

Rajiv Gandhi University of Health Sciences, Karnataka Third Semester B. Pharm Degree Examination – 04-Jan-2020

Time: Three Hours Max. Marks: 75 Marks

Physical Pharmaceutics - I Q.P. CODE: 5010

Your answers should be specific to the questions asked Draw neat labeled diagrams wherever necessary

LONG ESSAYS (Answer any Two)

 $2 \times 10 = 20 \text{ Marks}$

- 1. Explain in detail factors influencing solubility of drugs.
- 2. Define Refractive Index. Discuss in detail working of Abbe's refractometer.
- 3. Define surface tension. Explain in detail measurement of surface tension by capillary rise method.

SHORT ESSAYS (Answer any Seven)

 $7 \times 5 = 35 \text{ Marks}$

- 4. Explain real solution with examples.
- 5. Define dielectric constant. Write a note on its application in pharmacy.
- 6. Describe the Griffin's HLB scale.
- 7. Describe the solubilisation process.
- 8. Write a note on complexation.
- 9. Describe pH titration method for analysis of complexes.
- 10. Describe in detail electrometric determination of pH.
- 11. Write a note on Buffer capacity.
- 12. Write a note on critical solution temperature and its applications.

SHORT ANSWERS (Answer All)

 $10 \times 2 = 20 \text{ Marks}$

- 13. Write the solubility expressions for the solubility of solids in liquids.
- 14. What do you mean by glassy states?
- 15. Write any two limitations of Freundlich adsorption isotherm.
- 16. What are real and ideal solutions?
- 17. What are chelates?
- 18. Write a note on inclusion complexes.
- 19. Define the term isotonicity with examples.
- 20. Write any two applications of buffered isotonic solutions in pharmacy.
- 21. What is Henderson Hasselbalch equation? Give its applications in pharmacy.
- 22. Mention the applications of optical rotation in pharmacy.
