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[LC 814] APRIL 2013 Sub. Code: 3814

## DOCTOR OF PHARMACY (PHARM. D) DEGREE EXAMINATION THIRD YEAR

## PAPER II – PHARMACEUTICAL ANALYSIS

Q.P. Code: 383814

Time: Three Hours Maximum: 100 marks

**Answer All questions** 

I. Elaborate on:  $(2 \times 20 = 40)$ 

- 1. Explain the working and components of gas chromatograph with help of a neat, labeled, schematic block diagram. Write a note on classification, principle, factors influencing the separation and criteria to perform the Gas chromagraphy.
- 2. a) Draw an UV-Visible double beam spectrophotometer with a neat, labeled, block diagram and explain its each component operational mode.
  - b) Why a double beam spectrophotometer gives more precise, reliable and reproducible results in comparison to a single beam spectrophotometer?
  - c) What are the factors influencing the absorption of radiant energy?

II. Write notes on:  $(10 \times 6 = 60)$ 

- 1. Give a brief and comprehensive account of application of conductometry?
- 2. State Bcer-Lambert's law. Derive it Write its application, deviation and limitation.
- 3. Discuss the underlying principle and interfecence of atomic absorption spectrometry.
- 4. Name the detectors used in HPLC. Discuss anyone the of them.
- 5. Differentiate validation and calibration. Give a brief account of validation of analytical methods.
- 6. Explain the theory and procedure of DSC.
- 7. Discuss the fundamental theory of mass spectroscopy. What are the different ions produced in mass spectrum?
- 8. What is quenching? What are different type of quenching? Write a note on factors affecting the fluorescence intensity.
- 9. Explain the various method of sample handling in IR-Spectroscopy.
- 10. Give a brief account of:
  - a) Indicator electrode, b) Polarzable electrode, c) Spectrophotometric titration.

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