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[BPHARM 0921]

SEPTEMBER 2021 (FEBRUARY 2021 EXAM SESSION) Sub. Code: 4257

B. PHARMACY DEGREE EXAMINATION SECOND YEAR – (NON-SEMESTER)

PAPER II – PHARMACEUTICAL ANALYSIS & PHYSICAL CHEMISTRY O.P. Code: 564257

Time: Three hours Answer ALL questions Maximum: 100 Marks

SECTION-A (PHARMACEUTICAL ANALYSIS)

I. Elaborate on: $(1 \times 20 = 20)$

 a) Illustrate briefly on Neutralization curves and the choice of indicators in acid base titrations.

Explain briefly about Argentometric titrations and its various types.

II. Write notes on: $(4 \times 5 = 20)$

- Write a note on the types of Solvents used in non-aqueous titrations.
- Discuss briefly about Masking and Demasking agent with examples.
- Discuss about Nemst equation.

4.

III. Short answers on:

 $(5 \times 2 = 10)$

- 1. Common ion effect.
- Solubility product.
- Co-precipitation.
- Universal indicators.
- Significant figure.

SECTION-B (PHYSICAL CHEMISTRY)

I. Elaborate on: $(1 \times 20 = 20)$

 a) Define Colligative properties. Describe elaborately about the types of Colligative properties.

b) What is Adsorption? Explain briefly about Langmuir adsorption isotherm.

II. Write notes on: $(4 \times 5 = 20)$

- Write a note on Joule Thomson effect.
- 2. Derive Vant-hoff equation and mention its use.
- 3. Enumerate the theory and mechanism of Catalysis.
- Describe briefly theories involved in reaction kinetics.

III. Short answers on: $(5 \times 2 = 10)$

- Second law of Thermodynamics.
- Define the terms phase, component and degree of freedom.

3.

- Factors influencing adsorption.
- Refractive index.

