

DU MPhil PhD in Bio Physics N

Sr.No	Question Id	Question Description	Question Body	Options
1	1	DU_J19_MP HIL_BIOPHY _New6july_ Q01	Which of the following is an organelle enclosed by a single membrane:	1:Lysosomes, 2: ribosomes, 3: mitochondria 4: chloroplast,
2	2	DU_J19_MP HIL_BIOPHY _New6july_ Q02	Chlorophyll consists of two parts, a metal ion of Magnesium and an organic portion termed as:	5: dextran, 6: globin, 7: porphyrin, 8: sphingolipid,
3	3	DU_J19_MP HIL_BIOPHY _New6july_ Q03	Intellectual Property rights granted over creations like music, novels, paintings and cinematic work is classified as:	10: Copyright, 11: Trademark 12: Certification 9: Creative patent
4	4	DU_J19_MP HIL_BIOPHY _New6july_ Q04	"Dextrose" is an example of which type of macromolecule:	13: protein, 14: carbohydrate 15: lipid, 16: vitamin,
5	5	DU_J19_MP HIL_BIOPHY _New6july_ Q05	The secondary structure of proteins is stabilized mainly via :	17: hydrogen bonds between main chain atoms of amino acids . 18: hydrogen bonds between side chains of amino acids . 19: hydrogen bonds between main chain atoms of amino acids . 20: ionic bonds between atoms of amino acids
6	6	DU_J19_MP HIL_BIOPHY _New6july_ Q06	The genetic code is known to be degenerate with several three letter codons coding for the same amino acid. How many codons code for the Methionine amino acid :	21: 3, 22: 1, 23: 2, 24: 0,

7	7	DU_J19_MP HIL_BIOPHY _New6july_ Q07	Which of the following is a method for determining the three dimensional structure of proteins.	25:Isothermal Calorimetry, 26: X-ray crystallography, 27: Dynamic light scattering, 28:Optical microscopy
8	8	DU_J19_MP HIL_BIOPHY _New6july_ Q08	In a NATIVE-PAGE experiment, the proteins are separated:	29:only on basis of molecular weight 30: only on basis of isoelectric point 31: only on the basis of quaternary structure 32:both the basis of molecular weight and isoelectric point
9	9	DU_J19_MP HIL_BIOPHY _New6july_ Q09	A mRNA of about 1.5kb is expected to code for a protein of the following length:	33:~ 500 amino acids 34: ~ 200 amino acids 35: ~1000 amino acids 36:~ 1500 amino acids
10	10	DU_J19_MP HIL_BIOPHY _New6july_ Q10	Which of the following model organisms is used routinely in biological sciences is actually a Frog:	37:Caenorhabditis elegans 38: Xenopus tropicalis 39: Saccharomyces cerevisiae 40:Danio Rerio
11	11	DU_J19_MP HIL_BIOPHY _New6july_ Q11	The bacterial genome typically codes for about _____ genes.	41:4,000, 42: 1,500, 43: 15,000 , 44:40,000,
12	12	DU_J19_MP HIL_BIOPHY _New6july_ Q12	Which of the followings DOES NOT have a membrane-enclosed nucleus in the cell:	45:Archaea, 46: Fungi, 47: Yeast, 48:Protist,
13	13	DU_J19_MP HIL_BIOPHY _New6july_ Q13	You need a protein sample with concentration of 50mg/ml for your experiment. You have 1000 μ L of this sample with protein concentration of 5 mg/mL. Which of the following would lead you to the desired concentration?	49:Concentration of 500 μ L, 50: Concentration of 100 μ L, 51: Concentration of 0.05 L, 52:Concentration of 1 L,

14	14	DU_J19_MP HIL_BIOPHY _New6july_ Q14	You have a 5 M solution of NaCl, which needs to be diluted to 1 M concentration. How much water do we add to 100 ml of such solution to make it correct molarity?	53:0.5 L, 54: 0.4 L, 55: 300 mL, 56:500 mL,
15	15	DU_J19_MP HIL_BIOPHY _New6july_ Q15	Proteins are known to undergo various modifications after their synthesis, known as post-translational modifications. How many such variations are currently known:	57: ~20, 58: ~200, 59: ~2000, 60: ~5,
16	16	DU_J19_MP HIL_BIOPHY _New6july_ Q16	Some proteins are known to carry out multiple functions in an organism. Such proteins are known as:	61: Universal p 62: Sunny pro 63: Moonlighti 64: Twinkling p
17	17	DU_J19_MP HIL_BIOPHY _New6july_ Q17	In the acronym "siRNA", the letter "si" stands for:	65: small inges 66: small inter 67: short inhib 68: short mRN
18	18	DU_J19_MP HIL_BIOPHY _New6july_ Q18	In protein structure visualization programs, the nitrogen atoms are usually depicted in this color:	69: Grey, 70: Yellow, 71: Red, 72: Blue,
19	19	DU_J19_MP HIL_BIOPHY _New6july_ Q19	In prokaryotes, the genes for related function are often present in genetic units that are regulated together. This arrangement is called as:	73: a linkage g 74: an Operon 75: a cistron, 76: a CDS,
20	20	DU_J19_MP HIL_BIOPHY _New6july_ Q20	Which of the following techniques can be used to find the secondary structure content of a protein molecule without any information of the three-dimensional structure information?	77:NMR (Nucle Resonance), 78: Circular di spectroscopy, 79: Size exclus 80:X-ray cryst
21	21	DU_J19_MP HIL_BIOPHY _New6july_ Q21	Which of the following statements is CORRECT for double-stranded nucleic acids i. Two strands are associated by hydrogen bonds ii. Sequences are complementary and antiparallel iii. The back-bones are made of phosphor-diester bonds iv. Numbers of hydrogen bonds between two nucleotides are not uniform	81:All of the al 82: All of the a 83: All of the a

				84:Only (i) and
22	22	DU_J19_MP HIL_BIOPHY _New6july_ Q22	A double-stranded RNA genome isolated from a virus in the stool of a child with gastroenteritis was found to contain 15% uracil. What is the percentage of guanine in the viral genome?	85:15, 86: 25 , 87: 35 , 88:75,
23	23	DU_J19_MP HIL_BIOPHY _New6july_ Q23	A gene encodes a protein with 150 amino acids. There is one intron of 1000bps, a 5'-untranslated region of 100bps and a 3'-untranslated region of 200bps. In the final processed mRNA, how many bases lie between the start and final termination codon?	89:1750, 90: 750, 91: 650, 92:450,
24	24	DU_J19_MP HIL_BIOPHY _New6july_ Q24	Western blot is used to probe	93:DNA, 94: RNA, 95:Protein, 96:Single stran
25	25	DU_J19_MP HIL_BIOPHY _New6july_ Q25	Some cells in the adult animals do not divide (e.g., heart cells). These cells enter an inactive stage of the cell cycle called as	100:G2 phase, 97:G0 phase, 98: G1 phase, 99: S phase,
26	26	DU_J19_MP HIL_BIOPHY _New6july_ Q26	Which of following is not a protein	101: Spider v 102: Rhino h 103: Cobra v 104: Jute,
27	27	DU_J19_MP HIL_BIOPHY _New6july_ Q27	In which of the following cases, drug resistance is known to occur	105: Mycobac 106: HIV, 107: Cancer, 108: All of th
28	28	DU_J19_MP HIL_BIOPHY _New6july_ Q28	What does pH 0 indicate?	109: 1 molar 110: 1 molar , 111: A very s 112: A buffer acid is aced,
29	29	DU_J19_MP HIL_BIOPHY _New6july_ Q29	What is the advantage of glycolysis, since it taps only a small fraction of the energy available in the glucose molecule?	113:It may be is unavailable.

		New6july Q29		114: It is cyclic substrate is re 115: It require ATP., 116:It is comp spontaneous re
30	30	DU_J19_MP HIL_BIOPHY _New6july_ Q30	Which of the following statements are correctly describing the transport system found in plants? i) Xylem: water and nutrients from root to shoots:: Phloem: food synthesized in leaves to other parts ii) Xylem: upward movement only :: Phloem: Both up and down movement iii) Xylem: outside of vascular bundle :: Phloem: centre of vascular bundle iv) Xylem: centre of vascular bundle : : Phloem: outside of vascular bundle	117:Statement 118: Statemen 119: Statemen 120:Statement
31	31	DU_J19_MP HIL_BIOPHY _New6july_ Q31	The attachment site for RNA polymerase on the DNA template is called as	121:Cistron, 122: Regulator 123: Promoter 124:Intron,
32	32	DU_J19_MP HIL_BIOPHY _New6july_ Q32	High level of one hormone/protein results in diminution of a second hormone/protein. This phenomena is called as	125:Negative f 126: Hermaph 127: Positive f 128:Covarianc
33	33	DU_J19_MP HIL_BIOPHY _New6july_ Q33	The variable region of an antibody is primarily responsible for	129:Specificity antigen, 130: Three-dir of antibody, 131: Transport distant location 132:Disulfide b
34	34	DU_J19_MP HIL_BIOPHY _New6july_ Q34	Gram staining is an example of	133:Differentia 134: Acid fast 135: Negative 136:Spore stai
35	35	DU_J19_MP HIL_BIOPHY	Kinetic Theory of Gases deals with	137: macrosc the system.,

		New6july Q35		138: microscop system., 139: both mi properties of th 140: neither m macroscopic pr system.
36	36	DU_J19_MP HIL_BIOPHY _New6july_ Q36	Thermodynamics deals with	141:macroscop system., 142: microscop system., 143: both micr of the system. 144:neither mi macroscopic pr system.
37	37	DU_J19_MP HIL_BIOPHY _New6july_ Q37	A system of ideal gas has undergone change from one state to another state. While undergoing the change in state, the work done in a reversible process	145:Is equal to an irreversible 146: Is greater done in an irre 147: Is lesser in an irreversib 148:Is either g than the work irreversible pro
38	38	DU_J19_MP HIL_BIOPHY _New6july_ Q38	The First Law of Thermodynamics deals with	149:Flow of en direction, 150: Increase system and the 151:Conservat work on or by 152:None,
39	39	DU_J19_MP HIL_BIOPHY _New6july_ Q39	Van der Waals distance between two molecules in a gas arises due to	153:Strong ele interaction bet .

		Q39		154: Strong in the nuclei of the 155: Non-negligible molecules, , 156: Negligible molecules.,
40	40	DU_J19_MP HIL_BIOPHY _New6july_ Q40	Maxwell Boltzmann distribution of kinetic energy of molecules is based on	157: Random distribution of velocities., 158: Equal distribution of velocities., 159: Linear distribution of velocities., 160: Power law distribution of velocities.,
41	41	DU_J19_MP HIL_BIOPHY _New6july_ Q41	Resonance occurs when	161: Components of different wavelengths, b 162: Components of different wavelengths, b 163: Components of different wavelengths and phases.. 164: Components of different wavelengths and phases..
42	42	DU_J19_MP HIL_BIOPHY _New6july_ Q42	Oxygen Molecule (O ₂) is	165: Diamagnetic 166: Paramagnetic 167: Ferromagnetic 168: None. ,
43	43	DU_J19_MP HIL_BIOPHY _New6july_ Q43	Purine is	169: An aliphatic compound 170: A heterocyclic compound , 171: A heterocyclic compound,

				172: A heterocyclic compound , 173: Symmetrical 174: Asymmetrical 175: No carbon 176: Double bond
44	44	DU_J19_MP HIL_BIOPHY _New6july_ Q44	Optically active organic compounds must have	
45	45	DU_J19_MP HIL_BIOPHY _New6july_ Q45	The frequencies of the following electromagnetic radiations are of the order	177: Visible > ultraviolet 178: X-ray > ultraviolet 179: Ultraviolet 180: X-ray > visible
46	46	DU_J19_MP HIL_BIOPHY _New6july_ Q46	The sum of the series $1, 2, 4, 8, 16, \dots, 2^n$ is	181: N_2 , 182: $2n$, 183: $2n-1$, 184: None,
47	47	DU_J19_MP HIL_BIOPHY _New6july_ Q47	Three resistors 1Ω , 2Ω , 3Ω are combined in series. What is the equivalent resistance of the combinations?	185: 10Ω , 186: 6Ω , 187: 5Ω , 188: 25Ω ,
48	48	DU_J19_MP HIL_BIOPHY _New6july_ Q48	The curve $x^2/4 + y^2/9 = 1$ has major & minor axes	189: 2 & 2 respectively 190: 3 & 3 respectively 191: 2 & 3 respectively 192: 3 & 2 respectively
49	49	DU_J19_MP HIL_BIOPHY _New6july_ Q49	For a chemical reaction the Equilibrium constant is related to the Forward Rate Constant k_f and Backward Rate Constant k_b as below.	193: $K = k_f + k_b$ 194: $K = k_f - k_b$ 195: $K = k_f / k_b$ 196: $K = k_f \times k_b$
50	50	DU_J19_MP HIL_BIOPHY _New6july_ Q50	The total change in entropy for an irreversible process is	197: 0, , 198: Positive, , 199: Negative, , 200: both positive & negative