

0.3

Q.4

Seat No.: _ Enrolment No._ GUJARAT TECHNOLOGICAL UNIVERSITY BE - SEMESTER-VI(NEW) - EXAMINATION - SUMMER 2019 Subject Code:2163506 Date: 10/05/2019 **Subject Name: Unit Operations-II** Time:10:30 AM TO 01:00 PM **Total Marks: 70 Instructions:** 1. Attempt all questions. 2. Make suitable assumptions wherever necessary. 3. Figures to the right indicate full marks. (a) Explain Reverse osmosis with proper diagram. 03 Q.1* 04 (b) Differentiate Pyrolysis and Incineration. A batch of 120 Kg wet solid has initial moisture content of 0.4 Kg water/Kg 07 dry solid. The exposed area for drying is 0.05 m²/Kg dry solid. The rate of drying follows the following curve. Calculate the time required (in hours) for drying this batch to a moisture content of 0.1 Kg water/Kg dry solid. Rate of drying (kg water removed m-2 lu-1) 1.2 1.0 08 04 Moisture content, X (kg water/kg dry solid) (a) Explain Pneumatic Conveyers with proper diagram. 03 Q.204 Explain drying rate curve. Explain the troubles occurs in Rapid sand filters with proper diagram. 07 (c) OR Classify types of dryers and explain any two with proper diagram. 07 Differentiate cross flow and ded-end flow of membrane. 03 Q.3 (a) Explain different types of membrane. 04 Describe different types of membrane module (with figure) and 07 (c) working principal of microfiltration.

 $N = 0.0015 X \text{ for } X \le 0.3$

Explain fluidization process and derive the equation of pressure drop

Define (a) Equilibrium moisture (b) Free moisture (c) Bound moisture

Calculate the time required to reduce the moisture contents of a solid from

0.66 to 0.25 Kg moisture/Kg dry solid. The rate of drying N (Kg Water

Explain types of fouling in the membrane.

(b) Explain different types of membrane modules.

Explain selection criteria for absorbent.

evaporated/m² s) is given as following,

per unit length in the column.

(d) Un-bound moisture.

03

04

07

03

04

07

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= 0.00045 for X > 0.3

where X = Kg moisture/Kg dry solid. The drying surface is 0.025 m²/Kg dry solid.

OR

Q.4	(a)		03
	(b)		0.
	(c)		07
Q.5	(a)	Explain working principal of Microfiltration, Advantages and disadvantages.	03
	(b)		04
	(c)	ADSOrber: Gas in 0.250 m ³ /s at 26 °C. $P_1 = 1.07 \times 10^5 \text{ N/m}^2 \text{ Containing } 2.0\%$	07
		by volume of light oil vapors. The light oil will be assumed to be entirely	0,
		benzene and a 90% removal is required. The wash oil is to enter at 26 °C,	7.5
		containing 0.005 mole fraction benzene and has an average molecular weight	
		is 260. An oil circulation rate of 2 times the minimum is to be used. Wash oil benzene solution are ideal. Coloulete Astrol Visit G	
		benzene solution are ideal. Calculate Actual liquid flowrate and Mole fraction of benzene actually coming out of the tower in the liquid.	
		Data: The vapor pressure of benzene at 26 °C is 13330 Pa.	
Q.5	(a)	OR	
Ųω	(a)	Explain ultrafiltration.	03
	(b)	What are the types of scrubber and explain any one.	04
	(c)	In a Petrochemical Plant gas containing 4 % cyclohexane and rest is inert has	07
		to be treated with a non-volatile absorption oil in a packed tower. It is required	
		to removed 94 % of the cyclohexane of the feed gas. The feed solvent is free from cyclohexane. If the feed gas rate is 70 kmol/hr. Calculate the minimum	
		solvent rate at the equilibrium ralation 1: XX 0.2 X	
		solvent rate at the equilibrium relationship $Y = \frac{0.2 X}{1+0.8 X}$	