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Total No. of Pages : 02

Total No. of Questions : 09

# B.Tech. (ME) (2018 Batch) (Sem.-3) BASIC THERMODYNAMICS Subject Code : BTME305-18 M.Code : 76422

Time: 3 Hrs.

Max. Marks : 60

## INSTRUCTIONS TO CANDIDATES :

- 1. SECTION-A is COMPULSORY consisting of TEN questions carrying TWO marks each.
- 2. SECTION-B contains FIVE questions carrying FIVE marks each and students have to attempt any FOUR questions.
- 3. SECTION-C contains THREE questions carrying TEN marks each and students have to attempt any TWO questions.

## **SECTION-A**

## 1. Write briefly :

- a. State the conditions for a process to be reversible.
- b. Define Zeroth law of thermodynamics
- c. Define Thermodynamic work.
- d. Define Enthalpy.
- e. What is throttling process?
- f. What is heat pump? How it differs from a refrigerator?
- g. State Entropy principle.
- h. What is air standard efficiency?
- i. What is explosion ratio?
- j. How I.C. engines are classified?



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### **SECTION-B**

- 2. Explain the terms state, path, process and cycle.
- 3. Write the similarity and dissimilarities between work and heat.
- 4. State first law of thermodynamics and write its corollaries.
- 5. A reversible heat engine delivers 0.6 kW power and rejects heat energy to a reservoir at 300 K at the rate of 24 kJ/min. Make calculation for the engine efficiency and the temperature of the thermal reservoir supplying heat to the engine.
- 6. Explain the process of steam generation at constant pressure and show the various stages on p-v and T-v diagrams.

## **SECTION-C**

- 7. Derive analytical expression to find the work done, heat supplied and change in internal energy in adiabatic process.
- 8. Derive expression for steady flow energy equation and discuss its various engineering applications.
- 9. Define an expression for the air standard efficiency of the Diesel cycle in terms of the compression ratio, cut off ratio and the adiabatic index.

NOTE : Disclosure of Identity by writing Mobile No. or Making of passing request on any page of Answer Sheet will lead to UMC against the Student.