

PIC 2855 2012 2 S101.doc

Rajiv Gandhi University of Health Sciences, Karnataka I Year Pharma-D Examination - Sep 2012

Time: Three Hours Max. Marks: 70 Marks

PHARMACEUTICAL INORGANIC CHEMISTRY

Q.P. CODE: 2855

Your answers should be specific to the questions asked Draw neat labeled diagrams wherever necessary

LONG ESSAYS (Answer any Two)

 $2 \times 10 = 20 \text{ Marks}$

- 1. a) Define an antacid? Classify those giving examples. Why combination of compounds is preferred in antacid therapy (4 marks)
 - b) Write the method of preparation and principle involved in the assay of Sodium Bicarbonate (6 marks)
- 2. a) Explain the principle involved in the limit test for Arsenic, with reactions (5 marks)
 - b) Describe the Arsenic limit test apparatus, with neat labeled diagram (5 marks)
- a) What are non-aqueous titrations? Write the principle, with reactions, involved in the assay of Sodium Benzoate (5 marks)
 - b) Give any one compound which can be assayed by using modified Volhard's method. Write the principle and reactions involved in it (5 marks)

SHORT ESSAYS (Answer any Six)

 $6 \times 5 = 30 \text{ Marks}$

- 4. Define dental anti caries? Write a method of preparation, Properties, Storage conditions, use of Sodium Fluoride
- 5. Write preparation, principle involved in the assay and used of Boric acid
- 6. Name few extra and intracellular ions. Write the normal levels and physiological role of Calcium and Phosphate ions in the body
- 7. Write procedure and principle involved in Iron Limit Test
- 8. Define haematinics with examples? Write the method of preparation, principle involved in the assay of Ferrous Sulphate
- 9. Write preparation properties, storage, assay principle for chlorinated lime
- 10. What is saline cathartic? Write preparation and assay of Magnesium Sulphate
- 11. Write a note on theory of indicators

SHORT ANSWERS $10 \times 2 = 20 \text{ Marks}$

- 12. Write the composition of Barium sulphate reagent used in limit test for sulphates?
- 13. Give a method of preparation and use of Nitrous Oxide
- 14. Chloride limit test principle
- 15. Write the composition of Ringer lactate solution, mention its use
- 16. Complete and balance the following Na2S2O3+I2———
 Al(OH)3+HCl
- 17. Give one example for an acidifying agent and mention its use
- 18. What is the difference between Ferrous Sulphate and Dried Ferrous Sulphate?
- 19. Note on Barium Sulphate
- 20 Define the terms molarity, normality
- 21. Calculate the number of grams of NaCl in two litre of a 0.9% w/v solution
