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GUJARAT TECHNOLOGICAL UNIVERSITY

		BE - SEMESTER- V (New) EXAMINATION - WINTER 2019	
Subject Code: 2152306 Date: 21/11/2019			
Sub	iect]	Name: Chemistry of Plastic Materials	
Time: 10:30 AM TO 01:00 PM Total Marks: 7			
Instr	uctior	18:	
	1.	Attempt all questions.	
	2.	Make suitable assumptions wherever necessary.	
	3.	Figures to the right indicate full marks.	MADKS
			MARKS
Q.1	(a)	Give one example of alternating and block copolymers with structure.	03
	(b)	Define and Explain Homopolymer and co-polymer.	04
	(C)	Give classification of Polymer in detail with example.	07
Q.2	(a)	What is difference between Buna-N and Buna-S rubber?	03
c	(b)	The Mn of Polypropylene is 106gm/mol. Find the DPn.	04
	(c)	Discuss Carothers' Equation in detail.	07
		OR	
	(c)	Explain the mechanism of Anionic addition polymerization in detail.	07
0.2	(a)	List the generalized stops for polymorization	02
Q.3	(a) (b)	Give chemical structure of the following polymer: (1) PP (2) PVC (3)	03
	(0)	PET	U4
	(c)	Explain various types of bonding exist in polymers with example.	07
		OR	
Q.3	(a)	Write chemistry of formation of Melamine Formaldehyde (MF)	03
	(b)	Explain the manufacturing process of MF with flow diagram.	04
~ 4	(c)	Explain in brief about Tapping (Extraction) of Latex.	07
Q.4	(a)	What is natural rubber? Give the structure of polyisoprene.	03
	(D)	What is compounding. Explain vulcanization of rubber in detail.	04
	(C)	write structure, properties, and application of chlorinated rubber.	07
		OR	
Q.4	(a)	Calculate the number average degree of polymerisation of an	03
		equimolecular mixture of hexamethylenediamine and adipic acid for	
		the extents of reaction 0.5 and 0.8.	
	(b)	In the polymerisation of ω -hydroxy caproic acid, HO(CH2)5COOH, a	04
		2% impurity present. Calculate the degree of polymerisation of	
		polymer formed.	~ -
	(c)	Derive the following equation for free radical polymerization $Rp = Kp$	07
05	(\mathbf{a})	(Kd ¹² /Kt ¹²){(I[I] ¹²)[M]} Short note on: Natural Balymar Starah and Lianin	02
Q.5	(a) (b)	Give chemical structure of PTEE Discuss the properties of PTEE	03
	(U) (c)	Discuss chemistry, properties and application of HDPE	04
	(0)	OR	07
0.5	(ล)	Derive the equation between number average degree of polymerisation	03
×	()	(P) and Kinetic chain length (γ).	
	(b)	Differentiate: Crystalline Polymer and Amorphous Polymer.	04
	(c)	Short note on: Tacticity of Polypropylene.	07
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