

# Rajiv Gandhi University of Health Sciences, Karnataka

IV Year B.Pharm Degree Examination - 07-Jan-2020

Time: Three Hours Max. Marks: 70 Marks

## INSTRUMENTAL & BIO-MEDICAL ANALYSIS

(Revised Scheme - 3)

Q.P. CODE: 2617

Your answers should be specific to the questions asked Draw neat labeled diagrams wherever necessary

### LONG ESSAYS (Answer any Two)

 $2 \times 10 = 20 \text{ Marks}$ 

- 1. What are bonded phase supports? Describe its importance in HPLC. Add a note on different types of pumps and sample injection systems used in HPLC.
- 2. What are the requirements of a potentiometric titration? Explain any one classical potentiometric titration emphasizing on the electrodes used and methods commonly followed for determination of titration end point.
- 3. Explain different development and detection techniques used in paper chromatography.

#### **SHORT ESSAYS (Answer any Six)**

 $6 \times 5 = 30 \text{ Marks}$ 

- 4. Outline differences between flame emission and atomic absorption spectrometric methods. Write any two important applications for each of these techniques.
- 5. Write the effect of polar and non-polar solvents on absorption spectra of molecules.
- 6. Describe the principle of obtaining monochromatic radiations using Gratings with a neat diagram.
- 7. What is process validation? Explain its types.
- 8. Justify the statement "Fluorimetric methods are said to be sensitive and selective".
- 9. What are bulk property HPLC detectors? Explain the construction and working of any one of them
- 10. What is column chromatography? Explain the methodology and the principle involved in column chromatographic technique.
- 11. Explain with suitable titration curves, the method to determine end-points in displacement type of conductometric titrations.

#### SHORT ANSWERS

 $10 \times 2 = 20 \text{ Marks}$ 

- 12. What is E-band? Give an example.
- 13. What is the scope of derivatization in gas chromatography? Give an example.
- 14. Write the electrode equation for a potentiometric electrode system comprising of Calomel electrode and glass electrode in determination of hydrogen ion concentration of the solution.
- 15. Write any two applications of Nephelo-turbidimetric analysis.
- 16. What is the HR<sub>f</sub>? How and for what, compounds HR<sub>f</sub> is calculated.
- 17. Write the applications of ion-exchange chromatography.
- 18. What are dynodes in PMT? How is it different from the cathode present in such detectors?
- 19. What is the difference between frontal analysis and elution analysis?
- 20. What is concentration quenching and static quenching?
- 21. Write the composition and types of paper commonly used in paper chromatographic technique.

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