

GUJARAT TECHNOLOGICAL UNIVERSITY

BE - SEMESTER-IV (NEW) EXAMINATION – WINTER 2018

Subject Code: 2143606
Date: 12/12/2018
Subject Name: Advanced Organic Chemistry for Technologists
Time: 02:30 PM TO 05:00 PM
Total Marks: 70
Instructions:

1. Attempt all questions.
2. Make suitable assumptions wherever necessary.
3. Figures to the right indicate full marks.

		MARKS
Q.1	(a) Write short note on Markovnikov's Rule.	03
	(b) Write short notes on: Racemic Mixture & Meso compounds	04
	(c) Define Aromaticity. Explain Huckel rule in detail showing the behavioural approach of different Organic molecules in proving their aromatic characteristics.	07
Q.2	(a) State whether the following compounds are aromatic or non-aromatic	03
	(b) Explain Claisen - Dieckmann condensation in details.	04
	(c) Explain why,	07
	1. Pyridine is more basic than pyrrole. 2. Pyridine is less basic than aliphatic amine. 3. Pyridine is more basic than aniline	
	OR	
	(c) Discuss Pinacol-Pinacolone rearrangement in detail.	07
Q.3	(a) Write a note on conformational isomerism.	03
	(b) Explain Hydroboration reaction with suitable examples.	04
	(c) Write detailed mechanism of SN1 & SN2 reactions with suitable examples.	07
	OR	
Q.3	(a) Give Oxidation reaction in detail.	03
	(b) Identify enantiomeric and diastomeric pairs from following	04
	<div style="display: flex; justify-content: space-around; align-items: flex-end;"> <div style="text-align: center;"> $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{H} - \text{C} - \text{OH} \\ \\ \text{H} - \text{C} - \text{OH} \\ \\ \text{CH}_2\text{OH} \end{array}$ I </div> <div style="text-align: center;"> $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{HO} - \text{C} - \text{H} \\ \\ \text{HO} - \text{C} - \text{H} \\ \\ \text{CH}_2\text{OH} \end{array}$ II </div> <div style="text-align: center;"> $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{HO} - \text{C} - \text{H} \\ \\ \text{H} - \text{C} - \text{OH} \\ \\ \text{CH}_2\text{OH} \end{array}$ III </div> <div style="text-align: center;"> $\begin{array}{c} \text{CO}_2\text{H} \\ \\ \text{H} - \text{C} - \text{OH} \\ \\ \text{HO} - \text{C} - \text{H} \\ \\ \text{CH}_2\text{OH} \end{array}$ IV </div> </div>	
	(c) Write detailed mechanism and applications of Knoevenagel condensation.	07
Q.4	(a) Write a note on: Opposite behavior of Halogen group.	03
	(b) Write down electrophilic substitution reaction of Pyrrole.	04
	(c) Explain detailed mechanism of Clemmensen reduction.	07

OR

- Q.4** (a) Show optical activity of Lactic acid. **03**
(b) Explain detailed mechanism of Michael reaction. **04**
(c) Write a note on aromaticity of Heterocyclic compounds. Also write Nitration, Sulphonation and Acylation reactions of Furan. **07**
- Q.5** (a) Write a short notes on: Petroleum as sources of aromatic compounds **03**
(b) Enlist chemical reactions of Naphthalene. **04**
(c) Write detailed mechanism and applications of Leuckart reaction. **07**

OR

- Q.5** (a) Explain why, Nitro (-NO₂ group) acts as ortho-para director. **03**
(b) Write a note on Benzidine rearrangement. **04**
(c) Write a note on NGP. **07**

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