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rinted Pages: 02		Su	bject Code: RME301
aper Id: 140301	Roll No:		
	В ТЕСН		
(SEM-I	II) THEORY EXAMIN	,	19
2.11	MATERIAL SCIENC	Æ	T . 136 1 50
ime: 3 Hours			Total Marks: 70
otes: Assume any Missing Data.			
	SECTION-A		
Attempt all questions in brief.			(2x7=14)
a) Define repretallination to			
Define recrystallization ter			
b) Define 'critical cooling rat			
c) What are refractory materi			
d) Define the term strain hard			
e) Differentiate between toug		o define the end	urance limit.
f) Why hardening is followed			
g) What are Bravai's space la	ttices?	Engle.	
	SECTION B	18/1/2	
. Attempt any three parts of the	ne following :	3	(7x3=21)
) To 60 d d 100 1		0 . 1	
 a) Define the term diffusion. 	explain fick's law in case	e of steady state	diffusion.

2.

- b) What is non-destructive testing (NDT)? Explain in detail any two NDT methods.
- c) What is phase diagram? Draw a neat sketch of iron-carbon diagram and explain the various phases present in it and the important reactions occurring.
- d) What do you mean by heat treatment process? Name various types of hardening processes and explain in detail Nitriding process with its merits and denierits.
- What are Ceramics? What are different types of ceramics? Also write their properties.

SECTION C

3. Attempt any one part of the following:

(7x1=7)

- a) How many atoms per square millimeter are there on the (1 0 0) plane of lead. Lead has FCC structure. Assume the interatomic distance to be 3.499 Å.
 b) NaCl structure has FCC structure. The density of NaCl is 2.18 cm³. Calculate the distance between two adjacent of the control of the contro
- between two adjacent atoms.

- 4. Attempt any one part of the following: (7x1=7)
 a) What is a dislocation? What are different types of dislocation? Explain. Draw their neat sketches and mark burger vector in each case.
 - h) Explain the term creep, its mechanism and stages.

5. Attempt any one part of the following:

a) Write short notes on Solid Solution and its types.

b) What do you understand by lever rule? Determine the mass fraction of the phases present at 184°C in a sample of lead & tin with 45% tin in it.

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6. Attempt any one part of the following:

(7x1=7)

- a) Draw and explain the TTT diagram for eutectoid steel. Explain important transformations taking place in it on cooling.
 b) What are brasses and bronzes? How are they classified? Give the composition,
- properties, microstructure and applications of any two of each.

7. Attempt any one part of the following:

(7x1=7)

What are various methods of plastic processing? Explain in detail classification and applications of composite materials.

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