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Total No. of Questions: 10

B.Pharma (2012 to 2016) (Sem.-1) PHARMACEUTICAL CHEMISTRY-I (Inorganic Pharmaceutical Chemistry)

Subject Code : BPHM-102 M.Code : 46202

Time: 3 Hrs. Max. Marks: 80

INSTRUCTIONS TO CANDIDATES:

- SECTION-A is COMPULSORY consisting of FIFTEEN questions carrying TWO marks each.
- SECTION-B contains FIVE questions carrying FIVE marks each and students have to attempt any FOUR questions.
- SECTION-C contains FOUR questions carrying TEN marks each and students have to attempt any THREE questions.

SECTION-A

Write briefly :

- Define Emetics.
- b. What do you mean by light and heavy magnesium carbonate?
- Give different causes and mechanism of tooth decay.
- Werner coordination number.
- e. Define Adsorbents.
- f. Role of nitric acid in chloride limit test
- g. Use of pumice
- Define, example and use of non-systemic antacids.
- Necessary features of good antioxidant.
- Define and classify the protective agents.
- Full form and structure of EDTA.

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- Give the importance and mechanism of sodium nitrite as antidote.
- m. What do you mean by saline cathartics?
- Define and give example of Combination antacid.
- Assay of Hydrogen peroxide.

SECTION-B

- Method of preparation and assay of any two antimicrobial agents.
- 3. Give the importance and mechanism of electrolyte used in replacement therapy.
- Explain brief account of inhalants.
- Give name different types of water.
- Comment on "Hypophosphorous acid as an Antioxidant"

SECTION-C

- Give common sources of sulphate impurity. Discuss in detail the procedure and reactions in limit test of Sulphate.
- Write a short note on :
 - a. Astringents
 - Dentrifices containing fluorides
- Explain the various types of complexometric titrations in detail.
- Give a detailed account on the various sources in final pharmaceutical product.

NOTE: Disclosure of Identity by writing Mobile No. or Making of passing request on any page of Answer Sheet will lead to UMC against the Student.

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