

Time: Three hours

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Maximum: 100 marks

(LD 4251) AUGUST 2013 Sub Code: 4251

FIRST B.PHARM. EXAM

Paper I – PHARMACEUTICAL INORGANIC CHEMISTRY

Q.P. Code: 564251

I. Elaborate on: (2X20=40)

a) Explain the principle and procedure involved in the limit test for lea

- b) Write in detail the preparation, properties, identification tests, assay and uses of oxygen.
- a) Define and classify antacids. Explain the method of preparation, assay and uses of calcium carbonate.
 - b) What are the saline cathartics and write their mechanism of action.
 Explain the method of preparation and assay of magnesium sulphate.

II. Write Short Notes on:

 $(8 \times 5 = 40)$

- Define radiopharmaceuticals. Write about the diagnostic and therapeutic applications of radioisotopes.
- Explain about Indian Pharmacopoeia and monograph with examples.
- Write the principle involved in the assay, method of preparation and uses of boric aci
- 4. Write a short note on the electrolytes used for replacement therapy.
- Write the structure and applications of dimercaprol and pencillamine.
- Write about astringents. Describe the preparation and assay of zinc sulphate.
- Write a note on dental products and describe the role of fluorides as anticaries agent.
- What are expectorants? Write the preparation and assay of ammonium chloride.

III. Short Answers:

 $(10 \times 2 = 20)$

- Write a note on radioopaque contrast media.
- Define sedatives with examples.
- Write about adsorbents and protectives with examples.
- 4. How is acid neutralising capacity of aluminium hydroxide tested?
- Define antidotes with examples.
- Write about physiological role of potassium
- Write about primary and secondary standards with examples.
- Write a note on antioxidants with examples.
- What are pM indicators?
- Write about solutions of iodine.


