

Time : Three Hours]

[Maximum Marks : 80

**N.B. :—** (1) Question No. 1 is compulsory.(2) Solve any **four** questions from remaining.

(3) Draw neat labelled diagram wherever necessary.

(4) Assume suitable data wherever necessary.

1. Solve any **five** of the following :

4×5=20

(a) Why addition of sodium oleate to phenol water system lowers the critical solution temperature ?

(b) Define glass transition temperature (T<sub>g</sub>). Explain effect of cross linking on T<sub>g</sub> of polymer.

(c) What is active and passive diffusion ? What is osmosis and ultrafiltration ?

(d) Define solubility according to I.P.

(e) Define rate, order and molecularity of a reaction. What do you understand by pseudo-order reaction ?

(f) What are bulges and spurs in the hysteresis loop ?

(g) What are complexes ? Give one example.

2. Derive an equation for solubility of non-ionic solids in liquids.

15

3. Explain steady state diffusion and derive Ficks First and Second law of diffusion. Add a note on measurement of diffusion coefficient.

15

4. What are Non-Newtonian systems ? Explain their types and comment briefly on thixotropy.

15

5. (a) What is the influence of pH on solubility of weak acids and weak bases ?

8

(b) Give applications of polymers in Drug delivery systems.

7

6. (a) Write principle, working, disadvantages and modifications of Cup and Bob viscometer.

8

(b) Explain metal ion complexes.

7

7. (a) How the stoichiometric ratio and stability constant is determined by solubility and distribution method ?

8

(b) How the half-life and shelf-life of drugs can be calculated ?

7