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B.Tech I Year II Semester (R15) Supplementary Examinations December 2016

ENGINEERING CHEMISTRY

(Common to CE, EEE and CSE)

Time: 3 hours Max. Marks: 70

PART – A

(Compulsory Question)

- 1 Answer the following: $(10 \times 02 = 20 \text{ Marks})$
 - (a) Explain the principle of reverse Osmosis for desalination of water.
 - (b) What is the principle in estimation of hardness by EDTA method?
 - (c) What are ion exchange resins?
 - (d) How does Buna-S differ from Buna-N?
 - (e) What are inorganic polymers? Give two examples.
 - (f) Explain the working principle of $H_2 O_2$ fuel cells.
 - (g) What are the main constituents of coal? Explain their significance.
 - (h) Define the term cetane number. Mention one additive to improve cetane rating.
 - (i) What are refractory materials? Give one example.
 - (j) What are the characteristics of a good lubricant?

PART - B

(Answer all five units, $5 \times 10 = 50 \text{ Marks}$)

UNIT – I

- 2 Discuss the following process:
 - (a) Priming and foaming.
 - (b) Caustic embrittlement.
 - (c) Phosphate conditioning.

OR

A water sample on analysis was found to contain the following: $Mg(HCO_3)_2 = 22.5$ ppm, $CaSO_4 = 7.5$ ppm, NaCl = 3.6 ppm. Calculate the permanent and temporary hardness.

[UNIT – II]

What is Bakelite? Explain the manufacturing process and mention its uses.

OR

5 Explain the procedures used in the processing of natural rubber.

UNIT – III

What is Cathodic protection? How is it achieved by use of a sacrificial anode?

OR

- 7 Explain in detail about:
 - (a) Ni Cd.
 - (b) Lithium ion batteries and their significant uses.

UNIT – IV

8 Calculate the volume of air required for the complete combustion of 1 m³ of the gaseous fuel having the following composition by volume: $H_2 = 50\%$, $CH_4 = 36\%$, $N_2 = 15\%$, CO = 6%, $C_2H_4 = 4\%$ and H_2O vapor = 2.5%.

OR

9 Discuss the Fischer Tropsch synthesis with a neat diagram.

[UNIT – V]

Discuss the process of manufacturing of Portland cement with the help a diagram.

OR

Name two important solid lubricants and discuss their lubricating actions.