

## Rajiv Gandhi University of Health Sciences, Karnataka

II Year B.Pharm Degree Examination - 03-Jan-2020

Time: Three Hours Max. Marks: 70 Marks

## PHYSICAL PHARMACEUTICS (Revised Scheme 3)

Q.P. CODE: 2606

Your answers should be specific to the questions asked Draw neat labeled diagrams wherever necessary

## LONG ESSAYS (Answer any Two)

 $2 \times 10 = 20 \text{ Marks}$ 

- 1. Define Thixotropy. Draw thixotropic curves. How the extent of thixotropic breakdown can be measured?
- 2. Enumerate the derived properties of powders. How they are evaluated?
- 3. Discuss the influence of temperature on the rate of a chemical reaction. How the shelf life of a drug formulation can be arrived at with the principles of chemical kinetics?

## **SHORT ESSAYS (Answer any Six)**

 $6 \times 5 = 30 \text{ Marks}$ 

- 4. Write on Purification of hydrophobic colloidal dispersions.
- 5. Explain what do you mean by sedimentation parameters.
- 6. State Nernst Distribution Law and mention three application of knowing partition coefficient values.
- 7. What is steady state diffusion? Explain.
- 8. What is an adsorption isotherm? Write on Freundlich adsorption isotherm.
- 9. Write on interfacial film theory to account for the stability of emulsions.
- 10. Write on any two methods for the analysis of a complex.
- 11. Define critical solution temperature. Draw CST diagram for Phenol-Water, Triethylamine-Water, Nicotine Water systems.

SHORT ANSWERS 10 x 2 = 20 Marks

- 12. Define Hydrophile-Lipophile balance number.
- 13. How is Sorensen's pH scale constructed?
- 14. Define stokes diameter and surface diameter.
- 15. Define Poise unit.
- 16. What is Plug flow? How it can be over come?
- 17. Two approaches for Protection against hydrolysis.
- 18. What are micro emulsions?
- 19. What is micellar solubilization?
- 20. What are units for k in case of a zero-order reaction?
- 21. How hydrophobic colloids are stabilized?

\*\*\*\*