



**B.Sc. (Part-III) Semester-VI Examination**

**6S-PHYSICS**

**(Statistical Mechanics and Solid State Physics)**

Time : Three Hours]

[Maximum Marks : 80

**Note :-** Attempt **all** questions.

1. (A) Fill in the blanks : 2
- (i) The phase space of N particles have \_\_\_\_\_ dimensions.
  - (ii) Quantum statistics is applicable to identical and \_\_\_\_\_ particles.
  - (iii) Schottky defects are \_\_\_\_\_ defects.
  - (iv) Superconductors are perfectly \_\_\_\_\_.
- (B) Choose correct alternative of the following : 2
- (i) Pauli's exclusion principle is applicable to :
    - (a) M-B statistics
    - (b) B-E statistics
    - (c) F-D statistics
    - (d) None of these
  - (ii) Structure of NaCl crystal is :
    - (a) BCC
    - (b) FCC
    - (c) Simple cubic
    - (d) Hexagonal
  - (iii) For diamagnetic materials, magnetic susceptibility is :
    - (a) small and negative
    - (b) small and positive
    - (c) large and negative
    - (d) large and positive
  - (iv) Nanometer is equal to :
    - (a)  $10^{-3}\text{m}$
    - (b)  $10^{-6}\text{m}$
    - (c)  $10^{-7}\text{m}$
    - (d)  $10^{-9}\text{m}$
- (C) Answer the following in **one** sentence : 4
- (i) What are Bosons ?
  - (ii) What is point defect ?
  - (iii) What is Curie temperature ?
  - (iv) On what factors properties of nanomaterial depend ?



2. (A) Write expression for thermodynamic probability in M-B distribution and find expression for M-B distribution law. 6
- (B) By using M-B distribution law of molecular speed show that root mean square speed of gas molecule is given by  $V_{rms} = \sqrt{\frac{3kT}{m}}$ . 4
- (C) Find the thermodynamic probability for four distinguishable particles for the microstate (2, 2). 2

**OR**

3. (P) State and explain principle of priori probability. 4
- (Q) Show that most probable velocity of gas molecule is  $V_p = \sqrt{\frac{2kT}{m}}$ . 4
- (R) Explain the terms :  
 (i) Thermodynamic probability  
 (ii) Statistical weight. 4

**EITHER**

4. (A) What are the main postulates of Fermi-Dirac statistics? 4
- (B) Using Bose-Einstein distribution law, deduce an expression for Planck's law of energy distribution in black body radiation. 7
- (C) What is Fermi energy? 1

**OR**

5. (P) Explain the concept of distinguishable and indistinguishable particles. 3
- (Q) Obtain an expression for Fermi-Dirac distribution law by assuming thermodynamic probability. 5
- (R) What is Fermi function? How it behaves at absolute zero? 4

**EITHER**

6. (A) Define unit cell of crystal. Explain primitive and non-primitive unit cell. 4
- (B) Derive Bragg's law for diffraction of X-rays. 4
- (C) What is line defect in crystal? Explain the screw dislocation in crystal with neat diagram. 4

**OR**

7. (P) Give the names of seven crystal systems and state the parameters of their unit cells. 6
- (Q) What are Miller indices? Find Miller indices of the plane having intercepts (a, 2b, c/2). 4
- (R) What is reciprocal lattice? 2

**EITHER**

8. (A) Obtain an expression for electrical conductivity in terms of mean free path of electrons. 4
- (B) Discuss formation of insulator, semiconductor and conductor on the basis of band theory of solid. 6
- (C) What are conduction electrons? 2

9. (P) Explain the motion of electron in metals and hence derive an expression for drift velocity of electron. 5

(Q) Explain qualitatively conduction band, valence band and energy gap in solids. 4

(R) Explain nearly free electron model. 3

**EITHER**

10. (A) State properties of paramagnetic materials. 4

(B) Explain ferromagnetism on the basis of domain theory. 4

(C) What is magnetic dipole moment ? Obtain an expression for orbital magnetic dipole moment. 4

**OR**

11. (P) Give Langevin's theory of paramagnetism; hence prove that the susceptibility of paramagnetic material is inversely proportional to absolute temperature. 8

(Q) State properties of diamagnetic material. 4

**EITHER**

12. (A) Give brief idea of BCS theory of superconductivity. 6

(B) Explain type-I and type-II superconductor. 4

(C) Define : 2

(i) Critical temperature

(ii) Critical magnetic field.

**OR**

13. (P) Explain importance of surface to volume ratio and quantum size effect in nanomaterials. 6

(Q) State any four applications of nanomaterials. 4

(R) What are nanomaterials ? 2

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