

## Rajiv Gandhi University of Health Sciences, Karnataka I Year B. Pharm Degree Examination - 07-Jun-2023

Time: Three Hours Max. Marks: 70

## PHARMACEUTICAL INORGANIC CHEMISTRY (RS-4) Q.P. CODE: 2629

Your answers should be specific to the questions asked Draw neat, labeled diagrams wherever necessary

## LONG ESSAYS (Answer any Two)

2 x 10 = 20 Marks

- What are Antimicrobial agents? Classify with examples. Describe method of preparation and principle involved in the Assay of Hydrogen peroxide
- Explain the principle, reactions and apparatus involved in the limit test for Arsenic
- What are Haematinics? Describe the method of preparation, principle in the assay of Ferrous sulphate

## SHORT ESSAYS (Answer any Six)

6 x 5 = 30 Marks

- Define and classify Antidotes. Write the mode of action of Sodium thiosulphate as an Antidote
- 5. Write the principle and reactions involved in the Assay of Boric acid
- 6. Explain the theory of acid-base indicators in detail
- Write the procedure and principle involved in limit test for Iron
- 8. Add a note on electrolytes used in combination therapy
- 9. What is precipitation titration and give the principle involved in the assay of Sodium Chloride?
- Write a note physiological acid base balance
- Explain the principle involved in the complexometric titrations in detail and how will you estimate Magnesium Sulphate

SHORT ANSWERS 10 x 2 = 20 Marks

- 12. Write the role of hydrogen sulphide in the limit test for lead
- Write the storage condition and uses of Oxygen
- 14. Define protective and adsorbents and give examples
- 15. Define expectorants and give examples
- Define Dentifrices with two examples
- Define and classify Errors
- Define pharmaceutical aid with two examples
- Name two Acid-base indicators
- 20. What are primary and secondary standard solutions?
- Define Masking and de-masking agents with one example

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