

Printed Pages: 4

AS-203

(Following Paper ID and Roll No. to be filled in your  
Answer Books)

Paper ID : 199203

Roll No.

--	--	--	--	--	--	--	--	--	--

**B.TECH.****Theory Examination (Semester-II) 2015-16****ENGINEERING CHEMISTRY****Time : 3 Hours****Max. Mar : 100****Section-A****Q.1 Attempt all parts. All parts carry equal marks.****(2×10=20)**

- a. On the basis of MOT, prove that the molecule of oxygen is paramagnetic in nature.
- b. Describe the isomerism exhibited by maleic and fumaric acids.
- c. What is chemical shift? Explain.
- d. Define the terms chromosphere and auxochrome in UV spectroscopy.

1

(1)

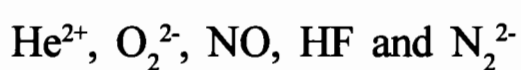
P.T.O.

- e. What is inductive effect? Give any two examples where this effect is operative.
- f. Explain why benzyl carbonium ion is more stable than ethyl carbonium ion is more stable than ethyl carbonium ion.
- g. Define reverse osmosis and cation conditioning.
- h. What are disadvantages of scale formation?
- i. How will you synthesise or/ on from acetylene?
- j. Differentiate between addition polymerisation and condensation polymerisation with suitable examples.

### Section-B

2. Attempt any five parts from this section. (10×5=50)

- a. With the help of molecular orbital diagram, calculate the bond order of the following:-



- b. Discuss in detail the case of an organic compound with two chiral centres.

- c. What is Beer Lambert law in UV-Vis-absorption spectroscopy?  
A compound having concentration  $10^{-3}$  g/l resulted absorbance value 0.20 at the  $\lambda_{\text{max}}$  510 nm using 1.0 cm. cell. Calculate its absorptivity and molar absorptivity values. Molecular weight of compound is 400/200.
- d. Give the mechanism of  $\text{SN}^1$  and  $\text{SN}^2$  organic reaction.
- e. Describe ion-exchange process of softening of water.
- f. Write note on:-
- (i) Reverse osmosis
  - (ii) Boiler corrosion
- g. Describe the process of galvanization of iron. How does it prevent the corrosion of iron?
- h. What is Ziegler-Natta catalyst? What is its significance in polymerization.

### Section-C

**Attempt any two questions from this section. (15×2=30)**

- Q.3 (a) What are corrosion inhibitors? Explain with examples how anodic and cathodic inhibitors provide protection against corrosion.

- (b) Why is it convenient to express hardness of water in terms of  $\text{CaCO}_3$  at the international level? Write other units also.

Q.4 (a) Differentiate between

- (i) Enantiomers and diastereomers
- (ii) Racemic mixture and Mesocompounds

(b) Write the preparation properties and uses of:-

- (i) Silicone rubber
- (ii) PMMA
- (iii) Bakelite

Q.5 (a) Discuss the principle and application of NMR spectroscopy in structure determination of organic compounds.

(b) Derive the rate expression for second order reaction, when the reactants are different.