

# Rajiv Gandhi University of Health Sciences, Karnataka I Year Pharma-D Examination - Mar 2013

Time: Three Hours Max. Marks: 70 Marks

## PHARMACEUTICAL ORGANIC CHEMISTRY

Q.P. CODE: 2854

Your answers should be specific to the questions asked Draw neat labeled diagrams wherever necessary

### LONG ESSAYS (Answer any Two)

2 x 10 = 20 Marks

- a) Give an account on Markonikov's and Anti Markonikov's additions to alkenes, giving examples.
  - b) Explain the mechanism and orientation involved in the hydration of alkenes
- Give the method of preparation and uses of the following.
  - a) Benzyl benzoate b) Lactic acid c) Saccharin sodium d) Methyl salicylate e) Dimercaprol
- 3. a) Discuss the resonance and orbital descriptions of Allyl cation.
  - Explain hyperconjugation and its importance in the study of stability of carbocations and free radicals.

#### SHORT ESSAYS (Answer any Six)

6 x 5 = 30 Marks

- 4. Discuss the mechanism of Friedel-Craft's alkylations in benzene. What are their limitations?
- Write a note on the acidity of carboxylic acids.
- Discuss the conditions that favour unimolecular substitution over bimolecular substitution in Alkyl halides.
- Write a note on Heat of hydrogenation and compare the stability of alkenes.
- Explain the concept of aromaticefy and Huckel's rule with examples.
- Define and classify isomerism with examples.
- Write a note on Elimination Vs Substitution.
- Explain the orientation and rearrangements involved in E<sub>1</sub> reactions.

#### SHORT ANSWERS 10 x 2 = 20 Marks

- Draw the structural formula for the following:
   a)trans-1,2-dichloroethene b) 1-Penten-4-yne
- Explain the term "Polarity of bonds".
- What is crossed Aldol condensation? Give the equation.
- 15. How will you assay dimercaprol?
- Draw the structural formula for the following:
   a) 2-Methyl-2butene b) 2,5-Dimethyl-2-hexene.
- Arrange the free radicals in their order of stability.
   a) Vinyl, CH<sub>2</sub>=CH
   b) Allyl, CH<sub>2</sub>=CH-CH<sub>2</sub>
   c) Benzyl, C<sub>6</sub>H<sub>6</sub>-CH<sub>2</sub>
- Give the Lowry-Bronsted and Lewis theory of acid and base.
- 19. Give the structure and uses of
  - a) Chlorbutol b) Tartaric acid
- 20. Classify amines with examples and structures.
- Distinguish between intra and intermolecular hydrogen bonding. Give examples.

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