

## Rajiv Gandhi University of Health Sciences, Karnataka First Semester B.Pharm Degree Examination - 05-Jun-2024

Time: Three Hours Max. Marks: 75 Marks

## PHARMACEUTICAL INORGANIC CHEMISTRY Q.P. CODE: 5004

Your answers should be specific to the questions asked Draw neat labeled diagrams wherever necessary All the questions are compulsory

LONG ESSAYS 2 x 10 = 20 Marks

Give principle, procedure, and role of reagents involved in the limit test for a) Iron b) Lead.

Discuss various types of physiological buffers. Explain the mechanism of their buffer action.

Explain the principle, reaction and procedure involved in assay of chlorinated lime and hydrogen peroxide.

SHORT ESSAYS 7 x 5 = 35 Marks

Write the principle and reactions involved in the limit test for sulphate.

OR

What are buffered isotonic solutions? Give details.

 What are dentifrices? Classify them with example. Write a note on role of fluoride as anticaries agent.

OR

What are antacids? Classify them with structural examples. State requirements for an ideal antacid.

- Describe the properties of α, β and γ radiations:
- Write a note on electrolyte replacement therapy, give compositions.
- Define and classify antidotes with structural examples. Write a note on activated charcoal.
- 8. What are emetics? Write the method of preparation and assay of copper sulphate.
- 9. Describe the various mechanism of action of inorganic anti-microbial agents.

SHORT ANSWERS 10 x 2 = 20 Marks

- Write the principles and reactions involved in the limit test for chloride.
- Write the Preparation and use of barium sulphate reagent.
- Differentiate between limit test and test for purity.
- Write the composition and uses of kaolin.
- Write the preparation of boric acid.
- Write a note on ORS
- Write the synonym for ferrous sulphate and copper sulphate & uses.
- 17. What is osmotic pressure and give buffer equation
- Write the storage condition of sodium iodide I<sup>131</sup>.
- Classify extra and intra cellular electrolytes with examples.

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