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Code: 15A51103

B.Tech I Year I Semester (R15) Supplementary Examinations November/December 2019

## **BASIC CHEMISTRY**

(Food Technology)

Time: 3 hours Max. Marks: 70

## PART - A

(Compulsory Question)

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- 1 Answer the following:  $(10 \times 02 = 20 \text{ Marks})$ 
  - (a) What is meant by entropy?
  - (b) What is meant by redox reaction?
  - (c) What is meant by order of reaction?
  - (d) What does valency bond theory explain?
  - (e) What are the important components in terms of instrumentation of Microwave spectroscopy?
  - (f) What is meant by IR spectroscopy? Give its important applications.
  - (g) What is meant by polymerization?
  - (h) What is meant by Friedel Craft Alkylation?
  - (i) Define the degree of freedom.
  - (j) What are important properties of colloids?

## PART - B

(Answer all five units,  $5 \times 10 = 50 \text{ Marks}$ )

UNIT – I

- 2 (a) What do you understand by thermodynamic system and surroundings?
  - (b) Describe the important applications of emf measurements.

OF

- 3 (a) Give an account of electrochemical series and its applications.
  - (b) Explain about electrochemical cells.

UNIT - II

- 4 (a) What do you understand by the term rate-determining steep of a complex reaction? What is steady-state hypothesis?
  - (b) What are bonding and antibonding molecular orbitals?

OR

- 5 (a) Write short notes on parallel reactions?
  - (b) Draw molecular orbital diagram for NO molecule.

UNIT – III

- 6 (a) Which selection rule is used in Infrared spectroscopy? Explain about finger print region.
  - (b) Explain why  $[Ni(CO)_4]$  is tetrahedral,  $[Ni(H_2O)_6]^{2^+}$  is square planar and  $[Ni(NH_3)_6]^{2^+}$  is octahedral using valence bond theory.

OR

- 7 (a) Briefly explain the Beer-Lambert's law and give its applications.
  - (b) Brief account on Haemoglobin with respect to metal ion.

UNIT – IV

- 8 (a) Write short notes on Perkin's reaction.
  - (b) Illustrate the details of Papaverine.

OR

- 9 (a) Write short notes on Toxol.
  - (b) What are Biomolecules? Explain aminoacids.

UNIT – V

- 10 (a) State and Explain the phase role.
  - (b) Explain basic principle of TLC and what are its usefulness?

OR

- 11 (a) Draw and explain the phase diagram of water system, i.e. one-component system.
  - (b) What are different separation techniques stratalize crystallization.