

Rajiv Gandhi University of Health Sciences, Karnataka I Year Pharm-D Examination - NOVEMBER 2015

Time: Three Hours Max. Marks: 70 Marks

PHARMACEUTICAL INORGANIC CHEMISTRY

O.P. CODE: 2855

Your answers should be specific to the questions asked Draw neat labeled diagrams wherever necessary

LONG ESSAYS (Answer any Two)

 $2 \times 10 = 20 \text{ Marks}$

- 1. Define Limit test? Write the principle and procedure involved in the limit test for Arsenic with a neat labeled diagram.
- 2. Describe the principle and procedure involved in the Mohr's and Volhard's method for the determination of Chlorides.
- 3. Describe the principle and reactions in the assay of a) Chlorinated lime b) Hydrogen peroxide.

SHORT ESSAYS (Answer any Six)

 $6 \times 5 = 30 \text{ Marks}$

- 4. Define Redox titrations and explain with an example.
- 5. Write the preparation storage, labeling and uses of oxygen.
- 6. What are Antacids? Classify them with an example for each class. Write the method of preparation of aluminium hydroxide gel.
- 7. Write the method of preparation and standardization of 0.1 N potassium permanganate.
- 8. Discuss the various types of solvents used in non-aqueous titrations.
- 9. What are saline cathartics? Give the method of preparation and use of magnesium sulphate.
- 10. Define and classify errors with examples.
- 11. Write the principle and reaction involved in the limit test for iron I.P.

SHORT ANSWERS 10 x 2 = 20 Marks

- 12. What are Anti-microbials? Give two examples.
- 13. Define expectorants with examples.
- 14. Define Primary and Secondary standards. Give examples.
- 15. Composition of ORS.
- 16. Define indicators with examples.
- 17. What is the importance of calcium in human nutrition?
- 18. Examples for haematinics.
- 19. Define normality and molarity.
- 20 Why lead acetate cotton plug is used in the limit test for Arsenic.
- 21. Name two Pharmaceutical Aids.
