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 $(10 \times 2 = 20)$

(LJ 4252) AUGUST 2016 Sub Code: 4252

B.PHARM. EXAMINATION FIRST YEAR PAPER II – PHARMACEUTICAL ORGANIC CHEMISTRY

Q.P. Code: 564252

Time: Three hours Maximum: 100 Marks

I. Elaborate on : $(2 \times 20 = 40)$

 Explain the Kekule's structure, aromaticity and resonance structures of benzene including Huckel's rule and stability.

Explain the mechanism, reactivity and orientation of E1 and E2 reaction.

II. Short notes on: $(8 \times 5 = 40)$

Describe the mechanism of addition reaction of conjugated dienes.

- Write the preparation and properties of phenanthrene.
- Mention the preparation, test for purity and medicinal uses of lactic acid and mephenesin.
- Give the nomenclature and preparation of Cycloalkanes.
- Describe Williamson synthesis.
- Describe the properties of Alpha (α) unsaturated carbonyl compounds.
- Write the preparation and synthetic utility of Grignard reagent.
- 8. Describe the methods of preparation of alcohols.

III. Short answers on :

Define Conjugation.

- Write about Mesomeric effect.
- Give the structural formula of Urethane and Acetanilide.
- Define Ozonolysis.
- Brief about reduction of Carboxylic acids.
- Give the structural formula and IUPAC name of Acetaldehyde.
- Define Hybridization.
- Define Intra-molecular forces.
- Define Nucleophiles.
- 10. Mention the medicinal uses of Vanillin and Carbromal.

