

Rajiv Gandhi University of Health Sciences, Karnataka I Year B. Pharm Degree Examination - 23-Jun-2022

Time: Three Hours Max. Marks: 70 Marks

PHARMACEUTICAL INORGANIC CHEMISTRY (RS4) Q.P. CODE: 2629

Your answers should be specific to the questions asked Draw neat, labeled diagrams wherever necessary

LONG ESSAYS (Answer any Two)

2 x 10 = 20 Marks

- Define limit test. Enlist the different types of limit test. Discuss in detail the limit test for sulphate and iron
- What are GIT agents? Write the principles and reaction for the assay of sodium bicarbonate
- What are Redox titrations? Explain the preparation, assay and use of copper sulphate and boric acid

SHORT ESSAYS (Answer any Six)

6 x 5 = 30 Marks

- Explain the different types of complexometric titrations with examples
- Compare and contrast Mohr's method and Volhard's method
- Discuss the theory of non-aqueous titrations
- 7. Explain various methods of minimization of errors
- 8. Write a note on electrolytes used in replacement therapy
- 9. Write a note on pharmaceutical aids
- 10. What are antimicrobials? Write a note on various iodine preparations
- Write a note on Ors (Oral rehydration salt)

SHORT ANSWERS 10 x 2 = 20 Marks

- 12. Give the uses of Oxygen and Carbon dioxide
- Write the molecular formula and uses of Milk of Magnesia
- 14. What are protective and adsorbents? Give two examples
- 15. Write the composition and application of zinc eugenol cement
- Define accuracy and precision
- 17. What is leveling and differentiating effects?
- 18. Give the role of starch as indicator in redox titrations
- 19. What are sequestering agents? Give two examples
- What are haematinics? Give two examples
- 21. What are anticaries agents? Give examples

