

(LM 4252) FEBRUARY 2018 Sub Code: 4252

## B.PHARM. DEGREE EXAMINATION FIRST YEAR PAPER II – PHARMACEUTICAL ORGANIC CHEMISTRY

Q.P. Code: 564252

Time: Three hours Maximum: 100 Marks

I. Elaborate on:  $(2 \times 20 = 40)$ 

 a) Define polynuclear hydrocarbons. Briefly explain the synthesis and chemical properties of naphthalene and diphenyl methane.

- b) Write briefly about Aldol condensation.
- a) Brief out in detail about SN1 and SN2 reactions.
  - b) Explain the mechanism of halogenations of alkanes.

II. Write notes on:  $(8 \times 5 = 40)$ 

- 1. State with example Markownikoff's rule.
- Write the preparation, test for purity and medicinal uses of sodium lauryl sulphate and iodoform.
- Explain the preparation and synthetic utility of diazonium salt.
- Write a detailed note on free radicals.
- 5. Explain various methods of synthesis and reactions of alkanes.
- Write a note on keto-enol tautomerism with examples.
- Describe briefly Diel's-Alder reaction with mechanism.
- 8. Differentiate primary, secondary and tertiary amines.

## III. Short answers on: $(10 \times 2 = 20)$

- Define hyper conjugation.
- 2. What is ozonolysis?
- 3. Huckel's rule of aromaticity.
- Write the medicinal uses of phenindione and urethane.
- Write the structure of acetic anhydride and propanal.
- Write on clemmenson reduction.
- Write about the types of bond fission.
- 8. Carbenes.
- Write one method of synthesis of amines.
- Oxidation of secondary alcohols.

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