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Sub Code: 4251

(LI 4251) FEBRUARY 2016

B.PHARM. EXAMINATION FIRST YEAR PAPER I – PHARMACEUTICAL INORGANIC CHEMISTRY

Q.P. Code: 564251

Time: Three hours Maximum: 100 Marks

I. Essays: $(2 \times 20 = 40)$

a) Write briefly about theory of Co-ordination compounds.

- b) Write the Application of Co-ordination compounds in pharmacy.
- c) Write notes on 10-Phenanthroline and Penicillamine.
- 2. a) What are Antacids? Give the classification of Antacids?
 - b) What are the qualities of an ideal Antacid?
 - Give the preparation, identification tests, assay and medicinal uses of dried Aluminium Hydroxide Gel.

II. Short notes: (8 x 5 = 40)

- 1. Explain the principle of Arsenic limit test with reactions.
- Give the properties and uses of Magnesia.
- Complete and balance the following equations:
 - a. $C_4H_2FeO_4+H_2SO_4 \rightarrow$
 - b. H₂S+SO₂ →
 - c. SrO+HCl→ FeCl₃+NaF→
 - e. Na₂HPO₄→
- Write the preparation, properties, identification, assay and uses of Hydrogen peroxide.
- Write short notes on Pharmacopoeia and Monograph.
- Write the properties and assay of Chlorinated lime.
- Explain about the preparation, assay and uses of the following:
 - (a) Oxygen, (b) Sodium Citrate
- What are the official preparations of Calcium? Give the assay and uses of any one compoun

III. Short answers: $(10 \times 2 = 20)$

- Define Molarity.
- Define Amphiprotic solvents with examples.
- Write any two precautions to be observed while handling radio active materials.
- Give the applications of Sodium Metabisulphite.
- Mention examples for adsorbent and saline Catharitic.
- Define Anti caries agents.
- 7. What are the official compounds of Iron?
- Define Sclerosing agent. Give examples.
- Mention the physiological role of Calcium and Potassium.
- Define chelating agents with examples.

