

# Rajiv Gandhi University of Health Sciences, Karnataka

II Year B.Pharm Degree Examination - 03-Jan-2020

Time: Three Hours Max. Marks: 80 Marks

## Physical Pharmaceutics (Revised Scheme - 2) Q.P. CODE: 1956

Your answers should be specific to the questions asked Draw neat labeled diagrams wherever necessary

### LONG ESSAYS (Answer any Two)

 $2 \times 10 = 20 \text{ Marks}$ 

- $^{1}\cdot$  What is an Adsorption Isotherm? Mention its applications. State and explain Langmuir Adsorption Isotherm.
- 2. List out the various methods to determine particle size. Explain in detail Coulter-Counter Apparatus.
- 3. Define and Classify Complexes. Define Organic Molecular Complexes with suitable examples.

#### SHORT ESSAYS (Answer any Eight)

 $8 \times 5 = 40 \text{ Marks}$ 

- 4. What is Fractional Distillation? Discuss the principle involved in this.
- 5. Explain the methodology of Accelerated Stability Testing with suitable graphs.
- 6. What is Contact Angle? Give its significance in wetting of solids with liquids with examples.
- Mention the factors effecting flow of powders. Name two agents used to improve flow properties.
- 8. What are Buffers? Write a note on physiological and pharmaceutical buffers.
- 9. What is Order of Reaction? Describe the various methods used to determine Order of Reaction.
- 10. What are Colloids? Describe different methods of purification of colloids.
- 11. What is Critical Solution Temperature? What is the effect of impurity on it?
- 12. Define and classify Emulsions. Write a note on evaluation of emulsions.
- 13. List the various apparatus used to determine Viscosity. Describe in detail Cone and Plate Method.

#### **SHORT ANSWERS**

 $10 \times 2 = 20 \text{ Marks}$ 

- 14. Azeotropic Mixtures
- 15. Pseudo Order Reaction
- 16. Structured Vehicles
- 17. Packing Arrangement of Powders
- 18. Coacervation
- 19. Dilatant Materials
- 20. DLVO Theory
- 21. Solubility Curves
- 22. Protective Colloids
- 23. BET Equation

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