrenal gland to **release Aldosterone**
- Aldosterone then **increase the peripheral vasoconstriction**
- Renin Angiotensin System **regulates Aldosterone hormone from Adrenal gland**
- During homeostatic mechanism of Water and Electrolyte Balance.

- Fall in E.C.F volume
- Decreases blood pressure
- Sensed by Juxtaglomerular apparatus of Nephrons of Kidneys to **secrete Renin**
- Renin then stimulate **Liver Angiotensinogen** to produce **Angiotensin –I**
- **Angiotensin I** is converted to **Angiotensin**
- **By Lung produced enzyme Angiotensin Converting Enzyme(ACE).**

- **Angiotensin II stimulates the release of Aldosterone from Adrenal gland.**

Renin-angiotensin-aldosterone system



4. Role Of Aldosterone

Aldosterone

- Aldosterone is a **Steroid Hormone**.
- It is a **Mineralocorticoid** produced by **Zona Glomerulus of Adrenal Cortex**.
- It has most important effect on **Mineral Metabolism**.

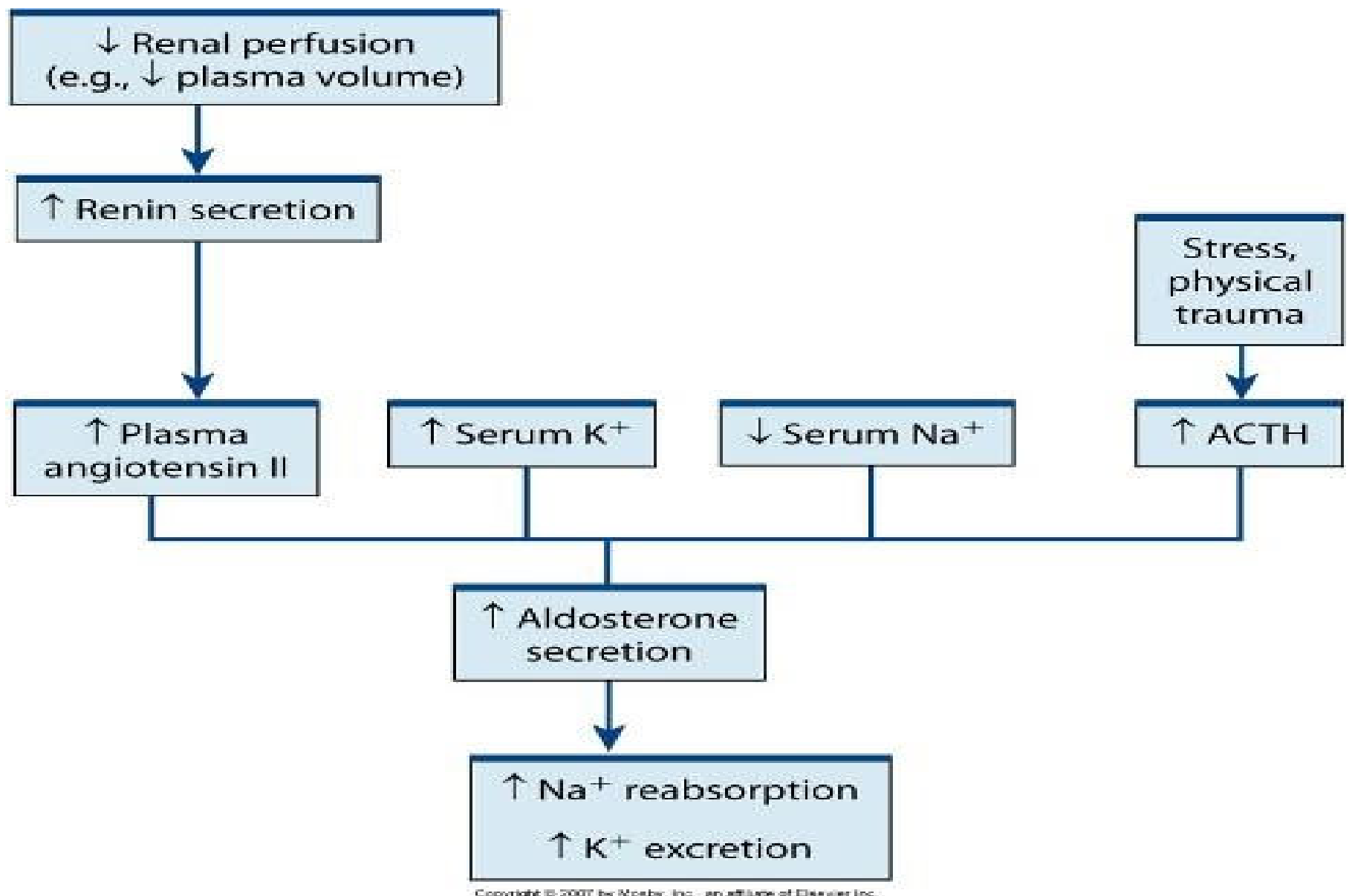
ALDOSTERONE

- Aldosterone is released as part of **Renin Angiotensin mechanism**
- Acts on renal distal convoluted tubule
- The hormone Aldosterone by its action:
 - **Increases the rate of reabsorption of Na^+ (95 %) and Cl^-**
 - **Increases K^+ loss through Urine**

Role Of Aldosterone

- Aldosterone **increases Sodium uptake from the tubular fluid**
- Regulates water reabsorption by renal tubules and add into the blood
- **Makes excretion of Potassium**
- Thus **Aldosterone maintain Water and Electrolyte Balance** by its action on renal tubules:
 - **Reabsorbs Sodium**
 - **Retains Water**
 - **Looses Potassium**

Factors Affecting Aldosterone Secretion



5. Atrial Natriuretic Peptide or Factor (ANP)/ (ANF)

Atrial Natriuretic Peptide or Factor (ANP)/ (ANF)

- ANP is a **Cardiac hormone**
- Secreted by **right Atrium of Heart.**
- Chemically ANP is a **Polypeptide hormone**

ANP

- ANP is released **in response to increased blood pressure** in the atria (due to increased blood volume)
- ANP opposes the Renin-Angiotensin-Aldosterone system**

- ❖ ANP suppresses Renin level
- ❖ Decreases the release of Aldosterone
- ❖ Decreases ADH release
- ❖ ANP stimulates excretion of Na and H₂O
- ❖ Reduces vascular resistance by causing vasodilation.

Role Of Atrial Natriuretic Peptide (ANP)

- ANP increases the urinary excretion of Na⁺ and regulates the electrolytes balance.

6. Role of Kinins

- **Kinins** are proteins in the blood
 - **Kinins** **cause inflammation** and affect blood pressure (especially **lowers the blood pressure**).
 - **Kinins** increases Salt and
-
- Water excretion.**

Osmolarity of a Solutions

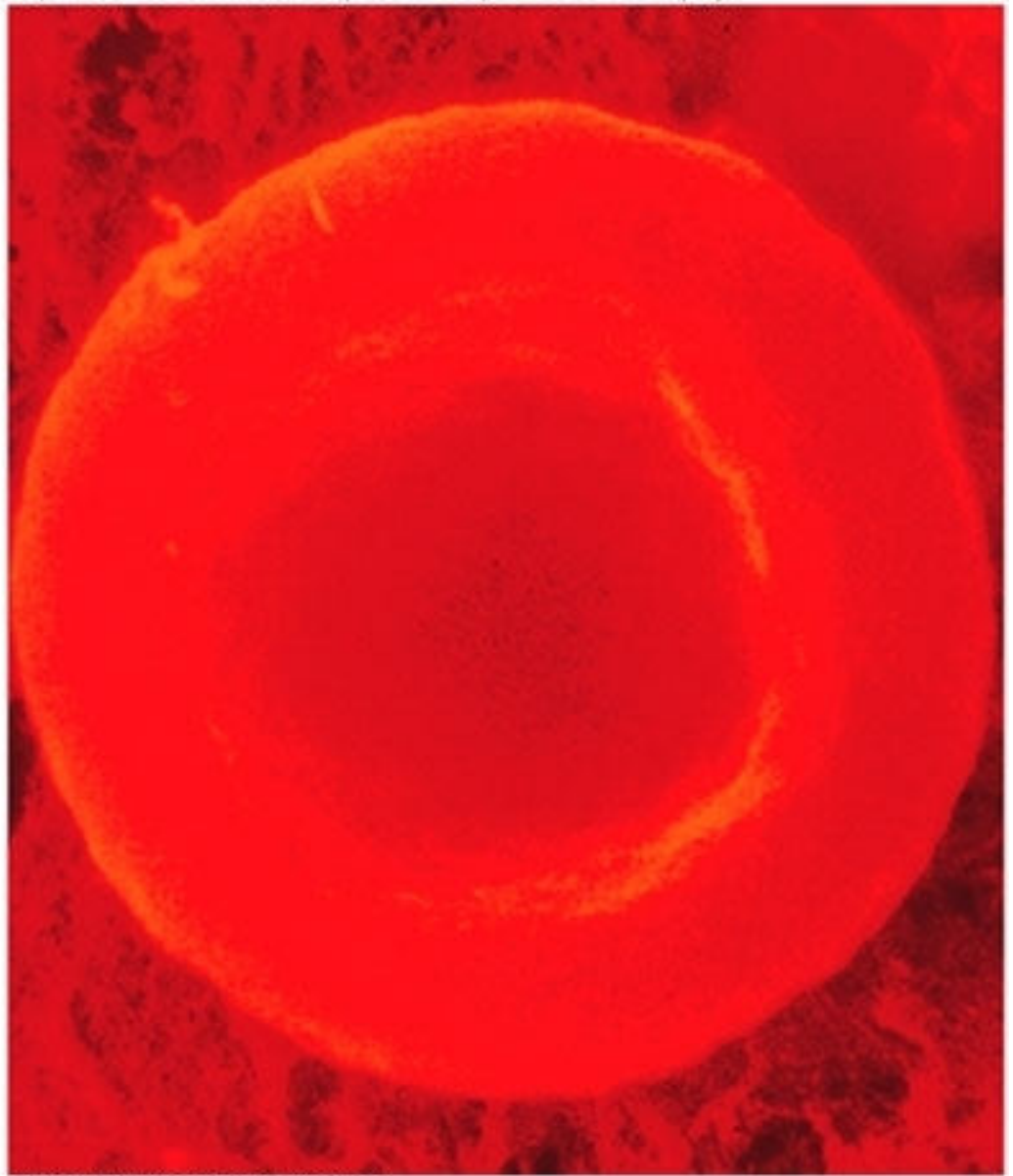
Osmolarity of Solutions

- **Isotonic Solution**— Same concentration of solutes as plasma
- **Hypertonic Solution**— Greater concentration of solutes than plasma
- **Hypotonic Solution**— Lesser concentration of solutes than plasma

Example Of

- 1 mol NaCl weighs 58.44g NaCl
- 1 mmol NaCl weighs 0.058 g NaCl
- One mmol of NaCl contains
 - 58.3 mg NaCl

**Cell in
Isotonic
Solution**



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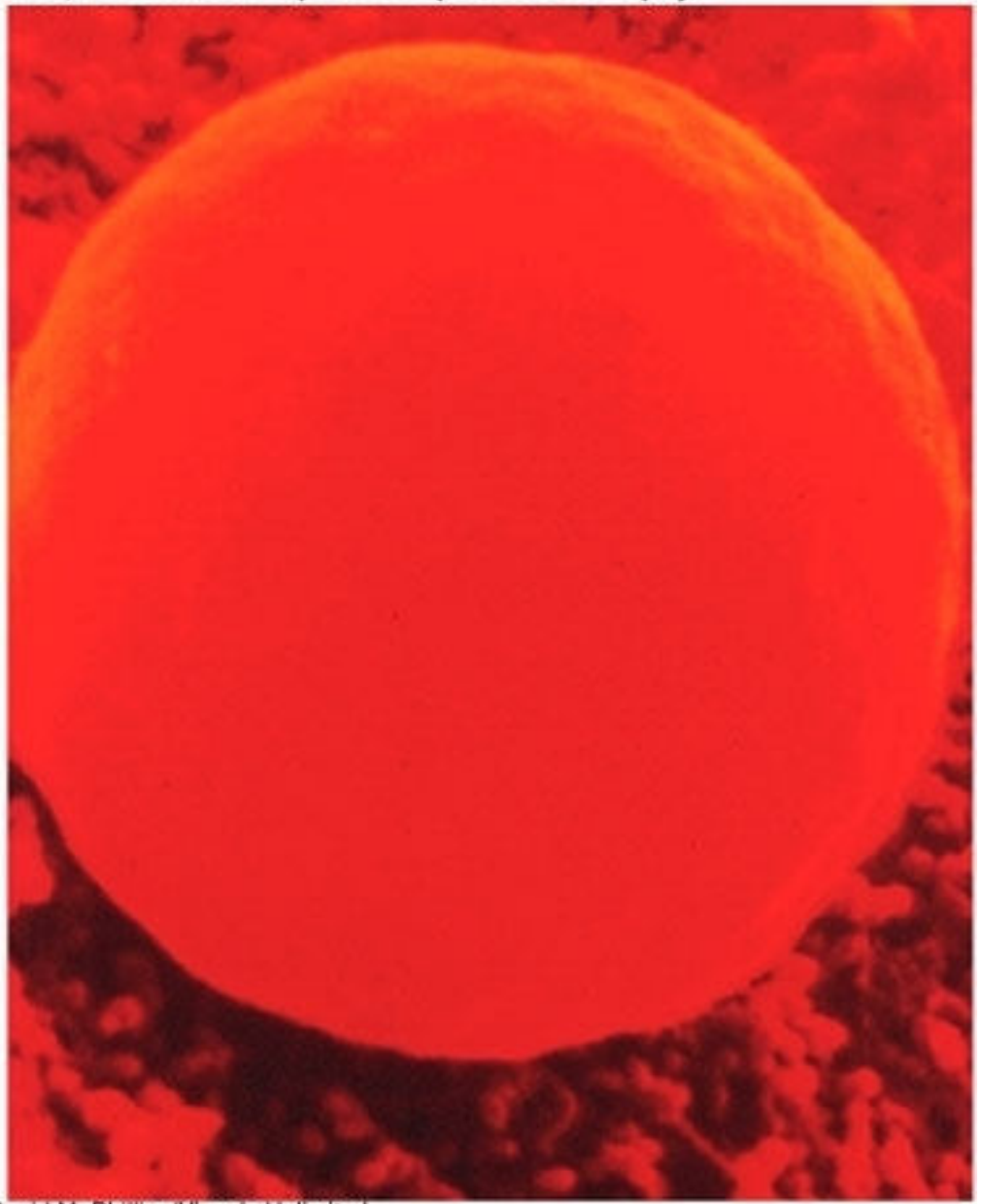
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HYPOTONIC SOLUTIONS

- **5% DEXTROSE & WATER**
- **0.45% SODIUM CHLORIDE**
- **0.33% SODIUM CHLORIDE**

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Cell in a hypotonic solution



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HYPERTONIC SOLUTIONS

- 3% SODIUM CHLORIDE
- 5% SODIUM CHLORIDE
- WHOLE BLOOD
- ALBUMIN
- TOTAL PARENTERAL NUTRITION
- TUBE FEEDINGS
- CONCENTRATED DEXTROSE (>10%)

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**Cell in a
hypertonic
solution**



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Water And Electrolyte Imbalances

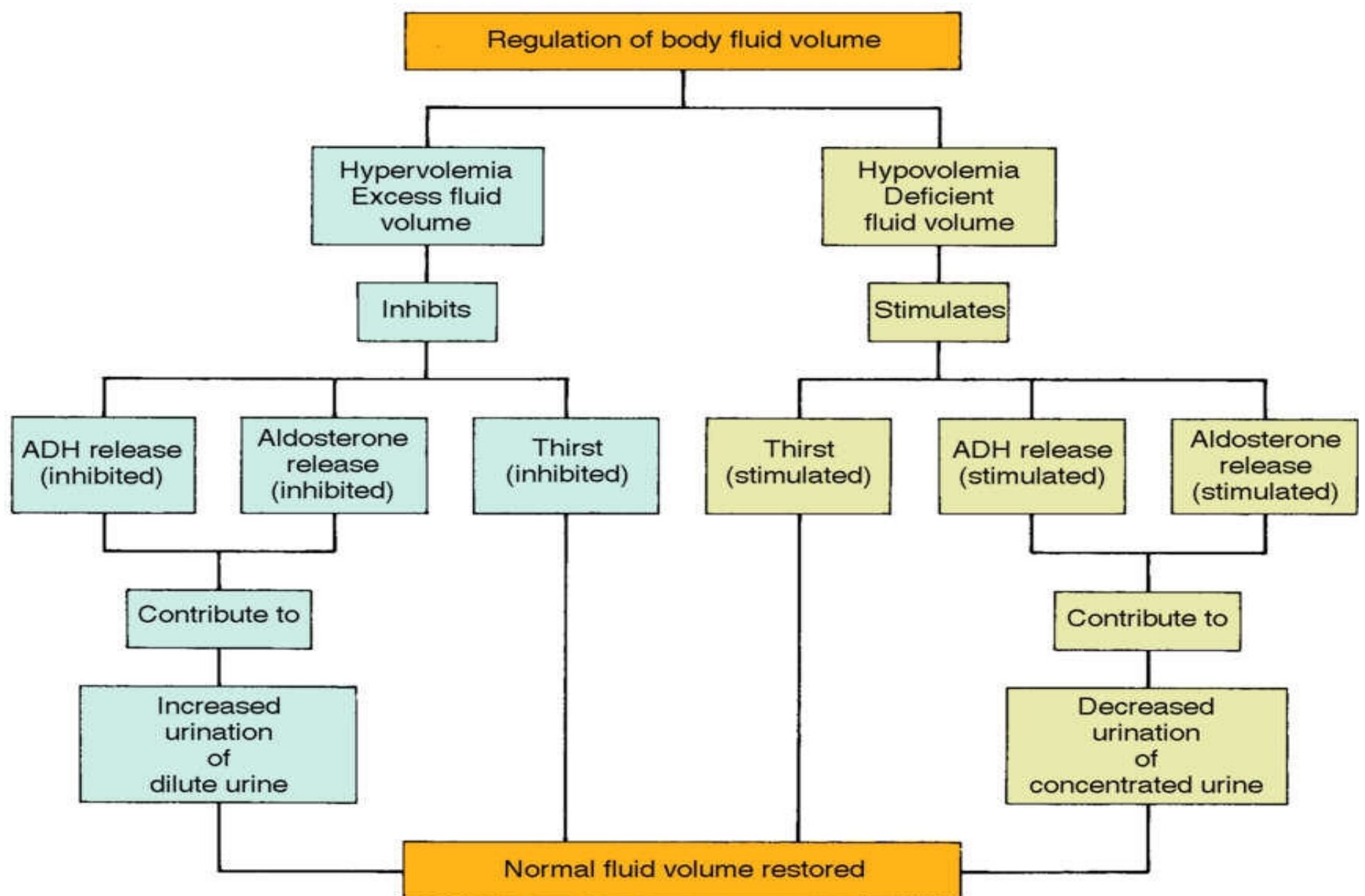
- In a human body **water distribution in compartments**
- **Is due to osmotic pressure produced by Electrolytes**
- **Since Water and Electrolytes balance in the body go together**
- **Hence imbalance in any one of it ,affects both.**
- In simple **words if there is Electrolyte imbalance, it directly affects and causes Water Imbalance.**

What Conditions Leads To Water And Electrolyte Imbalance?

Factors Involved In Water and Electrolyte Balance

- Water intake
- Electrolyte intake
- **Organs and Mechanisms**
 - Hypothalamus
 - Pituitary Gland
 - Kidneys
 - Liver
 - Lung
 - Adrenal glands
 - Cardiac tissue
 - Related Genes
 - Cholesterol
 - Amino acids

REGULATION OF FLUID VOLUME



- Any **defect in the organs associated to factor regulating Water and Electrolyte balance**
- May lead to Water and Electrolyte imbalance.

- Anything that **alters the concentrations of Electrolytes**
- Will alter the concentration of water, and vice versa.

Remember

- In condition of Water and Electrolyte imbalances
- **Care should be taken to manage both the entities simultaneously.**

Conditions Of Water and Electrolyte Imbalances

Water Electrolyte Imbalance Conditions

- Dehydration
- Over hydration

Dehydration

- Dehydration is a condition of Water imbalance.
- Dehydration is characterized by **disturbance of Water and electrolyte balance.**
- In a dehydrated body the output of water exceeds the water intake.
- This causes **reduction of body water below the normal level.**

Basic Cause Of Dehydration

- **No Ingestion of water**
- **Excessive Loss of body fluids**

Features Of Dehydration

- **Water Deficiency Condition**
- **Low body water**
- **Low blood volume**
- **Disturbance in body Electrolytes.**

Types Of Dehydration

Types Of Dehydration

- **Primary Dehydration** /Due to pure water depletion
- **Mixed Type Dehydration** /Due to both Water and Salt depletion
- **Secondary dehydration**/Due to pure salt depletion

Primary Dehydration OR Pure Water Depletion

Causes Of Primary Dehydration

- **Very weak or very ill patients** unable to ingest Water.
- **Mental Patients** who refuses to drink water
- In **Coma Condition**
- Person lost in desert or

Shipwrecked

- Defect to hypothalamus
- Patients with renal tubular disorder
- **Diabetes insipidus-ADH insufficiency.**

Biochemical Alterations

- **Pure Water depletion** occurs almost always:
 - Because of **lack of Water intake**
 - **Rather than** because of **excess looses from the body.**

- **Body water stores get depleted**
because of:
 - Continuing obligatory losses of water through
 - Urine and Feces to excrete metabolic waste.
- The only source of water supply in this condition is
- **Metabolic water obtained from oxidation of food stuffs.**

- The volume of ECF is maintained almost to normal in this condition.
- At the expense of **I.C.F which is grossly reduced in volume**
- **Causing intracellular dehydration.**
- Neural mechanism is activated
- **But if this is not responded**
- By oral Water ingestion
- Causes water depletion in body.

Clinical Manifestations

- **Dry tongue and pinched facies**
(Due to intense thirst)
- **Oligouria-** ADH secreted causes reabsorption of water from kidney tubules
- Causing a gradual **diminution of Urine volume.**

Management

- Give water to drink by mouth
- 5% Glucose by I.V (Hypotonic Soln)
- To correct intracellular dehydration.
- **Note: Never give Isotonic Saline.**
(0.9 % w/v NaCl solution).

Mixed Type Of Dehydration

Water and Salt Depletion

- This is the **most common type of dehydration condition.**

Causes of Mixed Type Dehydration

- **Severe Vomiting**
- **Severe Diarrhea**

Biochemical Alterations

- Mixed dehydration has **both low volume of both water and electrolytes.**
- Here the volume of fluids in both E.C.F and I.C.F is reduced.
- The **patient appears dehydrated and complains of thirst.**

Manifestations

- Feeling of thirst
- Low BP
- Increased blood Urea
- Urinary out put is diminished

Management

- Administering **Mixture of Saline** and **5 % Glucose in 1:1 proportion.**

Secondary Dehydration

Pure Salt Depletion

Pure Salt Depletion

Secondary Dehydration

- This occurs **when fluids of high Na/Cl content are lost from the body**
- And body is replaced by **Salt deficient fluids.**

Causes

- Excessive Sweating
- GIT loss of fluids during vomiting and diarrhea
- Continuous aspirations of G.I fluids
- In Addisons disease (Insufficiency of Adrenal Cortex)
- Vigorous use of diuretics

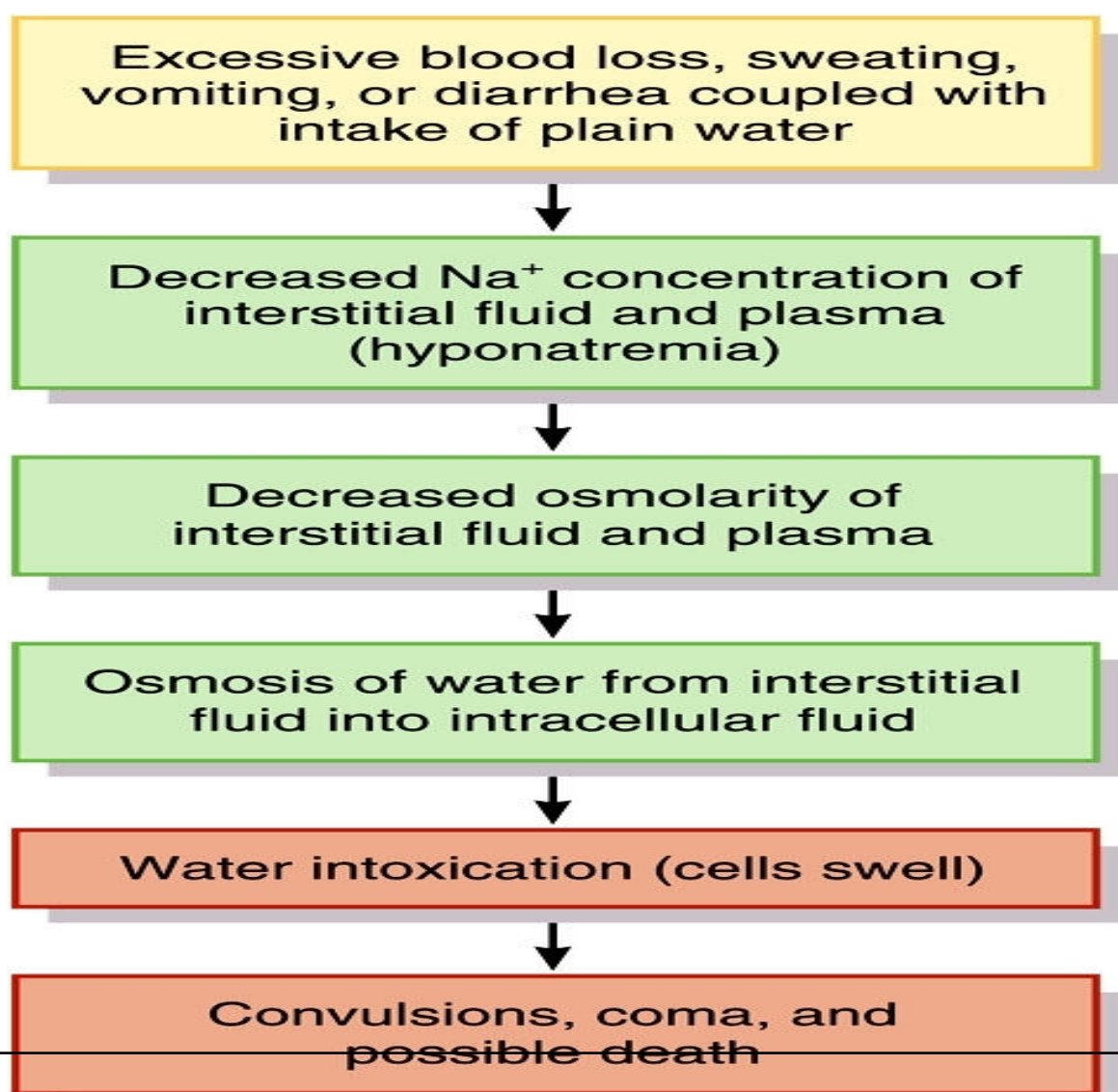
Biochemical Alterations

- The loss of body water and electrolytes
- **Replaced with water without salts** leads to **pure salt depletion**

- In pure salt depletion **E.C.F becomes hypotonic.**
- The **lowered osmotic pressure inhibits the release of ADH**
- In this response **Kidneys excrete water** in an **attempt to maintain normal extracellular Na concentration**
- This decreases plasma and interstitial fluids .
- **The water from ECF (hypotonic) flows into the I.C.F** of cells (hypertonic soln).
- **This further reduces the volume of ECF.**

- In this condition there is no cellular dehydration.
- No response to thirst centre.

Series of Events in Secondary Dehydration



Manifestations Of Secondary Dehydration

- Absence of Thirst (No intracellular dehydration)
- Patient is apathetic, listless
- Hallucinations and Confusions are common
- Anorexia and Nausea
- **Cramps in thigh**, abdominal and respiratory muscles.
- Sunken eyes, inelastic skin
- Low BP, decreased GFR and excretion

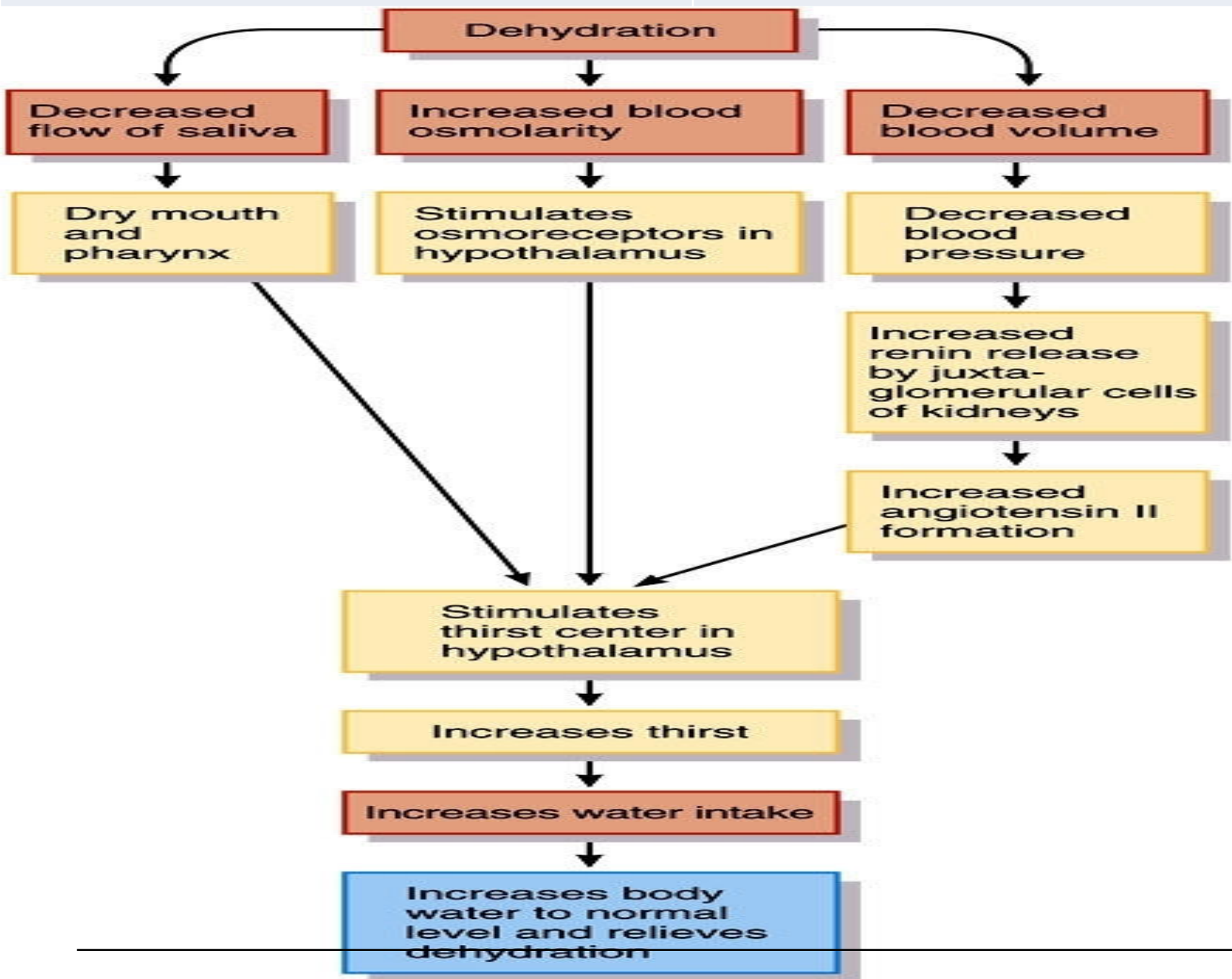
Management of Secondary Dehydration

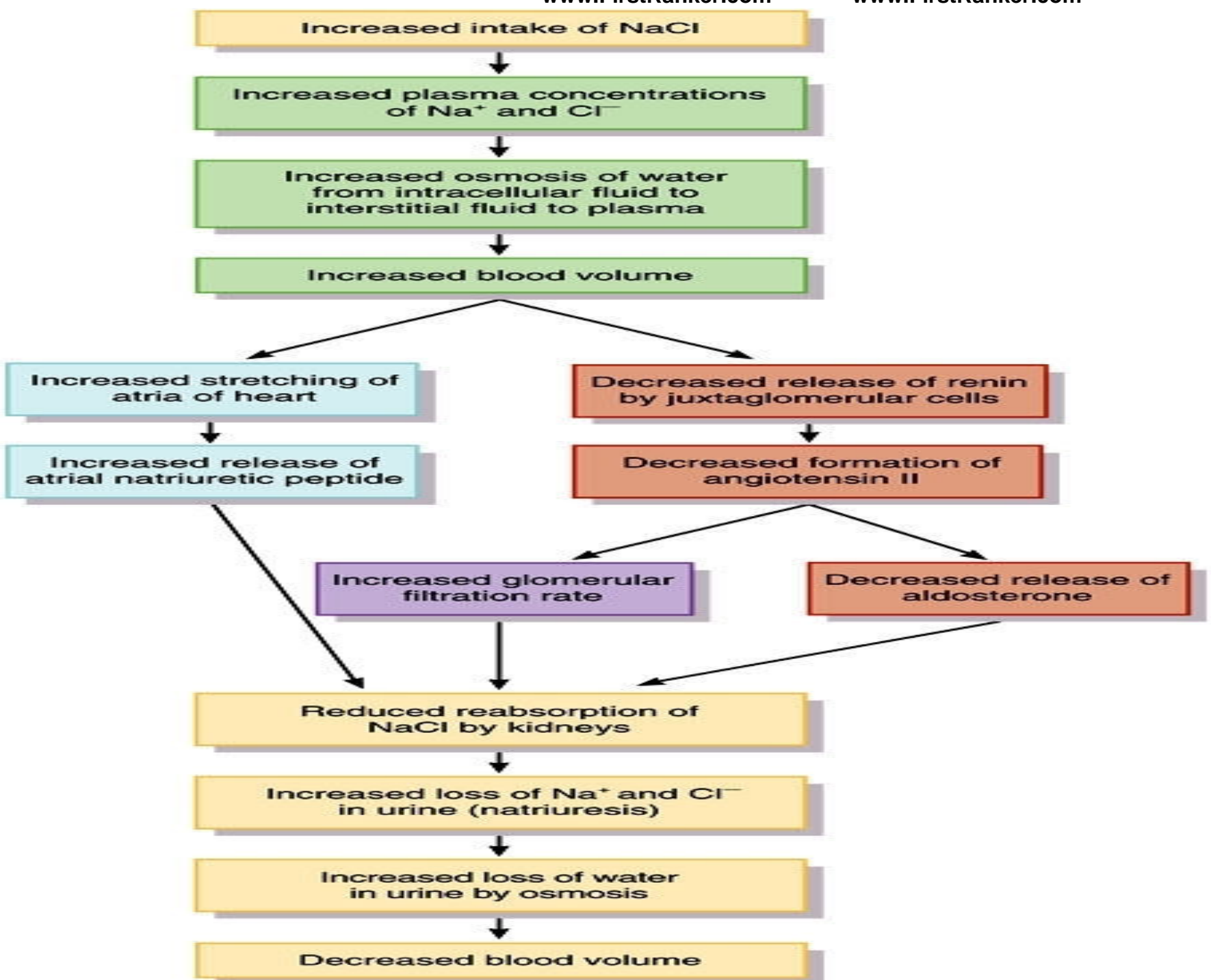
- By administering **Isotonic solution 0.9% NaCl.**

Differences In Primary And Secondary Dehydration

Primary Dehydration	Secondary Dehydration
Caused due to pure Water Depletion when water in take is stopped.	Caused due to pure Salt depletion when fluids of high salt content lost and replaced by salt deficient fluids
Noted in patients with dysphagia Comatose and Mental patients, Lost in deserts	Patients with excessive sweating,vomiting ,diarrhea in CCF, with no electrolytes replenished.
Feels Thirst , Cramps absent ,Pulse and B.P normal.	Thirst absent , cramps present, rapid and thready pulse with low B.P.

Primary Dehydration	Secondary Dehydration
Scanty Urine	Normal or increased Urine
ECF Hypertonic	ECF Hypotonic
Cellular Dehydration present	Cellular Edema present
Management by Water intake and Hypotonic soln	By infusing Isotonic solution





Over Hydration /Water Excess

Over hydration /Water Excess

- Here there is **excess of body water.**

Over Hydration / Water Intoxication

- Due to excess water intake
- Due to water retention

Causes Of Water Excess

- **Excessive administration of parenteral fluids.**
- Renal Failure (No/Less excretion of Urine)
- **Hypersecretion of ADH**
- Administration of Narcotics ,Anaesthesia causes secretion of ADH
- **Excess of Aldosterone (Conn's Syndrome)**

Manifestations

- Headache
- Nausea
- Incoordination of Movements
- Delirium

Management

- Withholding drinking of fluids.
- Administration of 3-5%
Hypertonic Saline
intravenously.

ELECTROLYTE IMBALANCES

ELECTROLYTE IMBALANCES

- Hyponatremia (sodium deficit $< 130\text{mEq/L}$)
- Hypernatremia (sodium excess $>145\text{mEq/L}$)
- Hypokalemia (Potassium deficit $<3.5\text{mEq/L}$)
- Hyperkalemia (Potassium excess $>5.1\text{mEq/L}$)
- Chloride imbalance ($<98\text{mEq/L}$ or $>107\text{mEq/L}$)
- Magnesium imbalance ($<1.5\text{mEq/L}$ or $>2.5\text{mEq/L}$)

**Common Conditions And Disorders
Associated
With Water And Electrolyte Imbalances**

Edema

Edema/Swelling

- Condition in which excess fluid accumulates in the **interstitial compartment**.
- **It is a response to inflammation and injury**



Edema (swelling) of the ankles and feet

Common Conditions Of Edema

- **Inflammation**
- **Infections**
- **Pregnancy**
- **Medications**

Causes Of Edema

- **Increased Hydrostatic Pressure**
- **Small blood vessels become leaky and releases fluid into nearby tissues.**
 - Venous obstruction, Lymphedema, CHF, Renal failure
- **Lowered Plasma Osmotic pressure (Protein loss)**
 - Liver failure, Malnutrition, Burns
- **Increased capillary membrane permeability**
 - Inflammation, Sepsis

Types Of Edema

- **Generalized Edema**
- **Organ specific Edema**
[cerebral, ascites , pleural)
- **Cutaneous Pitting Edema**
- **Non Pitting Edema**



Consequences Of Edema

- Imbalance in Water and Electrolyte distribution.
- **Impaired blood flow**
- Slow healing
- Increased risk of infections
- Pressure sores over bony prominences
- Impaired organ function
(Brain, Liver, Gut, Kidney)

Treatment Of Edema

- Treat the underlying cause of Edema.
- Defect in heart/Lungs/Liver /Kidney should be treated
- **Reducing amount of salt**

Diabetes insipidus

Diabetes insipidus

- **Endocrine Disorder**
- ADH insufficiency
- **Affecting Water and Electrolyte imbalance of the body.**

Diabetes insipidus

- **Diabetes insipidus (DI)** is a condition characterized by:
 - Excretion of large amounts of severely dilute urine
 - With excessive thirst

Incidence Of DI

- 3 in 100,000 individuals of general population.

Causes And Types of DI

Central Diabetes Insipidus

- Central Diabetes insipidus (CDI)
- Involves a deficiency of Vasopressin / Antidiuretic hormone

Nephrogenic Diabetes insipidus

- The second common type of DI is **Nephrogenic Diabetes insipidus** (NDI),
- Is due to kidney or nephron dysfunction
- Caused by an **insensitivity of the Kidneys or nephrons to ADH.**

Manifestations Of DI

- Polyuria with dilute urine.
- Due to osmotic diuresis
- Excessive thirst(Polydypsia)
- Dehydration
- Electrolyte imbalance

Diagnosis

- Urine Osmolarity
- Urine Specific gravity
- Electrolyte concentrations in serum and urine
- Fluid Deprivation test

Treatment

- Central DI respond to [Desmopressin](#) which is given as intranasal or oral tablets

Addisons Disease

Hypoadrenocorticism

Addisons Disease

- **Thomas Addison** first discoverer (1849)
- Autoimmune disease
- Endocrine disorder

Cause

- Defect in Adrenal glands
- Adrenal gland insufficiency
- Deficient Aldosterone and Cortisol

Incidence Of DI

- 1 in 100,000

Biochemical Alterations

- [Hypoglycemia](#) (reduced level of blood glucose)
- [Hyponatremia](#) (low sodium level in the blood)
- [Hyperkalemia](#) (elevated potassium level in the blood)
- [Hypercalcemia](#) (elevated calcium level in the blood)

Manifestations

- [Low blood pressure](#)
- [Syncope](#) (loss of consciousness and inability to stand)
- Confusion, [Psychosis](#), slurred speech
- Severe [Lethargy](#)
- [Convulsions](#)

Diagnosis

- Blood Electrolytes
(Na and K)

- Blood Glucose

- Blood Calcium

- Blood Cortisol levels

- ACTH Stimulation Test

- Uses synthetic pituitary ACTH hormone [Tetracosactide](#) used for diagnosis

Management

- Standard therapy involves intravenous injections of Glucocorticoids
- of Hydrocortisone tablets, Prednisone tablets
- Large volumes of intravenous saline solution with Dextrose/Glucose.
- Oral doses of **Fludrocortisone Acetate**

Cushings Syndrome

Hyperadrenocorticism

Cause

- Over activity of Adrenal glands
- Excess of Aldosterone and Cortisol

Incidence

- 1 in 100,000

Biochemical Alterations

- [Hyperglycemia](#) (Increased level of blood glucose)
- [Hypernatremia](#) (High Sodium level in the blood)
- [Hypokalemia](#) (Low potassium level in the blood)
- [Hypocalcemia](#) (Low Calcium level in the blood)

Manifestations

- [High blood pressure](#)
 - [Weight gain ,Central obesity. Buffalo Hump and Moon Face](#)
 - Insomnia
 - Excessive Sweating
 - Depression
 - Anxiety
-

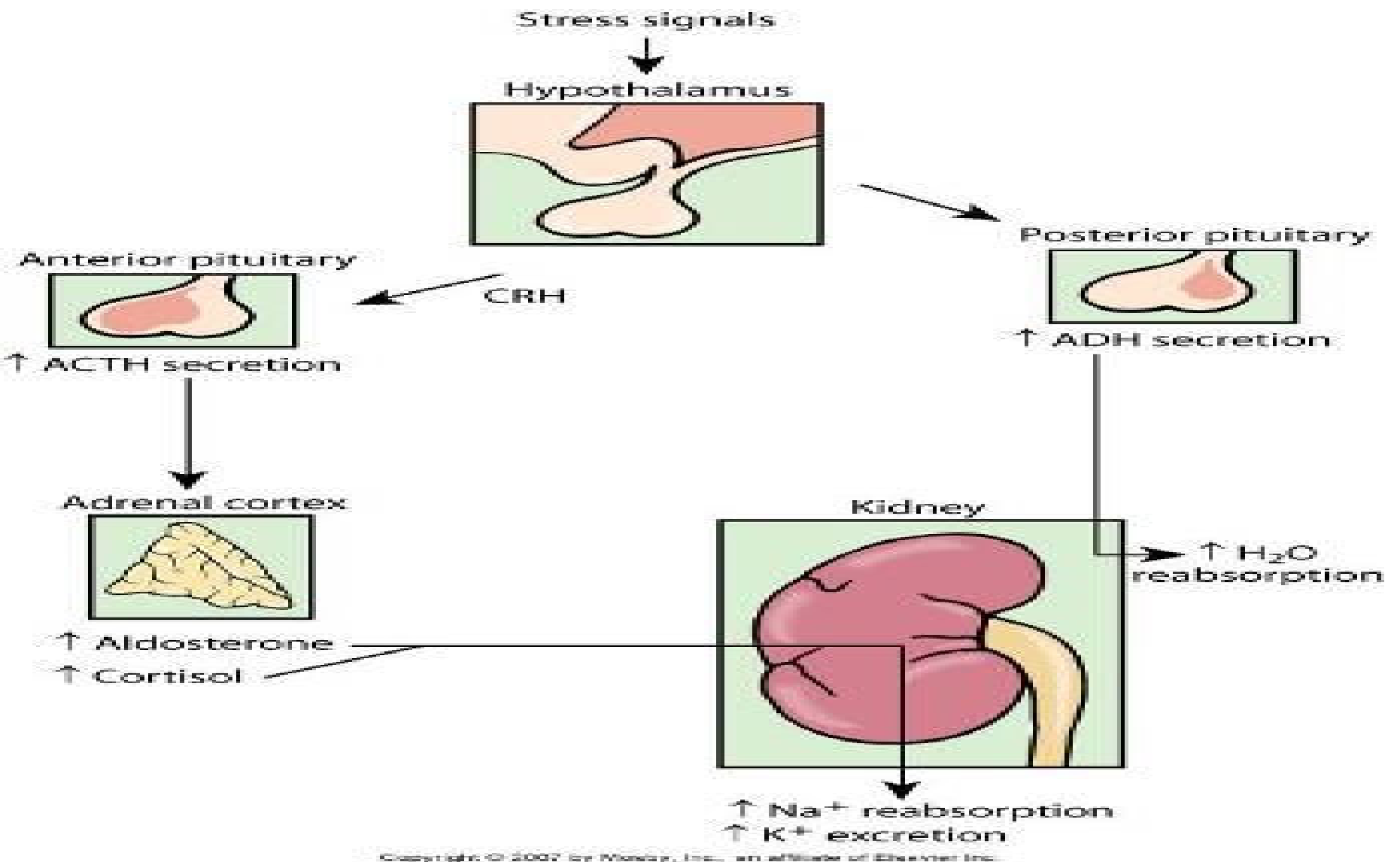
Diagnosis

- Blood Electrolytes
(Na⁺ and K⁺)
- Blood Glucose
- Blood Calcium
- Blood Cortisol levels

Management

- In adrenal Adenomas surgical removal.
- [Ketoconazole](#), [Metyrapone](#) inhibit Cortisol biosynthesis.
- [Mifepristone](#) is a powerful Glucocorticoid type II receptor antagonist

Effects of Stress on Water And Electrolyte Balance



- The Hypothalamus and Pituitary gland
- Integrate communication between nervous and endocrine systems
- Stress severely affect this coordination and affect Water and Electrolyte balance

Diagnostic Tests To Check for Fluid and Electrolytes Balance And Imbalance

Blood Investigations

- Serum Electrolytes
- Serum Creatinine = 0.6 – 1.5 mg/dl
- Blood Urea and BUN = 8-20 mg/dL
- Serum Osmolality
- Serum Albumin – 3.5-5.5 g/dL
- Serum Hematocrit = 40-54%/men, 38-47% for women (Decreased in Dehydration)

Urine Investigations

- Urine pH
- Urine specific gravity
- Urine Osmolarity
- Urine Creatinine Clearance
- Urine Sodium
- Urine Potassium

Questions

- Distribution & functions of Water in human body.
- What are Electrolytes? Give its distribution & functions related to human body.
- What is water electrolyte balance? Explain the factors involved in water electrolyte balance.
- What are disorders of Water and Electrolyte imbalances?

- Dehydration- types, causes & management.
- Differentiate between Primary and Secondary Dehydration
- Over hydration/Water Toxicity/Water Intoxication.
- Edema
- Diabetes Insipidus.
- Addison's & Cushing Syndrome.

Thank You