Code: 19A51102T

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B.Tech I Year I Semester (R19) Regular Examinations January 2020

CHEMISTRY

(Common to CSE and IT)

Time: 3 hours Max. Marks: 70

PART - A

(Compulsory Question)

- Answer the following: (10 X 02 = 20 Marks)
 - (a) Write Schrodinger equation.
 - (b) Write the electronic configuration and bond order of O₂⁺.
 - (c) Calculate the electrode potential of copper, if the concentration of CuSO₄ is 0.1 M. Given that $E^{0}Cu^{2+}/Cu = + 0.34 \text{ V}$
 - (d) Define battery? Give examples for secondary battery.
 - (e) What is stereospecific polymerization?
 - What is meant by functionality?
 - (g) What is the importance of fingerprint region in IR spectrum?
 - (h) State Beer's Lambert's law of absorption.
 - Define the term catenanes.
 - Explain the role of supramolecules as switching devices

PART - B

(Answer all five units, 5 X 10 = 50 Marks)

UNIT - I

Discuss the molecular orbital diagram of CO and find out its bond order. 2

- (a) Explain wave nature of electrons.
 - (b) Discuss π-molecular orbitals of benzene.

- (a) Derive Nernst equation for a electrochemical cell.
 - (b) Explain the working of methanol fuel cells.

OR

- (a) Explain conductometric titration of strong acid and strong base with example.
 - (b) What is electrochemical sensor? Explain amperometric sensors with examples.

UNIT - III

- 6 (a) Explain the preparation and applications of urea-formaldehyde.
 - (b) Write a short note on chain growth and step growth polymerization.

- 7 (a) Explain the conducting behaviour of polyaniline.
 - (b) Discuss any one mechanism of polymer formation.

UNIT - IV

- (a) Explain IR spectrophotometer with instrumentation.
 - (b) Discuss the principle involved in HPLC.

OR

- (a) Explain the principle of NMR spectroscopy.
 - (b) Describe in brief the terms: (i) Auxochrome. (ii) Chromophore.

UNIT - V

10 Explain briefly about basic lock and key mechanism principle of supramolecules.

(a) Write a note on self assembly in biological systems.

Explain the role of supramolecules as catalysts.

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