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(LL 4251) AUGUST 2017 Sub Code: 4251

B.PHARM. DEGREE EXAMINATION FIRST YEAR PAPER I – PHARMACEUTICAL INORGANIC CHEMISTRY

Q.P. Code: 564251

Time: Three hours Maximum: 100 Marks

I. Elaborate on: $(2 \times 20 = 40)$

a) What are limit tests? Explain the types of impurities present in pharmaceuticals.

- b) Enumerate the principle and procedure involved in the limit test for arsenic.
- What are Anti-microbials? Discuss the method of preparation, properties, assay and uses of: a) Iodine b) Silver Nitrate.

II. Write notes on: $(8 \times 5 = 40)$

- Write note on 1, 10 Phenanthroline and EDTA.
- 2. Illustrate the preparation and properties of Helium gas.
- Discuss on electrolytes used in combination therapy.
- State the physiological role of iron and coppe
- Define expectorant. Write the preparation and assay of ammonium chloride.
- Give in detail about Pharmacopoeia and Monograph.
- Write short notes on radio opaque contrast media.
- Write a note on dental products and describe the role of fluorides as anti-caries agent.

III. Short answers on: $(10 \times 2 = 20)$

- Write the official compounds of sodium along with its uses.
- 2. How will you assay calcium gluconate?
- Give the identification test for sulphate and barium.
- Define chelating agents with examples.
- Write about the solutions of iodine.
- Give difference between purified water and water for injection.
- Write the preparation and use of titanium dioxide.
- Define osmotic laxative. Give example.
- Give the preparation and use of selenium sulphide.
- 10. What are sedatives? Give examples.

