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B.Tech. Third Semester (Chem. / Poly / Food / Pulp / Oil / Petro) (Old)

Inorganic Chemistry 3 SCE 1

P. Pages: 2 Time: Three Hours

AW - 3550

Max. Marks: 80 Notes: 1. Due credit will be given to neatness and adequate dimensions. Assume suitable data wherever necessary. 2. Diagrams and chemicals equations should be given wherever necessary. 3. Discuss the reaction, mechanism wherever necessary. 4. Use of pen Blue/Black ink/refill only for writing the answer book. 5. SECTION - A Derive the Schrodinger wave equation. 1. 6 a) State and explain de Broglie's principle. b) 3 Write the postulates of Bohr's theory. c) OR 2. What are quantum numbers? Discuss and give the distribution of electrons with in the 6 a) orbital. Explain the following: b) Aufban principle Atomic number ii) State and explain Heisenberg's uncertainty principle. 3 c) 3. What is ionic bond? Explain with example and properties. a) b) Explain the following: 8 Covalent bond Metallic bond ii) iii) Bond energy iv) Bond length OR What is meant by lattice energy? Discuss the Born Haber cycle. 6 4. a) What is hydrogen bonds? How are they classified. b) What is hybridisation? Discuss SP³ hybridisation with example. c) 5. What is coordinate bond? Discuss valence bond theory. a) Write the applications of chelates. b) 3 What are the factors affecting the stability of complexes. c)

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