

GUJARAT TECHNOLOGICAL UNIVERSITY**BE - SEMESTER– III (New) EXAMINATION – WINTER 2019****Subject Code: 3132004****Date: 3/12/2019****Subject Name: Principles of Materials Science and Metallurgy****Time: 02:30 PM TO 05:00 PM****Total Marks: 70****Instructions:**

1. Attempt all questions.
2. Make suitable assumptions wherever necessary.
3. Figures to the right indicate full marks.

	MARKS
Q.1 (a) Differentiate material science and metallurgy.	03
(b) Write a short note on Creep and Fatigue.	04
(c) Describe material selection process for turbine blade.	07
Q.2 (a) Enlist three non-ferrous metals.	03
(b) What is APF and CN? State CN for BCC, SC and FCC unit cell system.	04
(c) What is the significance of crystallographic planes and directions? Enlist the steps to find out crystallographic planes within unit cell.	07
OR	
(c) Differentiate among single crystal and polycrystalline materials. Also explain isotropy and anisotropy of the materials.	07
Q.3 (a) Explain Bragg's law for crystal structure determination.	03
(b) Explain jominy hardenability test.	04
(c) Enlist and explain different point defects in metallic and ionic materials.	07
OR	
Q.3 (a) Define linear and planer density.	03
(b) Enlist the steps involved in liquid penetrant test. Explain each briefly.	04
(c) Give a detailed overview of ultrasonic testing method.	07
Q.4 (a) Explain sintering process in powder metallurgical process.	03
(b) Give the advantages and disadvantages of MPT.	04
(c) Explain the ductile and brittle failures with relevant diagram.	07
OR	
Q.4 (a) What is the requirement of performing jominy hardenability test?	03
(b) Give brief overview of flame hardening process.	04
(c) Define hardness. Explain Rockwell hardness test in detail.	07
Q.5 (a) State gibb's phase rule giving its significance in metallurgy.	03
(b) Explain significance of TTT diagram.	04
(c) Enlist the various techniques for powder production. Explain any one with neat sketch.	07
OR	
Q.5 (a) Enlist the steps involved in micro examination of specimen.	03
(b) Draw and explain isomorphous binary phase diagram.	04
(c) Draw iron-iron carbide equilibrium diagram with all necessary details. Briefly explain cooling of 0.8 % carbon steel.	07
