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## GUJARAT TECHNOLOGICAL UNIVERSITY

		<b>BE - SEMESTER- IV (New) EXAMINATION - WINTER 2019</b>	
Subject Code: 2141302 Date: 13/12/2019			
	Subject	Name: Environmental Sciences II	
	Time: 10:30 AM TO 01:00 PMTotal Marks: 70		
	Instruction	ns:	
	1.	Attempt all questions.	
	2. 3.	Figures to the right indicate full marks.	
		I gures to the right malcute run murks.	Marks
01	(a)	Define: Sol emulsion tyndall effect	3
<b>V</b>	(b)	Enlist the mechanisms by which colloids can be coagulated	4
	(c)	Explain Aliphatic and Aromatic compounds with examples.	7
	(-)		
Q.2	2 (a)	State Schulze-Hardy rule.	3
	<b>(b)</b>	The solubility of BaCO3 is 5.9 x 10-5 M at 25°C. Calculate the solubility product i.e. Ksp.	4
	( <b>c</b> )	Explain Ionization of weak acids with examples.	7
		OR	
	( <b>c</b> )	Write a short note on binary mixtures.	7
Q.3	<b>3</b> (a)	Explicate the basic differences between BOD and COD	3
	<b>(b</b> )	Define Colloids. Explain surface charge of colloids.	4
	(c)	Write a short note: Classes of amino acids	7
		OR 🔨	
Q.3	<b>3</b> (a)	Explain in brief TOC.	3
	<b>(b)</b>	Explain solubility product with appropriate examples.	4
	( <b>c</b> )	Explain the biochemistry of (i) Carbohydrates (ii) Proteins	7
Q.4	(a)	What are the factors affecting solubility of oxygen in water.	3
	(b)	What is common ion effect? Give its application in waste water treatment with examples	4
	( <b>c</b> )	Explain modified Winkler Method with Chemical Reaction for DO determination.	7
		OR	
<b>Q.</b> 4	(a)	How dilution water is prepared in laboratory for BOD?	3
	<b>(b)</b>	Why organic matter removal is important for wastewater treatment?	4
	( <b>c</b> )	How will you determine Oil and Grease in laboratory?	7
Q.5	5 (a)	Differentiate: physical adsorption and chemisorption	3
	<b>(b</b> )	Write brief note biological properties of pesticides.	4
	(c)	Describe in brief the methods used for the determination of volatile acids. Explain applications of volatile acids data.	7
		OR	
Q.5	5 (a)	Explain working principle of COD determination with examples.	3
	<b>(b)</b>	Define with suitable examples: Primary alcohol (ii) Secondary alcohol (iii) tertiary alcohol	4
	( <b>c</b> )	Explain the chemistry involved in Argentometric method for determination of chloride with pH consideration.	7