

Rajiv Gandhi University of Health Sciences, Karnataka I Year B.Pharm Degree Examination – 16-Sep-2020

Time: Three Hours Max. Marks: 80 Marks

PHARMACEUTICAL ORGANIC CHEMISTRY - I (Revised Scheme - 2) Q.P. CODE: 1954

Your answers should be specific to the questions asked Draw neat labeled diagrams wherever necessary

LONG ESSAYS (Answer any Two)

 $2 \times 10 = 20 \text{ Marks}$

- Define Electrophilic and Free radical addition reactions? Give the mechanism of Markovnikov's and anti-Markovnikov's addition.
- 2. What are cycloalkanes? Describe any four methods of preparation of cycloalkanes. Write a note on Bayer's strain theory of cycloalkanes.
- 3. Define and give the mechanism of following reactions (i) Cannizzaro reaction (ii) Fries re-arrangement

SHORT ESSAYS (Answer any Eight)

 $8 \times 5 = 40 \text{ Marks}$

- 4. Explain the mechanism and stereochemistry of SN¹ reaction.
- 5. Explain the aldol and crossed aldol condensation reaction.
- 6. Define and give the mechanism of Perkins reaction.
- 7. Describe the chlorination of methane along with its mechanism.
- 8. Write a note on acidity of carboxylic acids.
- 9. Explain the effect of halogen on electrophilic aromatic substitution reaction.
- 10. Define isomerism and list different types of isomerism with examples.
- 11. Explain the mechanism and importance of Sandmeyer reaction.
- 12. Explain Friedel crafts alkylation with its limitations.
- 13. Write note on dehydration of alcohols.

SHORT ANSWERS

 $10 \times 2 = 20 \text{ Marks}$

- 14. Define and explain diazotization reaction with an example.
- 15. Define carbocation and carbanion.
- 16. What are Lewis acids? Give examples.
- 17. Define resonance with suitable example.
- 18. Give Reformatsky reaction
- 19. Define free radicals with examples.
- 20. Write the structure and IUPAC names of a) Acetone b) Formic acid.
- 21. Define bond dissociation energy and dipole moment with example.
- 22. Write short notes on hydrogen bonding.
- 23. Give structure and uses of salicylic acid and dimercaprol.
