

**CODE NO: 6102/PCI** 

## **FACULTY OF PHARMACY**

M. Pharmacy I - Semester (Main & Backlog) Examination, January 2020 (Common Paper for all Except Pharmacy Practice)

**Subject: Modern Pharmaceutical Analytical Techniques** 

Time: 3 Hrs Max. Marks: 75

Note: Answer any Five Questions. All Questions Carry Equal Marks.

1.	(a) State and explain Beer- Lambert's law. Add a note on the deviations from Belaw.	er's 8
	<ul><li>(b) Explain solvents and the selection criteria for UV/Visible spectroscopy.</li><li>(c) What is solvent shift?</li></ul>	4
2.	(a) Explain the principle and instrumentation of FTIR with a neat labelled diagram.	8
	(b) Explain abt the sampling techniques and applications of FR spectroscopy	7
3.	(a) What is the principle of Fluorescence? Explain the radiative and non radiative pathways of relaxation.	tive 7
	(b) Add a note on the factors affecting fluorescence and quenchers in fluorescence.	6
	(c) What are the criteria for a molecule to exhibit the phenomena of fluorescence	2
4.	(a) Explain the principle of proton NMR spectroscopy.	5
	(b) What is the significance of chemical shift. What are the factors affecting chemical shift?	6
	(c) Explain abt spin -spin crippling and it's importance in NMR	4
5.	<ul><li>(a) Classify the ionization techniques in MS. Explain any three methods in detail.</li><li>(b) Differentiate between Base peak and molecular ion peak.</li></ul>	12 3
6.	<ul><li>(a) Explain HPLC instrumentation.</li><li>(b) What are the applications of HPLC?</li></ul>	10 5
7.	<ul><li>(a) Explain Braggs equation and derive the equation.</li><li>(b) What is the principle involved in rotating crystal technique?</li></ul>	8 7
8.	Explain the principle, working and applications of	
	(a) Capillary electrophoresis (b) Gel electrophoresis	7 <sup>1/2</sup> 7 <sup>1/2</sup>
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