

Rajiv Gandhi University of Health Sciences, Karnataka

II Year B.Pharm Degree Examination - DEC-2018

Time: Three Hours Max. Marks: 70 Marks

PHYSICAL PHARMACEUTICS (Revised Scheme 3)

Q.P. CODE: 2606

Your answers should be specific to the questions asked Draw neat labeled diagrams wherever necessary

LONG ESSAYS (Answer any Two)

 $2 \times 10 = 20 \text{ Marks}$

- 1. Explain non-Newtonian type of flow with rheograms, mechanism and suitable examples.
- 2. Define adsorption isotherm. Draw various types of adsorption isotherm and explain their behaviour.
- 3. Write the principle and method involved in the determination of particle size in a powder using Anderson apparatus.

SHORT ESSAYS (Answer any Six)

 $6 \times 5 = 30 \text{ Marks}$

- 4. What are the structured vehicles and give its importance?
- 5. What is Bancrofts rule and give its applications.
- 6. State Hardy-schulze's rule. Explain.
- 7. Write the different types of inclusion complexes with suitable examples.
- 8. Differentiate Newtonian and non-Newtonian systems.
- 9. Explain micellar solubilization in detail with examples.
- 10. Give the applications of solubility.
- 11. Write a note on method of determination of PH.

SHORT ANSWERS

 $10 \times 2 = 20 \text{ Marks}$

- 12. What is refraction?
- 13. Define conjugate solutions and consolute temperature.
- 14. Give any two applications of distribution law.
- 15. Give BET equation.
- 16. What is steady state diffusion?
- 17. Differentiate mobility and fluidity.
- 18. Zeta potential
- 19. Gold number
- 20. Give the uses of suspensions in ophthalmics.
- 21. Why colloids are charged? Give examples.
